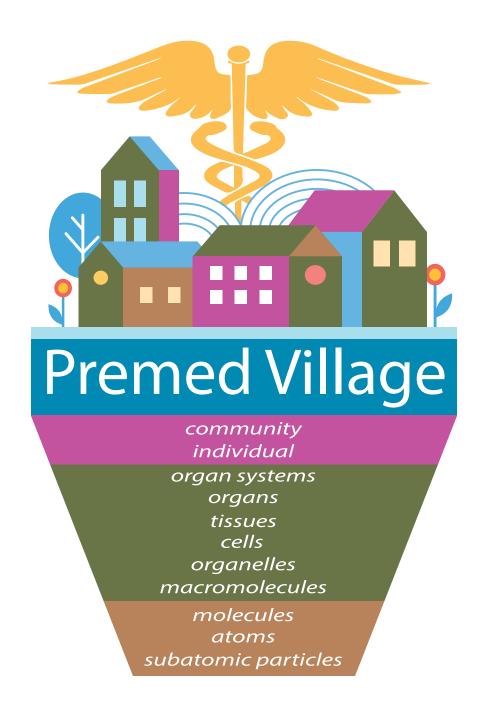
Crossword Puzzles



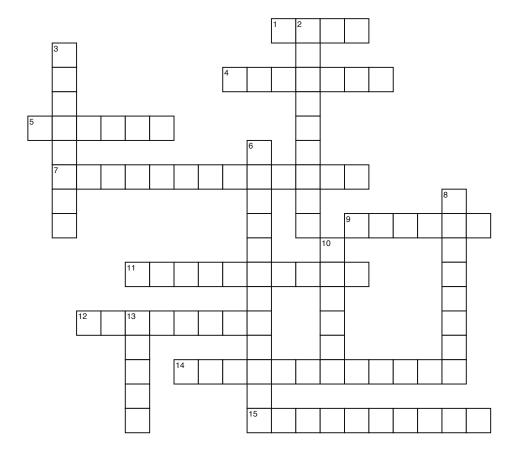
premedvillage.com

Share this book!



Puzzles

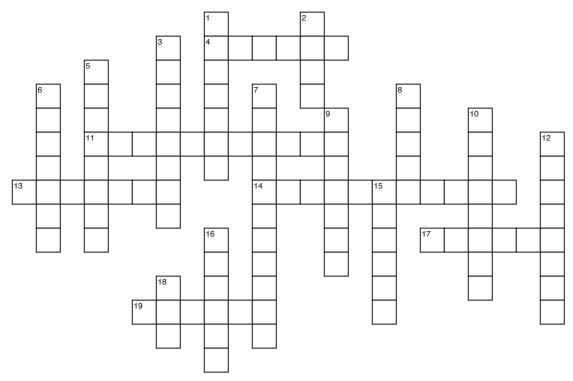
PHYSICS	ORGANIC CHEMISTRY	
Kinematics3	Functional Groups	35
Newton's Laws3	Stereochemistry	3
Work & Energy	Physical Properties	37
Harmonic Motion5	Organic Acids & Bases	
Fluic Mechanics6	Aromatic Compounds	
Waves	Organic Redox	
Temperature & Heat Flow8	Molecular Spectroscopy	
Ideal Gas9	Aldehydes & Ketones	
1st Law of Thermodynamics	Alcohols	
2nd Law of Thermodynamics	Carboxylic Acid Derivatives	
Electricity	,	
DC Current		
Magnetism 14		
Properties of Light	BIOCHEMISTRY & MOLECULAR CELL BIOLO	GY
Geometric Optics	Proteins	
Wave Optics	Carbohydrates	
Modern Physics	Nucleic Acids	
Nuclear Physics	Lipids	
Nocieur i nysics	Biological Membranes	
	Prokaryotes	
	Eukaryotes	
GENERAL CHEMISTRY	Energy Metabolism	
	Metabolic Integration	51 51
Atomic Theory	Gene Expression - Part 1	
	Gene Expression - Part 2	
Chemical Bonding	Cell Division	
Intermolecular Force	Transmission Genetics	
Stoichiometry	Mutation	
Thermochemistry	Genomics Laboratory	
States of Matter	Human Genetics	
Chemical Thermodynamics	Viruses	
Chemical Kinetics	Bacteria	
Water	bacieria	02
Solutions		
Acids & Bases		
Coordination Chemistry	HUMAN PHYSIOLOGY	
Oxidation - Reduction		
Electrochemistry34	Embryology	
	Animal Tissues	
	Nervous System	
	Sensory Systems	
	Endocrine System	
	Musculoskeletal System	
	Cardiovascular System	
	Blood	
	Respiratory System	
	Immunity	
	Urinary System	
	Digestive System	
	Reproductive System	75



1	fall is motion with no acceleration
	other than that provided by gravity.
	aivavilas mastiam dasasilas mastiam i

- 4 _____ circular motion describes motion in which an object moves with constant speed along a circular path.
- **5** _____ means a continuous change in the position of a body relative to a reference point.
- 7 The ______ System of Units (abbreviated SI) is the world's most widely used system of
- **9** A _____ is a physical quantity characterized by both magnitude and direction.
- 11 ______ is a branch of mechanics which provides the basic tools for describing the motion of objects.
- 12 _____ is a numerical description of how far apart objects are at any given moment in time.
- **14** _____ is defined as the rate of change of the velocity.
- **15** A _____ is the path a moving object follows through space.

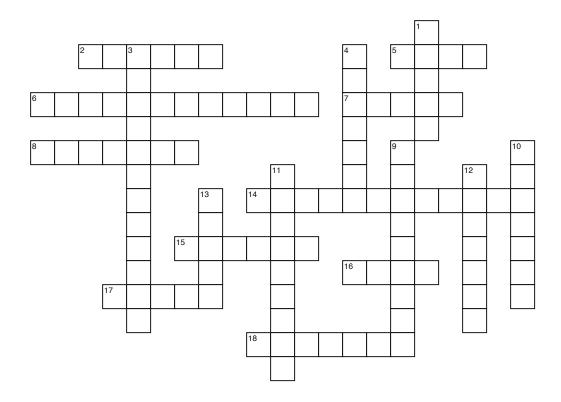
- 2 A frame of ______ is a particular perspective from which the universe is observed, providing a set of axes from which an observer can measure the position and motion of all points in a system.
- 3 _____ is defined as the rate of change of the position.
- 6 ______ is the vector that specifies the position of a point or a particle in reference to an origin or to a previous position.
- **8** A ______ vector represents the location of an object in space in relation to an arbitrary inertial frame of reference.
- 10 A ______ is a simple physical quantity that does not depend on direction, and is therefore not changed by coordinate system rotations.
- 13 _____ is the magnitude of the velocity.



4	The force is the component,
	perpendicular to the surface of contact, of the
	contact force exerted by the surface.
11	The force is the external force required
	to make a body follow a circular path at constant
	speed. The force is directed inward, toward the
	center of the circle.
13	Galilei was an Italian physicist,
	mathematician, astronomer, and philosopher who
	achieved the first systematic studies of uniformly
	accelerated motion, improved the telescope and
	supported Copernicanism.
14	A force or interaction is a mechanism by
	which particles interact with each other and which
	cannot be explained in terms of another
	interaction.
17	is a measurement of the gravitational
	force acting on an object.
19	The is the SI derived unit of force.
DC	OWN
1	is the property of an object to remain at
	constant velocity unless acted upon by an outside
	force.
2	is a fundamental concept in physics,
	roughly corresponding to the intuitive idea of how
	much matter there is in an object.

3 Newton's third law states that forces occur in pairs, one called the action and the other the _____.

5	is the force that opposes the relative
	motion or tendency toward such motion of two
	surfaces in contact.
6	A force is a force between two objects
	that are touching each other.
7	The of friction is a dimensionless
	quantity used to calculate the force of friction
	(static or kinetic).
8	is anything that can cause a massive
	body to accelerate. It may be experienced as a lift,
	a push, or a pull.
9	friction is when two solid surfaces slide
	against each other.
10	is the branch of classical mechanics that
	is concerned with the effects of forces on the
	motion of objects.
12	A frame of reference is one in which
	Newton's first and second laws of motion are valid,
4-	ie. that is neither rotating nor accelerated.
15	Newton's laws of are three physical laws
	which provide relationships between the forces
16	acting on a body and its movement through space.
10	Sir Isaac was an English scientist whose treatise Philosophiae Naturalis Principia
	Mathematica, published in 1687, described
	universal gravitation and the three laws of motion.
1Ω	A resultant or force is a vector produced
.0	when two or more forces act upon a single object.
	which two of more forces act apon a single object.



ACROSS DOWN

In physics and other sciences,	_ is defined
as a work one system does (or can do)	on another
system	
	as a work one system does (or can do)

- 5 Mechanical ______ is the amount of energy transferred by a force.
- **6** A ______ force is a force that does zero net work on a particle that travels along any closed path in an isolated system.
- 7 A ______ is a rigid object that is used with an appropriate fulcrum or pivot point to multiply the mechanical force that can be applied to another object.
- 8 The _____ energy is the energy which causes or is released by the physical distortion of a solid or a fluid
- 14 The _____ of energy states that the total amount of energy in an isolated system remains constant, although it may change forms.
- **15** A _____ is a wheel with a groove along its edge for holding a rope or cable or belt.
- **16** The _____ is the SI derived unit of power, equal to one joule per second.
- **17** The inclined _____ is a flat surface whose endpoints are at different heights.
- **18** A simple _____ is any device that only requires the application of a single force to work.

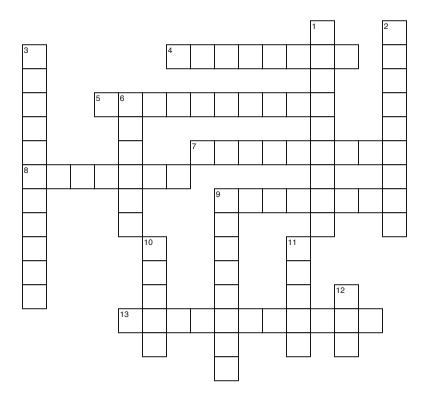
1 _____ is the rate at which work is performed or energy is transmitted. It is the amount of energy

3 The ______ is a unit of energy often used also in theoretical physics as a unit of mass. It is the amount of kinetic energy gained by a single unbound electron when it passes through an electrostatic potential difference of one volt, in vacuo.

required or expended for a given unit of time.

- 4 A ______ is a unit of measurement for energy equal to the amount of heat required to raise a gram of water one degree celsius. In most fields, it has been replaced by the joule.
- 9 Mechanical _____ is the factor by which a mechanism multiplies the force put into it.
- 10 The _____ energy of an object is the extra energy which it possesses due to its motion, defined as the work needed to accelerate the body from rest to its current speed.
- 11 _____ energy is energy stored within a physical system.
- 12 The gravitational ______ energy of an object consisting of loose material, held together by gravity alone, is the amount of energy required to pull all of the material apart, to infinity.
- 13 The _____ is the SI unit of energy.

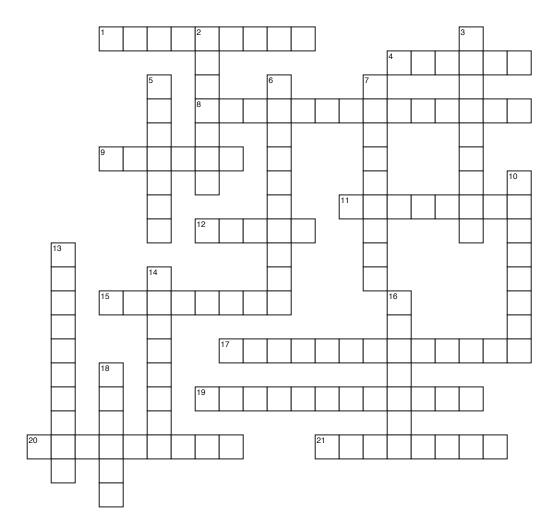
Harmonic Motion



ACROSS

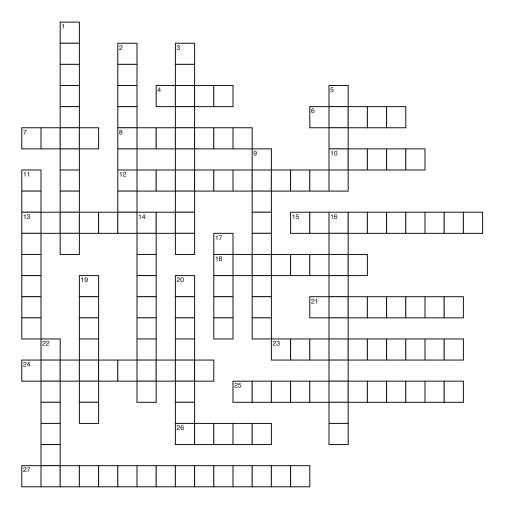
4 Simple _____ motion is the motion of a simple oscillator. **5** A harmonic _____ is a system which, when displaced from its equilibrium position, experiences a restoring force proportional to the displacement. 7 The _____ is a nonnegative scalar measure of a wave's magnitude of oscillation, the magnitude of the maximum disturbance in the medium during one wave cycle. 8 _____ is any effect that tends to reduce the amplitude of oscillations of an oscillatory system. 9 A _____ function is a function that repeats its values after some definite period has been added to its independent variable. 13 _____ is the variation of some measure about a central value (often a point of equilibrium) or between two or more different states.

1	refers to mechanical oscillations
	about an equilibrium point. The oscillations
	may be periodic such as the motion of a
	pendulum or random such as the
	movement of a tire on a gravel road.
2	is the measurement of the
	number of occurrences of a repeated
	event per unit of time.
3	is the quality of occurring at
	regular intervals or periods in time or
	space.
6	A is a flexible elastic object
	used to store mechanical energy usually
	made out of hardened steel.
9	A is an object that is attached to
	a pivot point so it can swing freely.
10	The of an oscillation or wave is
	the fraction of a cycle corresponding to an
	offset in the displacement from a specified
	reference point at time t = 0.
11	The is the SI unit of frequency.
	Its base unit is the cycle per second.
12	A is the weight on the end of a
	pendulum.



_ action is the ability of a substance to draw another substance into it such as with water and porous paper. 4 A ____ is a fluid that can freely form a distinct surface at the boundaries of its bulk material. 8 A _ __ flow is flow in which the divergence of velocity is zero. This is more precisely termed isochoric flow. is the SI unit of pressure or stress. 9 The is the upward force on an object produced by the surrounding fluid in which it is fully or partially immersed. 11 __ is defined as a substance that continually deforms under an applied shear stress regardless of the magnitude of the applied 12 A_ stress _____ is the force per unit area applied on a surface in a direction perpendicular to that surface. 15 _____ is fluid dynamics applied to liquids. 17 19 A flow is considered a _____ flow if the change in density of the flow with respect to pressure is non-zero. _ is a measure of the resistance of a fluid to deform under shear stress. 21 Identifying different flow regimes, such as laminar or turbulent flow, the _____ number is the ratio of inertial forces to viscous forces quantitfying the relative importance of these two types of forces for given flow conditions. **DOWN** or streamline flow occurs when a fluid flows in parallel layers, with no disruption between the layers.
 flow is a flow regime characterized by chaotic, stochastic property changes such as low momentum diffusion, high momentum

- convection, and rapid variation of pressure and velocity in space and time.
- ____ is mass per unit volume.
- 6 For a viscous liquid flowing through a cylindrical tube with constant circular cross-section, _______'s law relates the flow rate, pipe radius, pipe length, and the pressure difference between the two ends.
- ___'s Principle states that for an ideal fluid with no work being performed on the fluid, an increase in velocity occurs simultaneously with decrease in pressure or a change in the fluid's gravitational potential energy.
- 10 Fluid ______ is the sub-discipline of fluid mechanics dealing with fluids in motion.
- 13 The ___ _ flow rate, also volume flow rate, is the volume of fluid which passes through a given surface per unit time.
- 14 is a curve in the surface of a liquid produced in response to the surface of its container.
- ____ is an effect within the surface layer of a liquid that causes that layer to behave as an elastic sheet. 16 Surface _
- s law states that for all points at the same absolute height in a connected body of an incompressible fluid at rest, the fluid pressure is the same, even if additional pressure is applied on the fluid at some place.



- 4 A _______ is a mode of energy transfer from one place to another, often with little or no permanent displacement of the particles of the medium, but through oscillations around nearly fixed positions.
- 6 A ______ wave is a type of propagating disturbance characterized by an abrupt, nearly discontinuous, change in the characteristics of the medium.
- 7 In acoustics, a ______ is an interference between two sounds of slightly different frequencies, perceived as periodic variations in volume.
- 8 A _____ wave is a wave that travels through the Earth, most often as the result of a tectonic earthquake, sometimes from an explosion.
- 10 The _____ of sound describes how much distance a sound wave travels in a given amount of time.
- given amount of time.

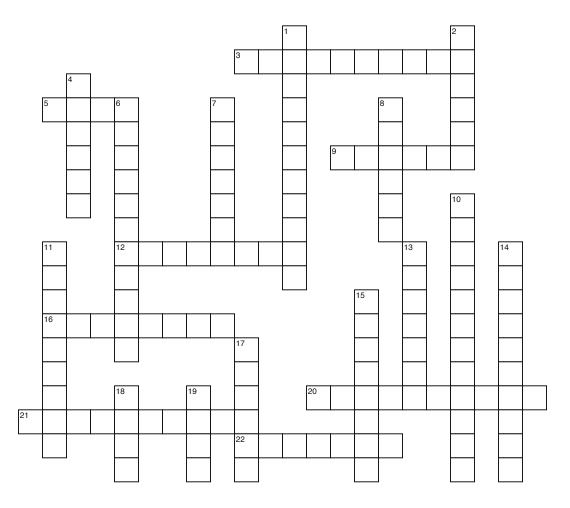
 12 ______ is the superposition of two or more waves that results in a new wave pattern.
- 13 _____ is the branch of physics concerned with the study of sound
- 15 ______ is the change in direction of a wave due to a change in its speed when a wave passes from one medium to another.
- 18 A _____ or harmonic is a natural

- resonance or vibration frequency of a system.
- 21 A ______ or overtone of a wave is a component frequency of the signal that is an integer multiple of the fundamental frequency.
- 23 _____ is the distance between repeating units of a propagating wave of a given frequency.
- 24 ______ is the change in direction of a wave front at an interface between two dissimilar media so that the wave front returns into the medium from which it originated.
- 25 _____ waves are waves that have vibrations along or parallel to their direction of travel.
- 26 A ______ is the point on a wave with the greatest positive value or upward displacement in a cycle.
- 27 _____ radiation, or light, is a selfpropagating wave in space with electric and magnetic components.

- 1 _____ refers to various phenomena associated with the bending, spreading and interference of waves passing by an object or aperture that disrupts the wave.
- 2 _____ refers to a shortening of a transmitted signal's wavelength.

- 3 A _____ wave is a wave that causes vibration in the medium in a perpendicular direction to its own motion.
- 5 The ______ of an oscillation or wave is the fraction of a complete cycle corresponding to an offset in the displacement from a specified reference point at time t = 0.
- 9 ______ is the measurement of the number of occurrences of a repeated event per unit of time.
- **11** A _____ or stationary wave is a wave that remains in a constant position.
- **14** The sound _____ is defined as the sound power per unit area.
- 16 The _____ tone is the lowest frequency in a harmonic series.
- 17 ______ is a disturbance consisting of vibrations traveling through matter as a longitudinal wave.
- 19 The ______ effect is the change in frequency and wavelength of a wave as perceived by an observer moving relative to the source of the waves.
- 20 The ______ seriew refers to the natural frequencies of an oscillator, limited to integer multiples of the lowest possible frequency.
- 22 The _____ is a logarithmic unit of measurement that expresses the magnitude of a physical quantity relative to a specified or implied reference level.

Temperature and Heat Flow



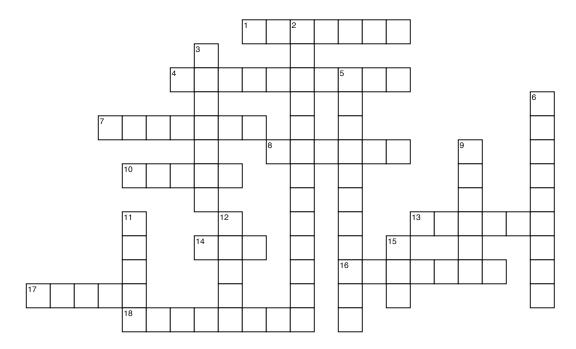
ACROSS

- 3 In the ______ scale, the melting point of water is 32 degrees and the boiling point is 212 degrees, placing the boiling and melting points of water exactly 180 degrees apart.
- 5 _____ is energy transferred from one body or system to another due to a difference in temperature.
- 9 The ______ is one of the seven SI base units. It corresponds to the absolute temperature scale where the coldest possible temperature is zero.
- 12 _____ zero describes a theoretical system that neither emits nor absorbs energy whose temperature is zero Kelvin.
- 16 _____ radiation is electromagnetic radiation of a wavelength longer than that of visible light, but shorter than that of radio waves.
- 20 The ______ of a material is the ratio of energy radiated by the material to energy radiated by a black body at the same temperature.
- 21 One of the major modes of heat transfer, ______, refers in the most general terms to the movement of currents within fluids.
- 22 A ______ is a unit of measurement for energy. In most fields, it has been replaced by the joule. However, a thousand-fold variation remains in common use within the field of nutrition.

- 1 A ______ is a device that measures temperature or temperature gradient.
- 2 The ______-Boltzmann law states that the total energy radiated per unit surface area of a black body in unit time is directly proportional to the fourth power of the black body's thermodynamic temperature.

- 4 The ______law of thermodynamics states that if two thermodynamic systems are in thermal equilibrium with a third, they are also in thermal equilibrium with each other.
- 6 ______ is defined as the average energy of microscopic motions of a single particle in the system per degree of freedom.
- 7 ______ energy is the energy portion of a system that increases with its temperature.
- 8 The ______-Petit law gives the classical expression for the specific heat capacity of a crystal due to its lattice vibrations.
- 10 Thermal ______, k, is the intensive property of a material that indicates its ability to conduct heat. It is used primarily in Fourier's Law for heat conduction.
- 11 Thermal ______ refers to electromagnetic waves emitted from the surface of an object which is due to the object's temperature.
- 13 Zero on the _____ scale was defined until 1954 as the melting point of ice and 100 degrees was defined as the boiling point of water under a pressure of one standard atmosphere. The definition is more formal today.
- 14 Heat ______ is the spontaneous transfer of thermal energy through matter.
- 15 ______ heat is the measure of the heat energy required to increase the temperature of a unit quantity of a substance by a certain temperature interval.
- 17 ______'s law describes the spectral radiance of electromagnetic radiation at all wavelengths from a black body at a given temperature as a function of frequency.
- 18 _____ transfer is the passage of thermal energy from a hot to a cold body.
- 19 _____'s displacement law states that there is an inverse relationship between the wavelength of the peak of the emission of a black body and its temperature.

Answer key - pg 82 Ideal Gas



ACROSS

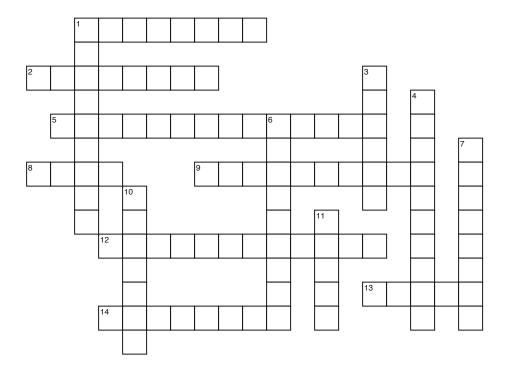
- 1 In statistical mechanics, a degree of ______ is a single scalar number describing the classical micro-state of a system.
- 4 ______'s law states that the ratio between the combining volumes of reagent gases and product can be expressed in small whole numbers.
- 7 ______ theory of gases attempts to explain macroscopic properties of gases by considering their molecular composition and motion.
- 8 ______'s law of partial pressures states that the total pressure exerted by a gaseous mixture is equal to the sum of the partial pressures of each component in a gas mixture.
- 10 The ______ gas law is the equation of state of a hypothetical gas consisting of identical particles of zero volume, with no intermolecular forces.
- 13 _____'s law of effusion states that the rate of effusion of a gas is inversely proportional to the square root of the mass of its particles.

- 14 _____ is one of the four major states of matter, consisting of freely moving atoms or molecules without a definite shape.
- 16 ______'s law states that at constant pressure, the volume of a given mass of an ideal gas increases or decreases by the same factor as its kelvin temperature increases or decreases.
- 17 A ______ gas is a hypothetical gas consisting of identical particles of zero volume with no intermolecular forces undergoing perfectly elastic collisions.
- 18 _____ is the process where individual molecules flow through a small pore without collisions.

- 2 The ______ theorem, which relates the temperature of a system to its average energies, depends on the idea that at thermal equilibrium, energy is shared equally among its various forms within the system.
- 3 Each gas in a mixture of ideal gases has a _____

- pressure which is the pressure which the gas would have if it alone occupied the volume.
- 5 ____ mechanics is the application of probability theory to the field of mechanics.
- **6** The _____ constant is the physical constant relating temperature to energy.
- 9 Root mean ______ speed is the measure of the speed of particles in a gas that is most convenient for problem solving within the kinetic theory of gases.
- 11 ______'s law states that the product of the pressure and volume for an enclosed ideal gas will be constant if temperature remains fixed.
- 12 The van der _____ equation is the general equation of state for a fluid composed of particles that have a non-zero size and a pairwise attractive inter-particle force.
- 15 The _____ constant is a physical constant used in equations of state. It is another name for the Boltzmann constant, though expressed in units of energy per kelvin per mole rather than energy per kelvin per particle.

1st Law of Thermodynamics



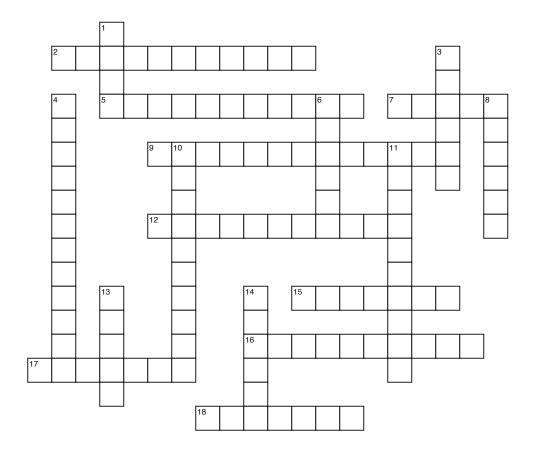
ACROSS

- 1 The ______ energy of a thermodynamic system is the total of the kinetic energy due to the motion of molecules and the potential energy associated with the vibrational and electric energy of atoms within molecules or crystals.
- 2 A ______ system, as contrasted with a open system, is a physical system that does not interact with its surroundings.
- 5 ______ is a branch of physics that studies the effects of changes in temperature, pressure, and volume on physical systems at the macroscopic scale by analyzing the collective motion of their particles.
- 8 Mechanical ______ is the amount of energy transferred by a force.
- 9 The mechanical ______ of heat was an expression of 19th century science stating that mechanical work may be transformed into heat, and conversely heat into work, with the magnitude of one always proportional to the other.
- 12 The _____ of energy states that the total amount of energy in any system remains constant, although it may change forms.
- **13** A thermodynamic ______ is the macroscopic condition of a thermodynamic system as described by its particular thermodynamic parameters.
- **14** A _____ process is a thermodynamic process in which the pressure stays constant.

1	A process	s, also called an isometric
	process or an isovol	umetric process, is a
	thermodynamic prod	cess that occurs without a
	change in volume.	
2	A thormodynamia	originally called

- **3** A thermodynamic ______, originally called a working substance, is defined as that part of the universe that is under consideration, separated by a real or imaginary boundary from the environment or surroundings
- 4 A ______ process is a thermodynamic process in which the temperature of the system stays constant.
- **6** A ______ process or an isocaloric process is a thermodynamic process in which no heat is transferred to or from the working fluid.
- 7 Heat ______ is the passage of thermal energy from a hot to a cold body.
- **10** A thermodynamic _____ may be defined as the evolution of a thermodynamic system proceeding from an initial state to a final state.
- 11 The ______ law of thermodynamics states that the increase in the internal energy of a thermodynamic system is equal to the amount of heat energy added to the system minus the work done by the system on the surroundings.

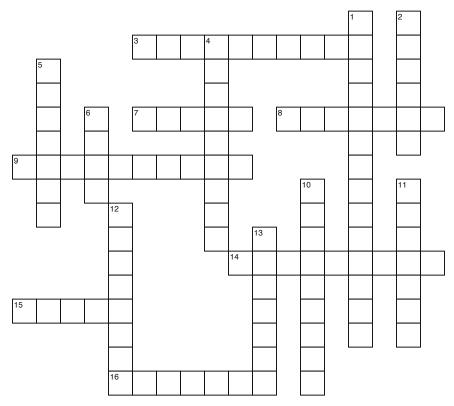
2nd Law of Thermodynamics



ACROSS

2	A thermodynamic system is said to be in thermodynamic when it's state is characterized by the minimum of a
	thermodynamic potential, such as the Helmholtz free energy. The coefficient of of a heat pump is the ratio of the output heat to the supplied work
	A thermodynamic is a series of thermodynamic processes which returns a system to its initial state is the process of removing heat from an enclosed space, or from a substance, and rejecting it elsewhere in order
12	to lower the temperature of the enclosed space or substance and then maintain that lower temperature. embodies the concept of a dynamical system where important mechanical modes, such as waves or oscillations,
12	lose energy over time, typically due to the action of friction or turbulence.
15	is a measure of the unavailability of a system's energy to do work.
16	A process, or cycle, can be reversed by means of infinitesimal changes in some property of the system without loss or dissipation of energy.
17	In thermodynamics, is often associated with the amount of order, disorder, and or chaos in a thermodynamic system.
18	The thermodynamic concept of can be described qualitatively as a measure of energy dispersal at a specific temperature.
DC	DWN
1	A heat is a machine or device that moves heat from one location to another via work.
3	The law of thermodynamics is an expression of the universal law of increasing entropy, stating that the entropy of an isolated system which is not in equilibrium will tend to increase over time, approaching a maximum value at equilibrium.
4	A probability is a probability measure defined over a state space instead of the sample space.
6	The cycle represents the most efficient cycle possible for converting a given amount of thermal energy into work
	or, conversely, for using a given amount of work for refrigeration purposes.
	A heat is a physical or theoretical device that converts thermal energy to mechanical output.
10	The thermal is a dimensionless performance measure of a thermal device such as an internal combustion
	engine, a boiler, or a furnace.
	In thermodynamics, a process is one during which the entropy of the system remains constant.
13	Thelaw of thermodynamics is an axiom of nature regarding entropy and the impossibility of reaching absolute zero of temperature.
14	's theorem sets a limit on the maximum amount of efficiency any possible engine can obtain based on the

difference between the hot and cold reservoir temperatures.



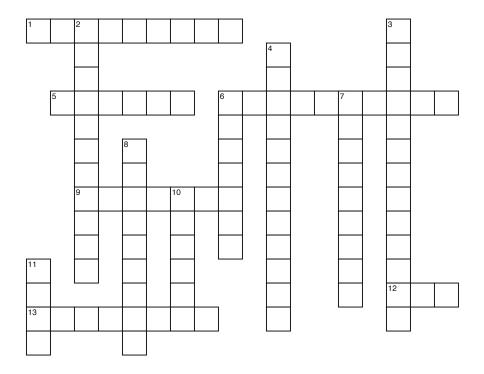
3	energy refers to the energy due to the
	interaction of electric charges with an electric field,
	and the energy stored in that field.

- 7 The _____ is the SI unit of capacitance.
- 8 _____ is the difference of electrical potential between two points of an electrical or electronic circuit, expressed in volts.
- **9** A ______, or electrical insulator, is a substance that is highly resistant to the flow of an electric current
- 14 Electric _____ is the positional energy per unit of charge associated with a static electric field, typically measured in volts.
- **15** A _____ charge is an idealized model of a charged particle as being located within a mathematical point with no dimensions.
- 16 ______'s law states that the magnitude of the electrostatic force between two point electric charges is directly proportional to the product of the magnitudes of each charge and inversely proportional to the square of the distance between the charges.

1	is the	oranch of physics that deals with
	the electrical phe	nomena which can be observed
	with stationary el	ectric charges.
2	Electric	is a fundamental conserved
		and the second of the second o

- property of some subatomic particles, which determines their electromagnetic interaction.

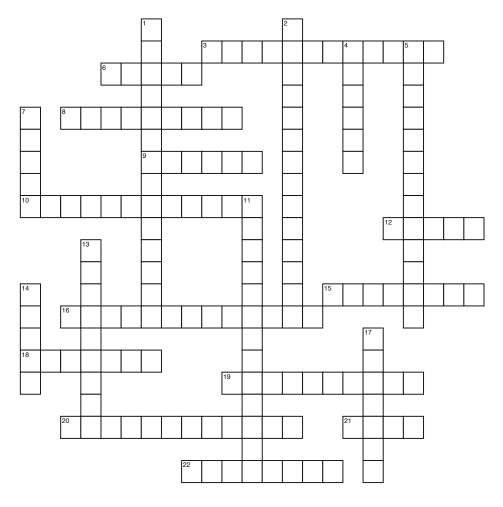
 4 A ______ is an electrical device that can store
- energy in the electric field between a pair of closely spaced conductors.
- 5 The linear, surface, or volume charge _____ is the amount of electric charge in a line, surface, or volume, measured in coulombs per metre, square metre, or cubic metre, respectively.
- **6** The _____ is the SI derived unit of electric potential difference or electromotive force.
- **10** Electrical _____ difference is the voltage present between two points, or the voltage drop transversely over an impedance.
- 11 Michael _____ was an English chemist and physicist who contributed significantly to the fields of electromagnetism and electrochemistry. He established the basis for the magnetic field concept in physics.
- 12 The space surrounding a charged particle or in the presence of a time-varying magnetic field has a property called a ______ field, which exerts force on other charged particles.
- 13 The _____ is the SI unit of electric charge.



1	's circuit laws are a pair of laws
	that deal with the conservation of charge
	and energy in electrical circuits.
5	current is the constant flow of
	electric charge.

- 6 Conduction is the movement of electrically charged particles through a transmission medium which can form an electric current in response to an electric field.
- **9** A ______ source is any device or system that produces an electromotive force between its terminals OR derives a secondary potential from a primary source of the electromotive force.
- _____'s law states that, in an electrical circuit, the current passing through a conductor between two points is proportional to the potential difference across the two points, and inversely proportional to the resistance between them.
- **13** A ______ is an electrical instrument that measures electrical resistance, the opposition to the flow of an electric current.

2	Electrical is a measure of how
	strongly a type of material opposes the
	flow of electric current.
3	A is a solid that has electrical
	conductivity in between that of a conductor
	and that of an insulator, and can be
	controlled over a wide range, either
	permanently or dynamically.
4	Electrical or specific is a
	measure of a material's ability to conduct
	an electric current.
6	Electric is the flow of electric
	charge.
7	Conductors, such as copper or aluminum,
	are materials with atoms having loosely
	held valence electrons.
8	A is a very high resistance
	ammeter used for measuring the electrical
	potential difference between two points in
	an electric circuit.
10	The is the SI unit of electric
	current.
11	Voltage is the reduction in
	voltage in an electrical circuit between the
	source and load



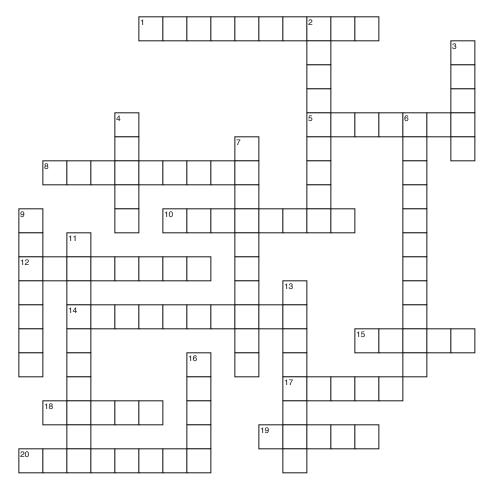
- 3 A ______ is an electromechanical transducer that produces a rotary deflection, through a limited arc, in response to electric current flowing through its coil.
- 6 The ______ point of a ferromagnetic material is the temperature above which it loses its characteristic ferromagnetic ability.
- 8 A ______ is a type of particle accelerator in which a perpendicular magnetic field causes the particles to spiral almost in a circle so that they reencounter the accelerating voltage many times.
- 9 _____'s circuital law relates the circulating magnetic field in a closed loop to the electric current passing through the loop.
- 10 Resulting from changes in the orbital motion of electrons, _______ is a weak repulsion from a magnetic field only exhibited by a substance in the presence of an externally applied magnetic field.
- **12** The _____ is the cgs unit of magnetic field.
- **15** A closed circulation of electric current creates a _____ dipole.
- 16 _____ is a form of magnetism which occurs only in the presence of an

- externally applied magnetic field, but unlike ferromagnetism, does not result in any retained magnetization.
- 18 Force The ______ is the force exerted on a charged particle in an electromagnetic field refering to the combined effects of any electric field and magnetic field.
- 19 ______ is the state when the material cannot absorb a stronger magnetic field, such that an increase of magnetization force produces no significant change in magnetic flux density.
- 20 _____ is the degree of magnetization of a material that responds linearly to an applied magnetic field.
- 21 The _____-Savart Law is an equation in electromagnetism that describes the magnetic field vector B in terms of the magnitude and direction of the source electric current, the distance from the current, and the magnetic permeability.
- 22 The _____ constant is equal to the vacuum permeability, also known as the permeability of free space.

DOWN

 _____ is defined as the phenomenon by which materials, such as iron, in an external magnetic field become magnetized and remain magnetized for a

- period after the material is no longer in the field.
- 2 _____ is the study of static magnetic fields.4 The magnetic dipole _____ is a
- measure of the strength of a magnetic source.
- 5 A _______ is a type of magnet in which the magnetic field is produced by the flow of an electric current.
- 7 Magnetic _____s permeate space around electric currents, magnetic dipoles, and changing electric fields, exerting a magnetic force on moving electric charges and magnetic dipoles.
- 11 ______ is a property of some materials that describes to what extent they are affected by magnetic fields and what magnetic field the material itself creates.
- 13 _____ is the magnetization left behind in a medium after an external magnetic field is removed.
- **14** The _____ is the SI derived unit of magnetic field.
- 17 The ______ center is a point around which the motion in a magnetic field of an electrically charged particle can be treated as the superposition of a relatively fast circular motion and a relatively slow drift of this point.



- is the phenomenon that the phase velocity of a wave depends on its frequency.
- reflection is the reflection of light from an uneven or granular surface such that an incident ray is seemingly reflected at a number of angles.
- ____s are a subcategory of radio waves with wavelengths shorter than one meter and longer than one millimeter.
- _ radiation is electromagnetic radiation of a wavelength longer than that of visible light, but shorter than that of radio waves. The name means below
- 12 The electromagnetic ____ range of all possible electromagnetic radiation.
- is the change in direction of a wave front at an interface between two dissimilar media so that the wave front returns into the medium from which it originated.
- _'s law is a formula used to angles of incidence and refraction for waves passing through a boundary
- describe the relationship between the between two different isotropic media. __ waves are electromagnetic

waves occurring within the portion of the electromagnetic spectrum which is a lower frequency than infrared radiation.

___ is electromagnetic radiation.

__ rays are forms of electromagnetic radiation or light emissions of a specific frequency produced from sub-atomic particle interaction, such as electron-positron

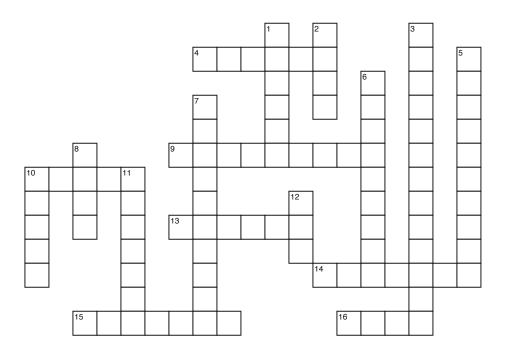
annihilation and radioactive decay.

reflection is the perfect, mirror-like reflection of light from a surface, in which light from a single incoming direction is reflected into a single outgoing direction.

- 2 The angle of _____ is a measure of deviation of a ray to a surface from straight on.
- 3 An optical is a glass or plastic strand of material designed to guide light along its length.
- _s are a form of electromagnetic radiation with a wavelength in the range of 10 to 0.01 nanometers. They are a form of ionizing radiation.
- _ light is electromagnetic radiation with a wavelength shorter than that of visible light, but longer than soft X-ravs.

- _ is the change in direction of a wave due to a change in its speed.
- 9 The _____ spectrum is the portion of the electromagnetic spectrum that can be detected by the human eye.
- 11 The ___ _ index of a medium is a measure for how much the speed of light is reduced inside the medium.
- 13 Total ____ reflection is an optical phenomenon that occurs when a ray of light strikes a medium boundary at an angle larger than the critical angle with respect to the normal to the surface.
- _ is a device that produces coherent radiation, typically in the form of a narrow, low-divergence beam and with a well-defined wavelength.

Geometric Optics

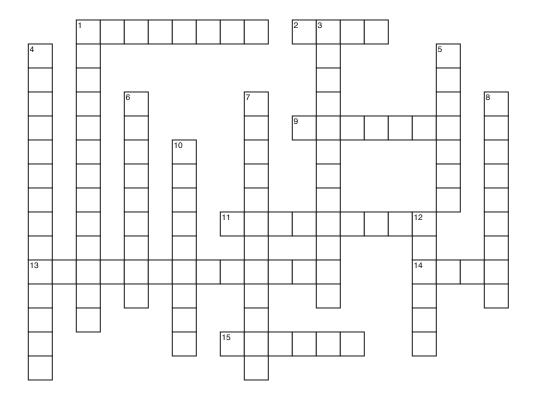


ACROSS

- 4 A ______ or singlet lens consists of a single uncomplicated element. Examples include a magnifying glass or a lens in a pair of simple reading glasses.
- **9** A ______ is an instrument designed for the observation of remote objects and the collection of electromagnetic radiation.
- 10 The _____ of view is the angular extent of the observable world that is seen at any given moment.
- 13 Also called near- or short-sightedness, _____ is a refractive defect of the eye with image focus in front of the retina when accommodation is relaxed.
- 14 Ray _____ is a general technique from geometrical optics of modeling the path taken by light by following rays of light as they interact with optical surfaces.
- 15 A _____ image is an image in which the outgoing rays from a point on the object never actually intersect at a point.
- **16** A _____ lens is a lens with a thicknessthat is negligible compared to the focal length of the lens.

1	is a branch of physics that describes the
	behavior and properties of light and the interaction
	of light with matter.
2	A image is a representation of an actual
	object formed by rays of light passing through the
	image.
3	is the process of enlarging something
	only in appearance, not in physical size.
5	A or refractor telescope is a dioptric
	telescope that uses a lens as its objective to form
	an image.
6	Also known as hypermetropia, farsightedness or
	longsightedness, is a defect of vision
	caused by an imperfection in the eye causing
	inability to focus on near objects.
7	•
	that occurs with aging.
8	A is an optical device with perfect or
•	approximate axial symmetry which transmits and
	refracts light, concentrating or diverging the beam.
10	The length of an optical system is a
	measure of how strongly it converges or diverges
	light.
4.4	A is a unit of measurement of the optical
	·
	power of a lens or curved mirror, which is equal to
	the reciprocal of the focal length measured in
	meters.
12	A is an idealized narrow beam of light.

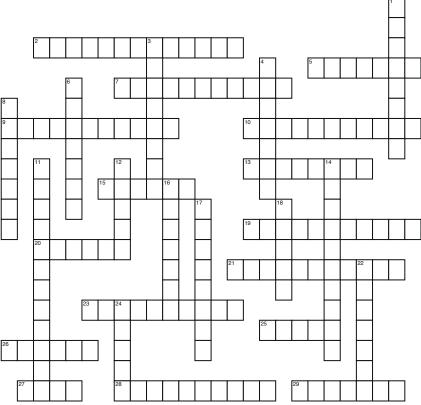
Wave Optics



ACROSS

- 1 At the angle of incidence known as ______'s angle, light for which the electric field of the light waves lies in the same plane as the incident ray and the surface normal cannot be reflected with polarization.
- 2 The _____ disc is the diffraction pattern resulting from a uniformly illuminated circular aperture.
- 9 A diffraction ______ is an optical component with a surface covered by a regular pattern of parallel lines, typically with a distance between the lines comparable to the wavelength of light.
- 11 The ______ interferometer is a common configuration for optical interferometry in which an interference pattern is produced by splitting a beam of light into two paths, bouncing the beams back and recombining them.
- **13** _____ refers to electromagnetic radiation of a single wavelength.
- 14 Physical or _____ optics, is the branch of optics which studies interference, diffraction, polarization, and other phenomena for which the ray approximation of geometric optics is not valid.
- 15 In the _____-slit experiment, light is shone at a solid thin plate that has two slits cut into it. A photographic plate is set up to record what comes through those slits.

- 1 ______ is the decomposition of a ray of light into an ordinary ray and an extraordinary ray when it passes through certain types of material, such as calcite crystals or boron nitride, depending on the polarization of the light.
- 3 _____ is the superposition of two or more waves that results in a new wave pattern.
- 4 _____ is the technique of superposing two or more waves, to detect differences between them.
- 5 The ______-Fresnel principle recognizes that each point of an advancing wave front is in fact the center of a fresh disturbance and the source of a new train of waves.
- **6** ______ is the property of wave-like states that enables them to exhibit interference. It is a measure of how perfectly the waves can cancel due to destructive interference.
- 7 ______ is the property of electromagnetic waves that describes the direction of the transverse electric field.
- **8** A ______ is a device that converts an unpolarized or mixed-polarization beam of electromagnetic waves into a beam with a single polarization state.
- **10** A _____ is the locus of points in a wave having the same phase.
- 12 The phenomenon of ______'s rings is an interference pattern caused by the reflection of light between two surfaces a spherical surface and an adjacent flat surface.



- 2 The _______effect is a quantum electronic phenomenon in which electrons are emitted from matter after the absorption of energy from electromagnetic radiation.
- 5 Einstein's ______ theory of relativity not only widened the postulate of relativity but added the second postulate that all observers will always measure the speed of light to be the same no matter what their state of uniform linear motion is.
- 7 Length _____, according to the special theory of relativity, is the physical phenomenon of a decrease in length detected by an observer in objects that travel at any non-zero velocity relative to that observer.
- 9 The Heisenberg _____ principle gives a lower bound on the product of the standard deviations of position and momentum for a system, implying that it is impossible for a particle to have an arbitrarily well-defined position and momentum simultaneously.
- 10 Electron is a technique used to study matter by firing electrons at a sample and observing the resulting interference pattern.
- 13 The _____ is the antiparticle of the electron.
- 15 The _____ constant is a physical constant that is used to describe the sizes of quanta, which plays a central role in the theory of quantum mechanics.
- 19 Mass-energy _____ is the concept that any mass has an

- associated energy and vice versa.
- 20 _____'s law describes the spectral radiance of electromagnetic radiation at all wavelengths from a black body at a certain temperature.
- 23 A ______ particle or fundamental particle is a particle not known to have substructure; that is, it is not known to be made up of smaller particles.
- 25 The speed of _____ in vacuum is the speed in a vacuum of anything having zero rest mass.
- 26 A _____ is a volume of space that is essentially empty of matter, such that its gaseous pressure is much less than standard atmospheric pressure.
- 27 The ____ interaction is one of the four fundamental interactions of nature. It is due to the exchange of the heavy W and Z bosons. Its most familiar effect is beta decay.
- 28 The ______ of simultaneity is the concept that simultaneity is not absolute, but dependent on the observer.
- 29 The de ______ hypothesis is the statement that all matter (any object) has a wave-like nature, in other words, wave-particle duality.

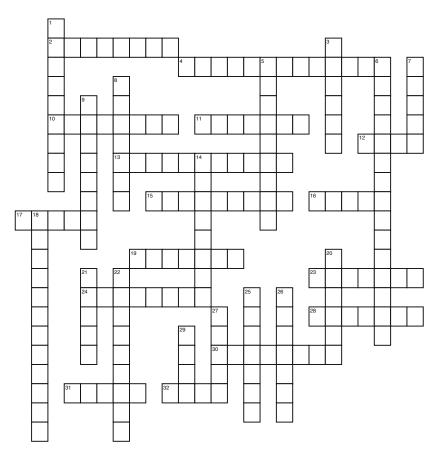
DOWN

 Time ______ is the phenomenon whereby an observer finds that another's clock which is physically identical to their own is ticking at a

- slower rate as measured by their own clock.
- 3 Albert _______ (1879 1955) was a German-born theoretical physicist. He is best known for his theory of relativity and, specifically, mass-energy equivalence.
- 4 The event ______ is a general term for a boundary in spacetime, such as an area surrounding the black hole, beyond which events cannot affect an outside observer.
- 6 The _____ transformation converts between two different observers' measurements of space and time in a manner consistent with special relativity, where one observer is in constant motion with respect to the other.
- 8 theory is the branch of physics which is based on quantization, which began in 1900 when Max Planck published his theory explaining the emission spectrum of black bodies.
- 11 Corresponding to most kinds of particles is an associated _____ with the same mass and opposite charges.
- 12 A _____ hole is a region of space in which the gravitational field is so powerful that nothing can escape after having fallen past the event horizon.
- 14 A principle of _____ is a criterion for judging physical theories, stating that they are inadequate if they do not prescribe the exact same laws of physics in certain similar situations.
- 16 _____scattering is the decrease in energy of an X-ray or gamma ray photon, when it interacts with matter.

- 7 The _____Model of particle physics is a quantum field theory developed between 1970 and 1973 which describes three of the four known fundamental interactions between the elementary particles that make up all matter.
- 18 The ______ is one of the two basic constituents of matter in particle physics (the other is the lepton).
- 2 _____ relativity is the geometrical theory of gravitation published by Albert Einstein in 1915-16.
- 24 In the late 19th century, luminiferous _____ was the term used to describe a medium for the propagation of light. Today this theory is regarded as a superseded scientific theory.

Nuclear Physics



ACROSS

- 2 Proton ______, also known as proton radioactivity, is a type of radioactive decay in which a proton is ejected from a nucleus.
- 4 Nuclear _____ is the conversion of one chemical element or isotope into another, which occurs through nuclear reactions.
- 10 _____ uranium is a sample of uranium in which the percent composition of uranium-235 has been increased through the process of isotope separation.
- 11 _____ physics is the branch of physics concerned with the nucleus of the atom.
- 12 In nuclear physics, _____ decay is a type of radioactive decay in which an electron or a positron is emitted.
- 13 A quantity is said to be subject to ______ decay if it decreases at a rate proportional to its value.
- 15 _____ in nuclear physics describes energy in the form of waves or moving subatomic particles.
- 16 _____ water is water which contains a higher

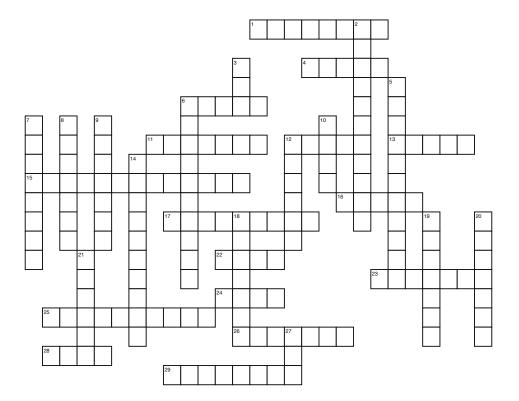
- proportion than normal of the isotope deuterium as deuterium oxideor as deuterium protium oxide.
- 7 _____ rays are forms of electromagnetic radiation of a specific frequency produced from sub-atomic particle interaction, such as electronpositron annihilation and radioactive decay.
- 19 _____ is a radioactive isotope of hydrogen with a nucleus containing one proton and two neutrons.
- 23 The _______ force or residual strong force is the force between two or more nucleons. It is responsible for binding of protons and neutrons into atomic nuclei.
- 24 _____ capture, sometimes called inverse beta decay, is a decay mode for isotopes that will occur when there are too many protons in the nucleus of an atom and insufficient energy to emit a positron.
- 28 In nuclear engineering, a
 _____ material is one that
 is capable of sustaining a
 chain reaction of nuclear
 fission.
- of beta decay, sometimes

- referred to as beta plus decay.
- 31 The ______ is a unit of radioactivity which is roughly the activity of 1 gram of the radium isotope 226Ra.
- 32 _____ particles are highenergy, high-speed electrons or positrons emitted by certain types of radioactive nuclei.

- 1 The ______ is the SI derived unit of radioactivity equal to one nucleus decay per second.
- 3 The _______ interaction is today understood to represent the interactions between quarks and gluons as detailed by the theory of quantum chromodynamics.
- 5 In nuclear engineering, a neutron ______ is a medium which reduces the velocity of fast neutrons, thereby turning them into thermal neutrons capable of sustaining a nuclear chain reaction.

6	is the process of
	creating new atomic nuclei
	from preexisting nucleons.
7	particles consist of

- two protons and two neutrons bound together into a particle identical to a helium nucleus.
- 8 A _____ is a collective name for two baryons: the neutron and the proton.
- 9 A ______ mass is the smallest amount of fissile material needed for a sustained nuclear chain reaction.
- 14 Neutron ______ is a type of radioactive decay in which an atom contains excess neutrons and a neutron is simply ejected from the nucleus.
- 18 ____s, the antiparticles of neutrinos, are neutral particles produced in nuclear beta decay.
- 20 Nuclear ______ is the process by which multiple atomic particles join together to form a heavier nucleus.
- 21 Radioactive ______ is the process in which an unstable atomic nucleus loses energy by emitting radiation in the form of particles or electromagnetic waves.
- 22 ______, also called heavy hydrogen, is a stable isotope of hydrogen with a natural abundance in the oceans of Earth of approximately one atom in 6500 of hydrogen.
- 25 Nuclear _____ is the splitting of the nucleus of an atom into lighter nuclei often producing photons in the form of gamma rays, free neutrons and other subatomic particles as by-products.
- 26 A _____ rod is a rod made of chemical elements capable of absorbing many neutrons without fissioning themselves. They are used in nuclear reactors to affect the rate of fission.
- 27 ______ decay is a type of radioactive decay in which an atomic nucleus emits two protons and two neutrons bound together into a particle identical to a helium nucleus.
- 29 The half-_____ of a quantity, subject to exponential decay, is the time required for the quantity to decay to half of its initial value.

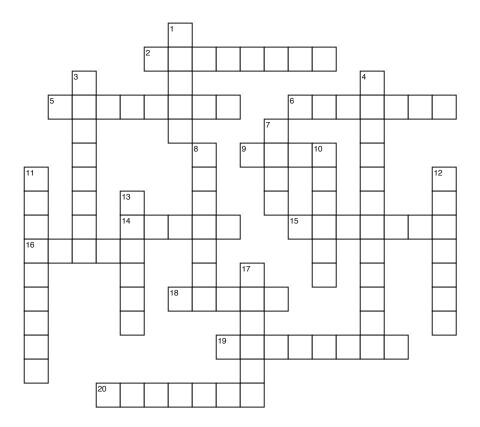


- 1 An element's emission ______ is the relative intensity of electromagnetic radiation of each frequency it emits when it is excited.
- 4 A ______ body is an object that absorbs all electromagnetic radiation that falls onto it. No radiation passes through it and none is reflected.
- 6 An electron _______, also known as a main energy level, is a group of atomic orbitals with the same value of the principal quantum number.
- 11 _____ mechanics is the study of the relationship between energy quanta and matter, in particular between photons and valence shell electrons.
- 12 _____ particles consist of two protons and two neutrons bound together into a particle identical to a helium nucleus.
- 13 The _____ exclusion principle explains why matter occupies space exclusively for itself and does not allow other material objects to pass through it, while at the same time allowing light and radiation to pass.
- 15 The electron ______ is the arrangement of electrons in an atom, molecule, or other physical structure such as a crystal.
- 16 The quantum _____ of a system corresponds to a set of numbers that fully describe a quantum system.
- 17 The _____ quantum number has the greatest correlation to energy of the quantum numbers describing the unique quantum state of an electron in an atom.
- 22 A _____ is the smallest particle still characterizing a chemical element
- 23 _____s are any of the several different forms of an element with nuclei having the same number of protons but

- different numbers of neutrons.
- 24 The _____ model depicts the atom as a small, positively charged nucleus surrounded by electrons that travel in circular orbits around the nucleus.
- 25 The _____ model showed that the plum pudding model of the atom of J. J. Thomson was incorrect, presenting the atom as containing a central charge concentrated into a very small volume in comparison to the rest of the atom.
- 26 A ______ state of a system is any quantum state of the system that has a higher energy than the ground state.
- 28 ______'s rules are a simple set of rules used to determine the term symbol that corresponds to the ground state of a multi-electron atom.
- 29 _____ is a chemical element represented by the symbol H and an atomic number of 1.

- 2 The Heisenberg _____ principle gives a lower bound on the product of the standard deviations of position and momentum for a system, implying that it is impossible to have a particle that has an arbitrarily well-defined position and momentum simultaneously.
- 3 The purpose of Robert Millikan and Harvey Fletcher's ______-drop experiment (1909) was to measure the electric charge of the electron.
- 5 The law of definite ______ states that a chemical compound always contains exactly the same proportion of elements by mass.
- 6 Observation of the phenomenon of Rutherford ______ of alpha particles incident on gold foil led to the development of the orbital theory of the

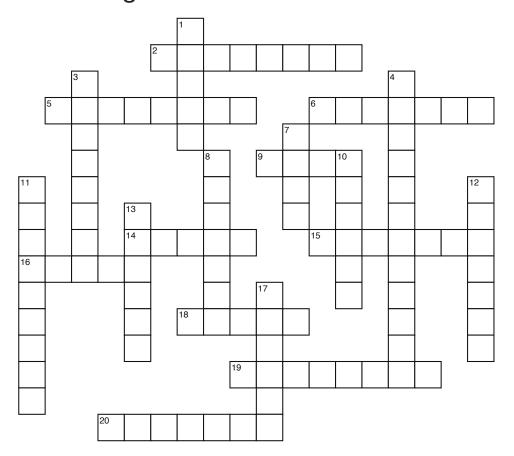
- atom.
- 7 The ______ is a fundamental subatomic particle that carries a negative electric charge.
- 8 A _____leap is a change of an electron from one energy state to another within an atom.
- 9 An atomic ______ is a mathematical description of the region in which an electron may be found around a single atom.
- 10 _____ is the angular momentum intrinsic to a body, as opposed to orbital angular momentum, which is the motion of its center of mass about an external point.
- 12 The ______ is used to determine the electron configuration of an atom, molecule or ion, postulating a hypothetical process in which an atom is built up by progressively adding electrons.
- 14 Ernest ______ was a nuclear physicist who pioneered the orbital theory of the atom through his discovery of scattering off the nucleus with his gold foil experiment.
- 18 _____ rays are streams of electrons observed in vacuum tubes.
- 19 The ______ is a subatomic particle with no net electric charge and a mass that is slightly more than a proton
- 20 ______ electrons are the electrons contained in the outermost electron shell of an atom.
- 21 The ____ is a subatomic particle with an electric charge of one positive fundamental unit, a diameter of about 1.5 fm femtometer, and a mass that is about 1836 times the mass of an electron.
- 27 A ______ is an atom or molecule which has lost or gained one or more electrons, making it negatively or positively charged.



- is defined as a sufficiently stable electrically neutral group of at least two atoms in a definite arrangement held together by _____ is a concept in chemistry which describes how bonding electrons may or may not be equally shared between atoms. 5 Chemical
- 6 A molecular _____ is a region in which an electron may be found in a molecule.
- pair is a valence electron pair without bonding or sharing with other atoms.
- $_$ rule is a simple chemical rule of thumb that states that atoms tend to combine in such a way that they each have a noble gas configuration in their valence shells.
- 15 Molecular ___ __ theory is a method for determining molecular structure in which electrons are not assigned to individual bonds between atoms, but are treated as under the influence of the nuclei in the whole molecule.
- _ is the number of bonds between a pair of atoms.
- _ structures, also called electron-dot structures or electron-dot diagrams, are diagrams that show the bonding between atoms of a molecule, and the lone pairs of electrons that may exist in the molecule.
- 19 bonding is a form of attraction-to-repulsion stability that forms between atoms when they share electrons.
- 20 electrons are the electrons contained in the outermost electron shell of an atom.

- bond (or electrovalent bond) is a type of chemical bond based on electrostatic forces between two oppositely-charged ions.
- 3 A chemical ______ is a substance consisting of two or more elements chemically-bonded together in a fixed proportion by mass.
- energy is defined as the standard enthalpy change when a bond is cleaved by homolysis at 0K (absolute zero).
- is the physical process responsible for the attractive interactions between atoms and molecules which confers stability to diatomic and polyatomic chemical compounds.
- _ bond theory explains the nature of a chemical bond in a molecule in terms of atomic valencies.
- is the enthalpy change involved with breaking up a neutral molecule into subtitutent neutral elements. **10** Bond
- is a tool used to represent and model certain types of non-classical molecular structures arising when no single conventional model showing electrons shared exclusively by two atoms can actually represent the observed molecule.
- is a measure of the number of chemical bonds formed by the atoms of a given element.
- $_$ charge is a partial charge on an atom in a molecule assigned by assuming that electrons in a chemical bond are shared equally between atoms, regardless of relative electronegativity.
- __ moment is a measure for the polarity of a chemical bond within a molecule.

Chemical Bonding

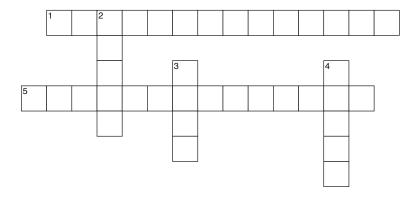


ACROSS

is defined as a sufficiently stable electrically neutral group of at least two atoms in a definite arrangement held together by strong chemical bonds. 5 Chemical ______ is a concept in chemistry which describes how bonding electrons may or may not be equally shared between atoms. 6 A molecular _____ is a region in which an electron may be found in a molecule. 9 A _____ pair is a valence electron pair without bonding or sharing with other atoms. ____ rule is a simple chemical rule of thumb that states that atoms tend to combine in such a way that they each have a noble gas configuration in their valence shells. theory is a method for determining molecular structure in which electrons are not assigned to individual bonds between atoms, but are treated as under the influence of the nuclei in the whole molecule. is the number of bonds between a pair of atoms. 18 ______ structures, also called electron-dot structures or electron-dot diagrams, are diagrams that show the bonding between atoms of a molecule, and the lone pairs of electrons that may exist in the molecule. 19 ______ bonding is a form of attraction-to-repulsion stability that forms between atoms when they share electrons. electrons are the electrons contained in the outermost electron shell of an atom. **DOWN** bond (or electrovalent bond) is a type of chemical bond based on electrostatic forces between two oppositely-charged ions. ____ is a substance consisting of two or more elements chemically-bonded together in a fixed proportion by mass.

- ____ energy is defined as the standard enthalpy change when a bond is cleaved by homolysis at 0K (absolute zero). ___ is the physical process responsible for the attractive interactions between atoms and molecules which confers stability to diatomic and polyatomic chemical compounds. 8 ______ bond theory explains the nature of a chemical bond in a molecule in terms of atomic valencies. 10 Bond ____ $_$ is the enthalpy change involved with breaking up a neutral molecule into subtitutent neutral elements.
- is a tool used to represent and model certain types of non-classical molecular structures arising when no single conventional
- model showing electrons shared exclusively by two atoms can actually represent the observed molecule.
- ____ is a measure of the number of chemical bonds formed by the atoms of a given element.
- ___ charge is a partial charge on an atom in a molecule assigned by assuming that electrons in a chemical bond are shared equally between atoms, regardless of relative electronegativity.
- _ moment is a measure for the polarity of a chemical bond within a molecule.

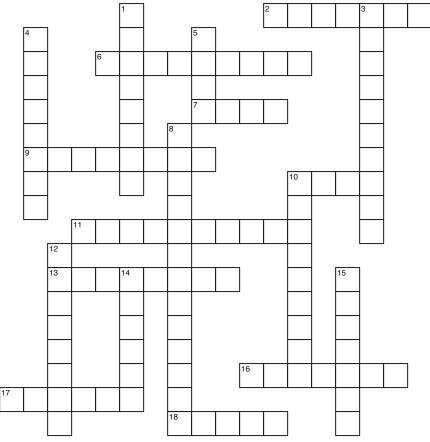
Intermolecular Force



ACROSS

- 1 _____ refers to the physical property of a molecule that is repelled from a mass of water
- 5 A ______ force is a force that acts between stable molecules or between functional groups of macromolecules which are generally much weaker than the chemical bonding forces.

- 2 The water _____ consists of two water molecules loosely bound by a hydrogen bond. It is the smallest water cluster.
- 3 A hydrogen ______ is a special type of dipole-dipole interaction that exists between an electronegative atom and a hydrogen atom bonded to another electronegative atom.
- 4 The name van der _____ force is sometimes used as a synonym for the totality of noncovalent forces which act between stable molecules.



2	A reactant or
	is a
	substance consumed
	during a chemical
	reaction.
6	The
	formula of a chemical
	compound is a simple
	expression of the
	relative number of
	each type of atom in
	it.
7	The law of
	conservation of
	states that
	the total amount of
	matter within a closed
	system will remain
	constant, regardless
	of the processes
	acting inside the
	system.
9	The law of
	proportions states
	that a chemical

	contains exactly the
	same proportion of
	elements by mass.
10	The is the
	SI base unit that
	measures an amount
	of substance, equal to
	Avogadro's number of
	entities.
11	The
	formula of a chemical
	compound is a
	graphical
	representation of the
	molecular structure
	showing how the
	atoms are arranged.
13	A chemical
	is a symbolic
	representation of a
	chemical reaction.
16	A chemical
	is a concise way of
	expressing
	information about the

atoms that constitute

compound always

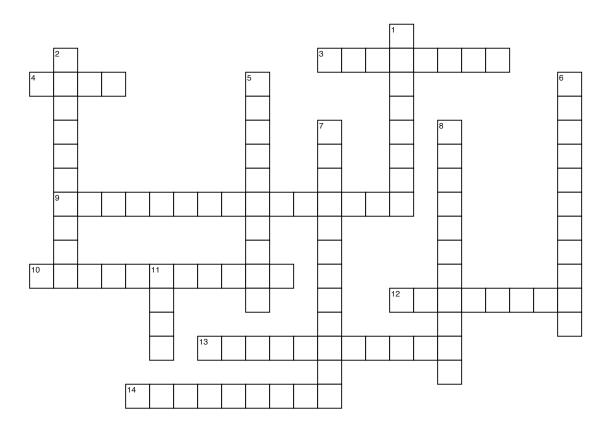
	15	
		8
17	a particular chemical compound. Gram mass is the mass in grams of one mole of atoms in an element.	10
18		
	obtained in a chemical reaction.	12
DC	OWN	
1	The reagent is the chemical that determines how far a reaction would go	14
	because the chemical in question is the reagent that would get completely used up, causing the	15

reaction to stop. 3 The _____ is a measurement unit

used in chemistry and

	the biological
	sciences, which
	measures of a
	substance's ability to
	combine with other
	substances, an
	expression frequently
	used in the context of
	normality.
4	's number is
	the number of entities
_	in one mole
5	Although the definition is more
	formal now, the
	was
	originally defined as
	the absolute weight of
	a volume of pure
	water equal to the
	cube of the hundredth
	part of a metre, and at
	the temperature of
	melting ice.
8	is the
	calculation of
	quantitative relationships of the
	reactants and
	products in chemical
	reactions.
10	The mass
	of a substance is the
	mass of one molecule
	of that substance,
	relative to the unified
	atomic mass unit
	(equal to 1/12 the
	mass of one atom of carbon-12).
12	
12	is a process that
	results in the
	interconversion of
	chemical substances.
14	The mass
	is the mass of an
	atom at rest, most
	often expressed in
4-	unifed amu.
15	
	substance that forms as a result of a
	chemical reaction.
	chombal rodollom.

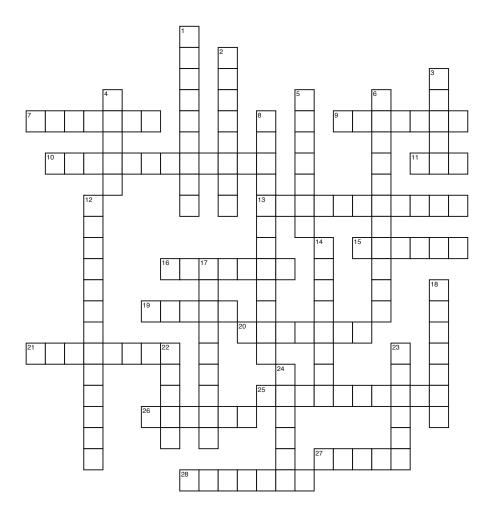
Thermochemistry



ACROSS

- 3 The ______ enthalpy of combustion is the change in enthalpy of the total reacting system when one mole of a substance completely reacts with oxygen, and is observed at 298K and 1 atmospheric pressure
- The ______-Haber Cycle is an approach to analyzing reaction energies involving the formation of an ionic compound from the reaction of a group I or group II metal with a non-metal.
- **9** _____ is the study of the heat evolved or absorbed in chemical reactions.
- **10** _____ describes a process or reaction that absorbs energy in the form of heat.
- 12 Standard temperature and ______ is a standard set of conditions for experimental measurements, to enable comparisons to be made between sets of data.
- **13** A reaction ______ is an instrument that enables the energy being released or absorbed by a reaction to be measured.
- 14 The standard enthalpy of ______ of a compound is the change of enthalpy that accompanies forming 1 mole of a substance in its standard state from its constituent elements in their standard states

- 1 The _____ or heat content is a quotient or description of thermodynamic potential of a system equivalent to the sum of the internal energy of the system plus the product of its volume multiplied by the pressure exerted on it by its surroundings.
- 2 The heat of _____ is the energy released when a compound undergoes complete combustion with oxygen.
- 5 _____ describes a process or reaction that releases energy in the form of heat.
- **6** A ______ is a device used for measuring the heat of chemical reactions or physical changes as well as heat capacity.
- 7 Bond ______ energy is defined as the standard enthalpy change when a bond is cleaved by homolysis, with reactants and products of the homolysis reaction at 0K (absolute zero).
- **8** _____ is the science of measuring the heat of chemical reactions or physical changes.
- 11 Developed through conceptualizing cyclic reaction processes in which the return path is different than the forward path, _______'s Law of Heat Summation is used to predict the enthalpy change regardless of the path through which it is to be determined.



- 7 _____tension is an effect within the surface layer of a liquid that causes that layer to behave as an elastic sheet.
- In a mixture of ideal gases, each gas has a _____ pressure which is the pressure which the gas would have if it alone occupied the volume.
- 10 The enthalpy of _____ is the energy required to transform a given quantity of a substance into a gas, measured at the boiling point of the substance.
- 11 The _____ point is the temperature to which a given parcel of air must be cooled, at constant barometric pressure, for water vapor to condense into water.
- 13 _____ is the process by which

- molecules in a liquid state spontaneously become gaseous without being heated to boiling point.
- 15 A _____ is typically an ionized gas, considered to be a distinct state of matter, apart from gases, because of its unique properties.
- 16 A phase _____ is a type of graph used to show the equilibrium conditions between the thermodynamically-distinct phases.
- 19 _____ is the gas phase component present along with a solid or liquid sample of matter which does not completely fill its container.
- 20 The _____ point of a substance is the maximum temperature at which a liquid can remain a liquid at a given pressure.
- **21** A _____ is a

- curve in the surface of a liquid and is produced in response to the surface of the container or another object.
- 25 _____ is the onset of a phase transition in a small region such as with the formation of a bubble or of a crystal from a liquid.
- 26 A ______ is a fluid that can freely form a distinct surface at the boundaries of its bulk material.
- 27 A _____ of matter is one of the many ways that matter can interact with itself to form a macroscopic, homogenous phase.
- 28 The ______ point of a crystalline solid is the temperature range at which it changes state from solid to liquid.

DOWN

1 _____ action is the ability of a

substance to draw another substance into it.

is the

process whereby a liquid turns to a solid.

3 A _______ is a set of states of a macroscopic physical system that have

relatively uniform

and physical

chemical composition

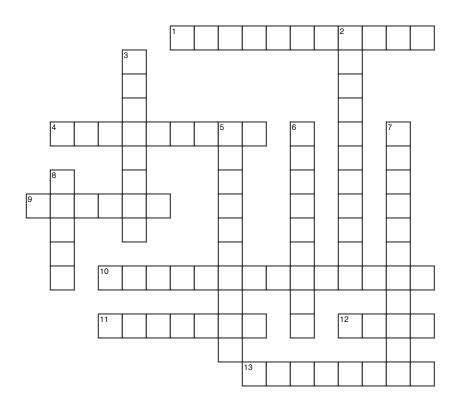
- properties

 4 _____ pressure is the pressure of a gaseous phase in equilibrium with its
- non-gaseous phases.

 5 A ______ structure is composed of a motif, a set of atoms arranged in a particular way, and a lattice.
- 6 _____solids are a class of solids that have regular or nearly-regular structures, meaning that the atoms in these solids are arranged in an orderly manner

- 8 ______ is the change in matter of a substance to a denser phase, such as a gas to a liquid.
- 12 A ______fluid is any substance at a temperature and pressure above its thermodynamic critical point.
- 14 A ______ point specifies the conditions (temperature, pressure) at which the liquid state of the matter ceases to exist.
- 17 A _____ solid is a solid in which there is no long-range order of the positions of the atoms.
- 18 A _____ curve is a line graph that represents the change of phase of matter, typically from either a gas to a solid or from a liquid to a solid.
- 22 A _____ object is in the states of matter characterized by resistance to deformation and changes of volume.
- 23 The _____ point of a substance is the temperature and pressure at which three phases (gas, liquid, and solid) of that substance may coexist in thermodynamic equilibrium.
- 24 The enthalpy of is the amount of thermal energy which must be absorbed or evolved at the melting point for 1 mole of a substance to change states from a solid to a liquid or vice versa.

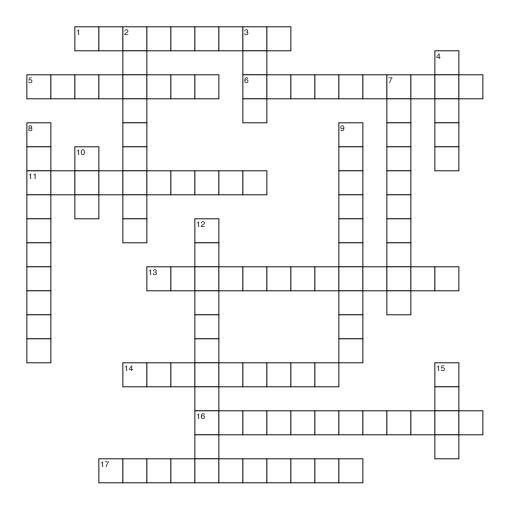
Chemical Thermodynamics



ACROSS

- 1 A ______ process is a chemical reaction in which a system releases free energy and moves to a lower, more thermodynamically stable, energy state.
- 4 Le _______'s principle states that if a chemical system at equilibrium experiences a change in concentration, temperature, volume, or total pressure, the equilibrium will shift in order to partially counter-act the imposed change.
- **9** The entropy of ______ is the change in the entropy when two different chemical substances or components are mixed.
- 10 Chemical ______ is the mathematical study of the interrelation of heat and work with chemical reactions or with a physical change of state within the confines of the laws of thermodynamics.
- **11** A _____ equilibrium occurs when two reversible processes proceed at the same rate.
- **12** The term thermodynamic _____ energy is a measure of the amount of work that can be extracted from a system.
- in chemistry is a measure of how different molecules in a non-ideal gas or solution interact with each other, extending the idea of concentration to more complex systems.

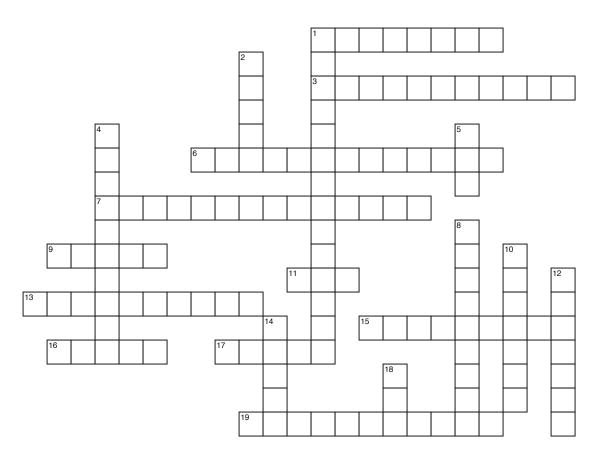
- 2 The _____ constant is the reaction quotient describing the state in which the chemical activities or concentrations of the reactants and products have no net change over time.
- 3 The reaction ______ is a quantitative measure of the extent of reaction, the relative proportion of products and reactants present in the reaction mixture at some instant of time.
- **5** A ______ reaction (also called an unfavorable reaction or a nonspontaneous reaction) is a chemical reaction in which the standard change in free energy is positive.
- **6** A ______ reaction is a chemical reaction where the variation of Gibbs free energy is negative.
- 7 An activity ______ is a factor used in thermodynamics to account for deviations from ideal behaviour in a mixture of chemical substances.
- **8** The ______ free energy is a thermodynamic potential which measures the useful or process-initiating work obtainable from an isothermal, isobaric thermodynamic system.



- 1 A reaction ______ is the step by step sequence of elementary reactions by which overall chemical change occurs.
- 5 Chemical ______ is the study of reaction rates in a chemical reaction.
- 6 A ______ reaction is a chemical reaction in which one or more chemical species react directly to form products in a single reaction step and with a single transition state.
- 11 The _______ state of a chemical reaction is a particular configuration along the reaction coordinate defined as the state corresponding to the highest energy along this reaction coordinate.
- 13 _____ catalysis describes catalysis where the catalyst is in a different phase to the reactants.
- 14 A ______ complex is a transitional structure in a chemical reaction that results from an effective collision between molecules and that persists while old bonds are breaking and new bonds are forming.
- **16** A ______ in a chemical reaction is a molecular entity with a lifetime appreciably longer than a molecular vibration that is formed from the reactants and reacts further to give the products of a chemical reaction.
- 17 _____ catalysis describes catalysis where the catalyst is in the same phase as the reactants.

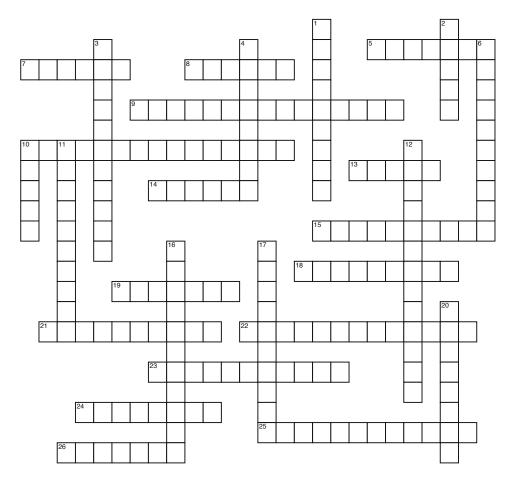
- 2 ______ is the increase in rate of a chemical reaction by means of a substance called a catalyst.
- **3** A reaction _______ of a chemical reaction is defined as an elementary reaction, constituting one of the stages of a reaction in which a reaction intermediate is converted into the next reaction intermediate in the sequence between reactants and products.
- 4 The _______ of reaction with respect to a certain reactant is defined, in chemical kinetics, as the power to which its concentration term in the rate equation is raised.
- 7 ______ state theory is a conception of chemical reactions or other processes involving rearrangement of matter as proceeding through a continuous change in the relative positions and potential energies of the constituent atoms and molecules.
- 8 ______ energy, also called threshold energy, is a term defined as the energy that must be overcome in order for a chemical reaction to occur.
- 9 A reaction ______ is an abstract one-dimensional trajectory representing progress along a reaction pathway.
- 10 The rate ______ or rate equation for a chemical reaction is an equation which links the reaction rate with concentrations or pressures of reactants and constant parameters.
- 12 The rate-_____ step is a chemistry term for the slowest step in a chemical reaction.
- 15 The reaction ______ for a reactant or product in a particular reaction tells you how fast a reaction takes place.

Answer key - pg 94 Water



ACROSS

1	A bond is a special type of dipole-dipole bond that exists between an electronegative atom and a hydrogen atom
	bonded to another electronegative atom.
3	(hypohydration) is the removal of water from a substance.
6	is any product of the condensation of atmospheric water vapor that is deposited on the earth's surface.
7	refers to the physical property of a moleculethat is repelled from a mass of water
9	Water is the gas phase of water.
11	is the name given to any one of the 14 known solid phases of water.
13	The self of water is the chemical reaction in which two water molecules react to produce a hydronium and a
	hydroxide ion.
15	is the common name for the cation derived from protonation of water.
16	In physical chemistry, and in engineering, refers to vaporized water.
17	water is water which contains a higher proportion than normal of the isotope deuterium, as deuterium oxide, or as
	deuterium protium oxide.
19	is water located beneath the ground surface in soil pore spaces and in the fractures of lithologic formations.
DC	DWN
	refers to a physical property of a molecule that can transiently bond with water through hydrogen bonding.
2	The Earth's water is always in movement, and the hydrologic or water describes the continuous movement of
	water on, above, and below the surface of the Earth.
	A is a chemical compound possessing both hydrophilic and hydrophobic properties.
	is a cloud in contact with the ground.
8	is the most common name for the diatomic anion, consisting of oxygen and hydrogen atoms, usually derived
	from the dissociation of a base.
10	A is an underground layer of water-bearing permeable rock or unconsolidated materials (gravel, sand, silt, or
	clay) from which groundwater can be usefully extracted using a water well.
12	is the net movement of water across a partially permeable membrane from a region of high solvent potential to
	an area of low solvent potential, up a solute concentration gradient.
14	is a common chemical substance that is essential to all known forms of life, typically referred to in its liquid form
	or state, but also having a solid state, ice, and a gaseous state, which are commonly encountered.
18	is water in the form of droplets that appears on thin, exposed objects in the morning or evening.



- **5** A ______ solution is a solution in which the solvent is water.
- 7 _______'s law states that the vapor pressure of an ideal solution is dependent on the vapor pressure of each chemical component and the mole fraction of the component present in the solution.
- The ______-ion effect is a term used to describe the effect on a solution of two dissolved solutes that contain the same ion.
- 9 ______ is the natural or artificial process of formation of solid crystals from a uniform solution.
- 10 The term ______ refers to a solution that contains more of the dissolved material than could be dissolved by the solvent under normal circumstances.
- 13 _____'s law states that at a constant temperature, the amount of a given gas dissolved in a given type and volume of liquid is directly proportional to the partial pressure of that gas in equilibrium with that liquid.
- 14 _____'s law states that the total pressure exerted by a gaseous mixture is equal to the sum of the partial pressures of each individual component in a gas
- 15 _____ equilibrium is any chemical equilibrium between solid and dissolved states of a compound at saturation.
- **18** Boiling-point _____ is a colligative property that states that a solution will

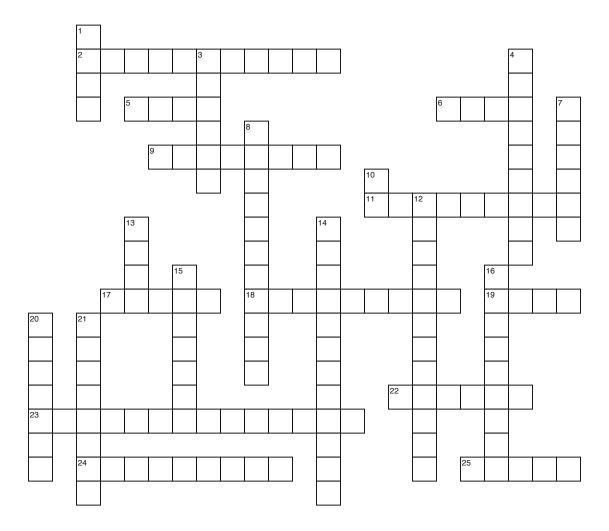
- have a higher boiling point than that of a pure solvent after the addition of a dissolved solute.
- 19 A ______, emulsion or dispersion is a type of heterogeneous mixture consisting of a dispersed phase made of tiny particles or droplets distributed evenly throughout a continuous phase.
- 21 Freezing-point ______ is the difference between the freezing points of a pure solvent and a solution mixed with a solute.
- 22 _____ is the measure of how much of a given substance there is mixed with another substance.
- 23 _____ occurs when carbon dioxide is dissolved in water or an aqueous solution.
- 24 The mole ______ of a component in a mixture is the relative proportion of molecules belonging to the component to those in the mixture, by number of molecules.
- 25 ______ is a general process in which ionic compounds separate or split into smaller molecules, ions, or radicals, usually in a reversible manner.
- 26 In chemistry, a ______ is a substance made by combining two or more different materials in such a way that no chemical reaction occurs.

DOWN

1 _____ or dissolution is the process of attraction and association of molecules of

- a solvent with molecules or ions of a solute.
- 2 A _____ solution is one that contains one mole of solute per liter.
- 3 A ______ is a substance containing free ions that behaves as an electrically conductive medium.
- 4 A _______, dispersion or colloid is a mixture of two immiscible substances in which a dispersed phase made of tiny particles or droplets is distributed evenly throughout a continuous phase.
- 6 _____ is a physical property referring to the ability for a given substance, the solute, to dissolve in a solvent.
- 10 A solvation ______ is a structure of any chemical species acting as a solvent, surrounding a solute species.
- 11 _____solution is a form of concentration expression often preferred to molarity within the biological sciences in which a 1 percent solution would have 1 g of solute dissolved in a final volume of 100 ml of solution.
- 12 _____ is the formation of a solid in a solution during a chemical reaction.
- 16 _____ properties are properties of solutions that depend on the number of particles in a given volume of solvent and not on the mass of the particles.
- 17 _____ water is plain water into which carbon dioxide gas has been dissolved.
- 20 A ______ is a homogeneous mixture composed of two or more substances.

Acids & Bases

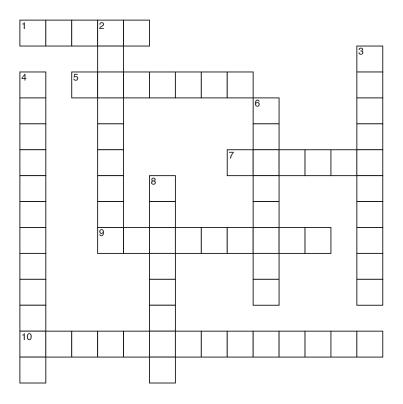


ACROSS

- 2 The ______ point or stoichiometric point occurs during a chemical titration when the amount of titrant added is equivalent, or equal, to the amount of analyte present in the sample.
- 5 A _____ acid is an acid that does not ionize in solution to a significant extent.
- 6 A _______ is traditionally considered any chemical compound that, when dissolved in water, gives a solution with a pH less than 7.0.
- 9 Hydrogen _____ has the formula HCl.
- 11 Sodium _____, also known as lye, caustic soda and sodium hydrate, is a caustic metallic base.
- 17 A ______ acid can accept a pair of electrons and form a coordinate covalent bond.
- 18 An acid-base ______ is a volumetric method in chemistry that allows quantitative analysis of the concentration of an unknown acid or base solution, making use of the neutralization reaction that occurs between acids and bases.
- **19** A ______-base reaction is a chemical reaction that occurs between a proton donor and a proton receiver.
- 22 A ______ acid is an acid that dissociates completely in an aqueous solution.
- 23 ______ is a chemical reaction, also called a water forming reaction, in which an acid and a base react and produce a salt and water.
- 24 A pH ______ is a halochromic chemical compound that is added in small amounts to a solution so that the pH of the solution can be determined easily.
- 25 A ______ base is any molecule or ion that can form a new coordinate covalent bond, by donating a pair of electrons.

- A ______ base is a chemical base that does not ionize fully in an aqueous solution
- 3 A ______ is a basic, ionic salt of a group I or group II element.
- 4 _____ is the common name for the cation derived from protonation of water. It is the simplest type of an oxonium ion.
- 7 _____ solutions are solutions that resist change in Hydronium ion and the hydroxide ion concentration (and consequently pH) upon addition of small amounts of acid or base, or upon dilution.
- 8 _____ is the addition of a hydrogen ion to an atom, molecule,
- 10 _____ is a measure of the acidity or alkalinity of a solution.
- 12 An acid _____ constant is an equilibrium constant for the deprotonation of a weak acid.
- **13** A ______ is most commonly thought of as a substance that can accept protons.
- 4 _____ acid is the aqueous solution of hydrogen chloride gas.
- 15 A ______ acid is an acid derived from inorganic substances by chemical reaction as opposed to organic acids.
- 16 Sodium _____ is a sodium salt of carbonic acid.
- 20 ______ is a compound consisting of one nitrogen atom singly bound to three hydrogen atoms, normally encountered as a gas with a characteristic pungent odor.
- 21 _____ acid is a strong mineral acid once known as oil of vitriol. Each molecule contains a sulfur atom, two hydrogen atoms and four oxygen atoms.

Coordination Chemistry



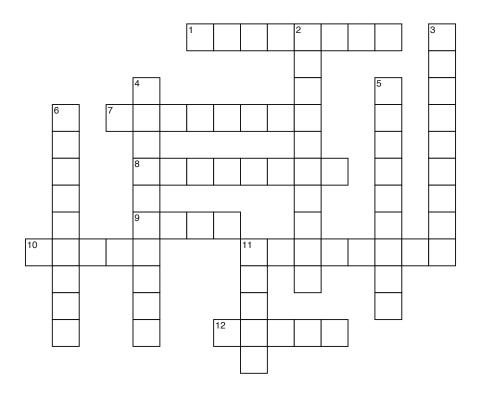
ACROSS

A ______ base is any molecule or ion that can form a new coordinate covalent bond, by donating a pair of electrons.
 A _____ is a molecule or ensemble formed by the combination of ligands and metal ions.
 A _____ is an atom, ion, or molecule donating one or more of its electrons through a coordinate covalent bond to one or more central atoms or ions
 ____ is the binding or complexation of a bi- or multidentate ligand.
 ____ chemistry is the study of chemical compounds containing bonds between carbon and a metal.

DOWN

2 The field of _____ chemistry covers all chemical compounds except the myriad organic compounds, which are the subjects of organic chemistry. 3 A _____ covalent bond is a type of covalent bonding between two atoms in which both electrons shared in the bond come from the same atom. _____ number is defined as the total number of neighbors of a central atom in a chemical compound. _____ ligand is an atom or a polyatomic entity that connects two or more metal centres in a complex. 8 Arising from the size of the valance shell of a transition metal, the 18-____ rule is a rule of thumb used primarily in transition metal chemistry for characterizing and predicting the stability of metal complexes.

Oxidation-Reduction



ACROSS

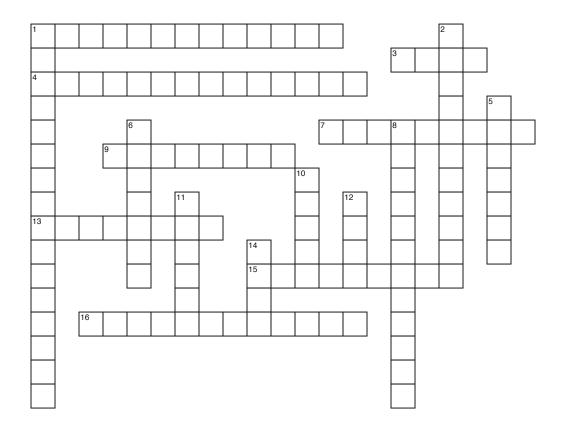
1 Electron _____ is the process by which an electron moves from one atom or molecule to another atom or molecule. **7** A _____ is a compound containing an oxygen-oxygen single bond. 8 A terminal electron _____ is a compound that receives or accepts an electron during cellular respiration or photosynthesis. **9** A _____ reaction is either the oxidation or reduction reaction component of a redox reaction. 10 The oxidation _____ is an indicator of the degree of oxidation of an atom in a chemical compound. It is usually numerically equal to the oxidation number. 11 A _____ agent is the element or a compound that reduces another species in a redox reaction. It is the electron donor in the redox. 12 In general terms, an electron ____

gives up an electron during cellular

respiration.

DOWN

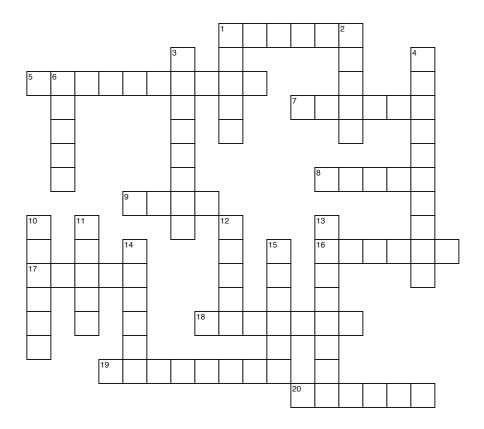
2 _____ is the free radical product of the one-electron reduction of dioxygen. **3** A _____ agent is a substance that gains electrons in a redox chemical reaction. 4 _____ is a bimolecular process involving the exchange of bonds between the two reacting chemical species, which results in the creation of products with similar or identical bonding affiliations. 5 Within a molecule or complex, the number of an element is the charge that it would have if the compound were composed of ions, with assignment of shared electrons based on electronegativity. 6 Standard reduction _____ is the tendency of a chemical species to acquire electrons and thereby be reduced. **11** _____, shorthand for reduction/oxidation reaction, is a term used to describe chemical reactions in which atoms have their oxidation state changed.



- force or potential of a body is the work done in joules to bring a unit electric charge from infinity to the body.
- 3 A _____ cell is a structure that contains a conductive electrode and a surrounding conductive electrolyte separated by a naturally-occurring Helmholtz double layer.
- 4 ______ is the process of using electrical current to coat an electrically conductive object with a relatively thin layer of metal.
- 7 The standard electrode ______ is the measure of individual voltage of any electrode at standard ambient conditions, which is at a temperature of 298K, solutes at a concentration of 1 M, and gases at a pressure of 1 bar.
- **9** A _____ cell (or voltaic cell) consists of two different metals connected by a salt bridge or a porous disk between the individual half-cells.
- 13 The standard ______ electrode is a redox electrode which forms the basis of the thermodynamic scale of oxidation-reduction potentials.
- **15** A _____ is an electrical conductor used to make contact with a nonmetallic part of a circuit.
- 16 ______ is a method of separating chemically bonded elements and compounds by passing an electric current through them.

- 1 ______ is a branch of chemistry that studies the reactions which take place at the interface of an electronic conductor and an ionic conductor.
- **2** A _____ is a substance containing free ions that behaves as an electrically conductive medium.
- 5 The _____ constant is the amount of electric charge in one mole of electrons.
- **6** A ______ is an electrode through which the positive direction of electric current flows out of a polarized electrical device.
- **8** _____ cells are composed of a vessel used to perform electrolysis and a cathode and anode.
- **10** A ______ is an electrode through which the positive direction of electric current flows into a polarized electrical device.
- 11 The ______ equation gives the electrode potential relative to the standard electrode potential of the electrode couple as a function of component concentrations.
- **12** A _____ bridge is a laboratory device used to connect the oxidation and reduction half-cells of a galvanic cell.
- 14 An electrochemical ______ is a device used for creating an electromotive force and current from chemical reactions.

Organic Functional Groups

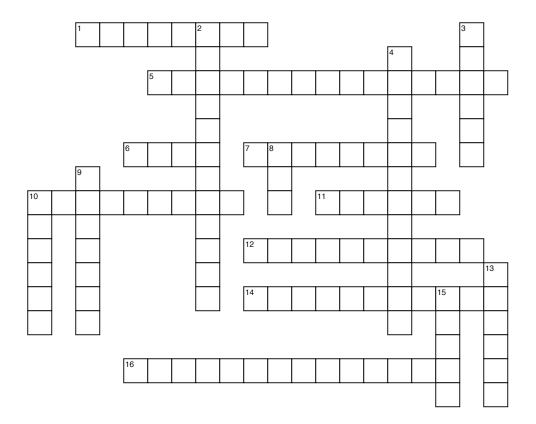


ACROSS

- 1 A _______, olefin, or olefine is an unsaturated chemical compound containing at least one carbon-to-carbon double bond.
 5 _______ acids are organic acids characterized by the presence of a carboxyl group.
 7 The ______ group consists of six carbon atoms are arranged in a cyclic ring structure. A member of the aromatic family, this functional group is highly stable.
- 3 A ______ is a compound that contains the functional group composed of a sulfur atom and a hydrogen atom. Often referred to as mercaptans, these compounds are the sulfur analogues of alcohols.
- 9 A ______ halide (also known as an acid halide) is a chemical compound derived from an acid by replacing a hydroxyl group with a halide group.
- 16 Traditionally known as acetylenes or the acetylene series, the _____s are hydrocarbons that have at least one triple bond between two carbon atoms.
- 17 ____s are a class of chemical compounds and functional groups consisting of an inorganic or organic acid in which at least one hydroxyl group is replaced by an alkoxy group.
- 18 A ______ is any organic compound in which a hydroxyl group (-OH) is bound to a carbon atom of an alkyl or substituted alkyl group
- 19 A _____ is an organic compound containing a terminal carbonyl group.
- 20 _____s, also known as Paraffins, are chemical compounds that consist only of the elements carbon and hydrogen linked together exclusively by single bonds.

- 1 _____s are organic compounds and a type of functional group that contain nitrogen as the key atom, structurally resembling ammonia, but one or more hydrogen atoms is replaced by alkyl and aryl groups.
- 2 ______ is the general name for a class of chemical compounds which contain an oxygen atom connected to two alkyl or aryl groups.
- 3 _____ in chemistry stands for a molecule consisting of an oxygen atom and a hydrogen atom connected by a covalent bond. When the oxygen atom is linked to a larger molecule it is a functional group.
- 4 The _____s (also known as halogenoalkanes or alkyl halides) are a group of chemical compounds, consisting of alkanes, such as methane or ethane, with one or more halogens linked, such as chlorine or fluorine, making them a type of organic halide.
- 6 A ______ is an organic functional group characterized by a carbonyl group linked to a nitrogen atom.
- 10 _____ (ethanoyl), is the term for a functional group which is the acyl of acetic acid.
- 11 A carbonate ______ is a functional group in organic chemistry consisting of a carbonyl group flanked by two alkoxy groups.
- is a four-carbon alkyl substituent derived from either of the two isomers of the alkane called butane.
- 13 A ______ group is a functional group composed of a carbon atom double-bonded to an oxygen atom.
- 14 _____ is the substituent form of the alkane propane.
- 15 A ______ is either the functional group characterized by a carbonyl group linked to two other carbon atoms or a chemical compound that contains this functional group.

Stereochemistry

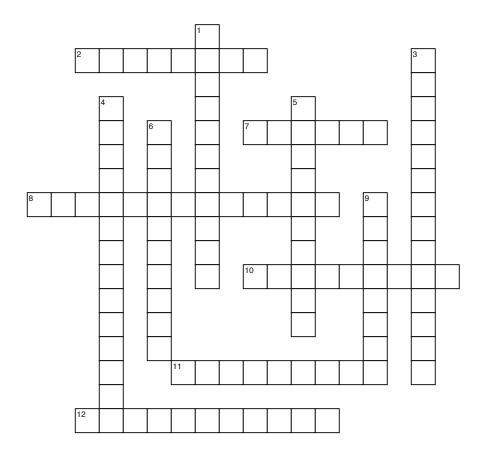


ACROSS

- _ conformation is a chemical conformation that exists in any open chain single chemical bond connecting two sp3 hybridised atoms as a conformational energy maximum.
- _____ involves the study of the relative spatial arrangement of atoms within molecules.
- compound is a chemical compound with molecules that contain 2 or more stereocenters but which is optically achiral because it contains an internal plane of symmetry.
- 7 Optical rotation or optical ______ is the rotation of linearly polarized light as it travels through certain materials.
- 10 A _____ conformation is a chemical conformation that exists in any open chain single chemical bond connecting two sp3 hybridised atoms as a conformational energy minimum.
- 11 The term _____ is used to describe an object that is non-superimposable on its mirror image.
- 12 ____s are stereoisomers that are nonsuperimposable complete mirror images of each other.
- 14 ______ is a cycloalkane containing 6 carbons and 12 hydrogens, which has the lowest angle and torsional strain of all the cycloalkanes.
- _____ isomerism is a form of stereoisomerism involving molecules with the same structural formula existing as different conformers due to atoms rotating about a bond.

- __ is any atom in a molecule bearing groups such that an interchanging of any two groups leads to a stereoisomer.
- 3 Van der Waals results from van der Waals repulsion when two substituents in a molecule approach each other with a distance less than the sum of their van der Waals radii.
- 4 _____s are stereoisomers that are not enantiomers.
 8 _____trans isomerism is a form of stereoisomerism describing the orientation of functional groups typically around double bonds which cannot rotate.
- 9 A _____ mixture is one that has equal amounts of left- and right-handed enantiomers of a chiral molecule.
- 10 _____ effects arise from the fact that if atoms are brought too close together, there is an associated cost in energy due to overlapping electron clouds.
- 13 A ______ projection visualizes chemical conformations of a carbon-carbon chemical bond from front to back, with the front carbon represented by a dot and the back carbon as a circle.
- 15 The presence of ______ strain in a molecule indicates that in a specific chemical conformation bond angles are deviating from the ideal bond angles required to achieve maximum bond strength.

Physical Properties



ACROSS

- 2 ______ organic compounds (VOCs) are organic chemical compounds that have high enough vapour pressures under normal conditions to significantly vaporize and enter the atmosphere.
- 7 A ______ solvent is a solvent that carries a hydrogen bond between an oxygen as in a hydroxyl group or a nitrogen as in an amine group.
- 8 ______ refers to the ability of a chemical compound to dissolve in fats, oils, lipids, and non-polar solvents such as hexane or toluene.
- 10 _____ compounds are organic compounds in which carbon atoms are joined together in straight or branched chains or in rings, that can be either saturated or unsaturated, but not aromatic.
- 11 A ______ or distribution coefficient is the ratio of concentrations of a compound in the two phases of a mixture of two immiscible solvents at equilibrium.
- **12** A _____ is an organic compound consisting entirely of hydrogen and carbon.

DOWN

the property of liquids to mix, forming a homogeneous solution.

3 ______ refers to the physical property of a molecule that is repelled from a mass of water

4 _____ refers to a physical property of a molecule that can transiently bond with water through hydrogen bonding.

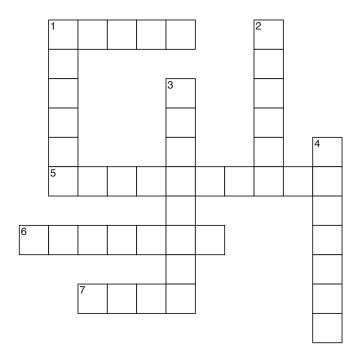
5 _____ is a measure of the speed at which a substance turns into a vapor from a solid or liquid state.

6 _____ is a term describing a chemical compound possessing both hydrophilic and hydrophobic properties.

is a common name for the group of

1 _____ is a term in chemistry that refers to

alkane hydrocarbons.



s are organic compounds and a type of functional group that contain nitrogen as the key atom.
Structurally they resemble ammonia, wherein one or more hydrogen atoms are replaced by alkyl and aryl groups.
acids are organic acids characterized by the presence of the compound of the presence o

characterized by the presence of a carboxyl group.

6 _____ acid is a colorless crystalline solid and the simplest aromatic carboxylic acid.

7 Amines and nitrogen-containing heterocyclic compounds are organic _____s.

DOWN

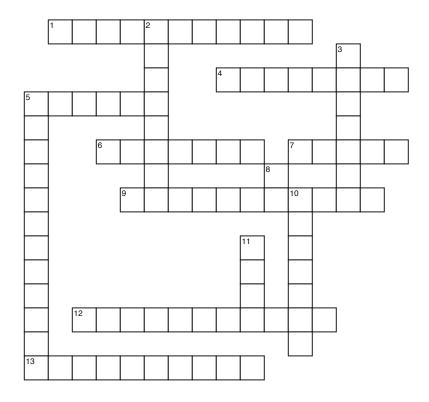
1 _____ acid, also known as ethanoic acid, is an organic chemical compound best recognized for giving vinegar its sour taste and pungent smell.

2 ______, also known under an older name of carbolic acid, possesses a structure consisting of a hydroxyl group bonded to a phenyl ring.

3 A ______ is the conjugate base of an alcohol and therefore consists of an organic group bonded to a negatively charged oxygen atom.

4 A ______, or ethanoate, is a salt or ester of acetic acid.

Aromatic Compounds

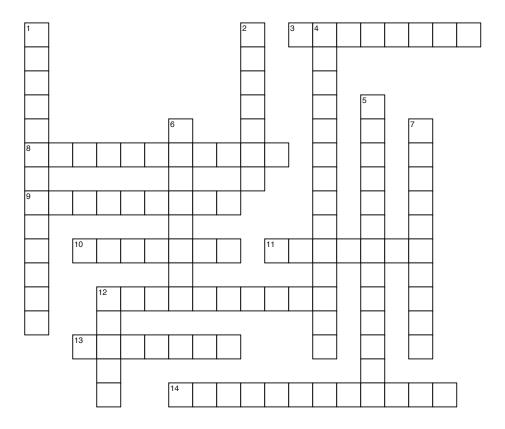


ACROSS

- 1 _____ electrons are electrons in a molecule that are not associated with a single atom or to a covalent bond.
- 4 _____ in supramolecular chemistry refers to a stacked arrangement of aromatic molecules, which interact through aromatic interactions.
- 5 ______'s rule estimates whether a planar ring molecule will have aromatic properties. It was first expressed succinctly as the 4n+2 rule by von Doering in 1951.
- 6 ______ is an organic aromatic chemical compound whose molecules contain six carbons and six hydrogens.
- 7 ____s are hydrocarbons which contain two double bonds.
- 9 ______ is a chemical property in which a conjugated ring of unsaturated bonds, lone pairs, or empty orbitals exhibit a stabilization stronger than would be expected by the stabilization of conjugation alone.
- 12 _____ is a crystalline, aromatic, white, solid hydrocarbon, best known as the primary ingredient of mothballs.
- 13 A chemically _____ system is a system of atoms covalently bonded with alternating single and multiple bonds in a molecule of an organic compound.

- 2 A set of points in space is ______ if the points all lie in the same geometric plane.3 Often compounds with extended conjugated
- systems, a ______ is a material that changes the color of light it reflects as the result of selective color absorption.
- 5 _____ compounds are organic compounds that contain a ring structure containing atoms in addition to carbon, such as sulfur, oxygen or nitrogen, as part of the ring.
- 8 ______ bonds are covalent chemical bonds where two lobes of one involved electron orbital overlap two lobes of the other involved electron orbital. Only one of the orbital's nodal planes passes through both of the involved nuclei.
- 10 An aromatic ring _____ is an effect observed in aromatic molecules if a magnetic field is directed perpendicular to the plane of the aromatic system.
- 11 In the context of organic molecules, _____ refers to any functional group or substituent derived from a simple aromatic ring.

Organic Oxidation Reduction

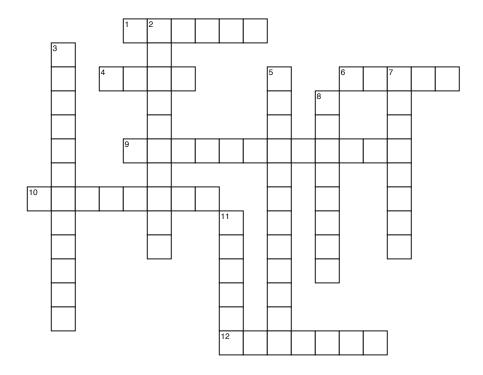


ACROSS

3	An important group of oxidizing agents, thes are salts of chromic acid.
8	Prepared by the reaction of sodium hydride on trimethylborate, sodium is a reducing agent often used to reduce
	aldehydes and ketones into alcohols.
9	The reverse of oxidative addition is elimination.
10	is the name given to the negative ion of hydrogen. Although this ion does not exist except in extraordinary
	conditions, the term is widely applied to describe compounds of hydrogen with other elements.
11	' reagent is usually ammoniacal silver nitrate, an oxidizing agent, which is itself reduced to silver metal. It is used
	as a test for aldehydes.
12	is the cleavage of an alkene or alkyne with ozone to form compounds in which the multiple carbon-carbon bond
	has been replaced by a double bond to oxygen.
13	Used for the hydrogenation of alkynes to alkenes, catalyst is a heterogeneous catalyst that consists of palladium
	deposited on calcium carbonate and treated with various forms of lead.
14	Potassium contains manganese bonded to four oxygens. Often employed for laboratory redox, it is a strong
	oxidizer with manganese in the +7 oxidation state.
DC	DWN
1	Employing borane in tetrahydrofuran in the first step, theoxidation reaction is a two-step organic chemical
_	reaction that converts an alkene into a neutral alcohol by the net addition of water across the double bond.
	Lithium aluminium is a powerful reducing agent used in organic chemistry.
4	is a catalytic chemical reaction whereby a molecule of hydrogen is added over a carbon-heteroatom single bond
_	effectively causing a lysis of the bond.
5	is a class of chemical reactions which result in an addition of hydrogen molecule usually to unsaturated organic
_	compounds.
6	Oxidative is an important classes of reactions in organometallic chemistry in which a metal complex with vacant
_	coordination sites and a relatively low oxidation state is oxidized by the insertion into a covalent bond.
1	or burning is a complex sequence of exothermic chemical reactions between a fuel and an oxidant accompanied
٠.	by the production of heat or both heat and light in the form of either a glow or flames.
12	Manganese is the chemical compound in which manganese is bound to two oxygens. It is often used to oxidize

allylic alcohols to the corresponding aldehydes.

Molecular Spectroscopy

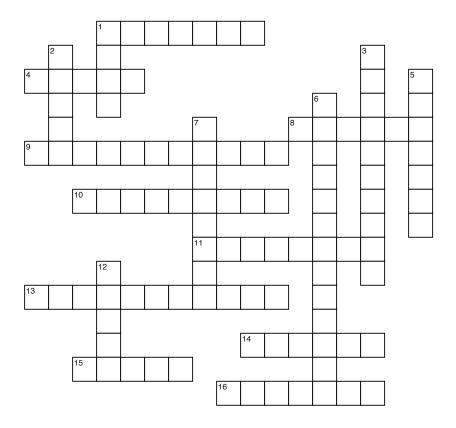


ACROSS

- precession refers to the precession of the magnetic moments of electrons, atomic nuclei, and atoms about an external magnetic field.
- 4 _____ spectrometry is an analytical technique used to measure the mass-tocharge ratio of ions.
- 6 In nuclear magnetic resonance (NMR), the chemical ______ describes the dependence of nuclear magnetic energy levels on the electronic environment in a molecule.
- **9** A ______ is an optical instrument used to measure properties of light over a specific portion of the electromagnetic spectrum, typically used in spectroscopic analysis to identify materials.
- 10 J-_____ describes the interaction between two nuclear spins due to the influence of bonding electrons on the magnetic field running between the two nuclei.
- 12 _____ magnetic resonance spectroscopy exploits the magnetic properties of certain nuclei. The most important applications for the organic chemist are proton NMR and carbon-13 NMR spectroscopy.

- 2 A material's ______ spectrum shows the fraction of incident electromagnetic radiation absorbed by the material over a range of frequencies.
- is the study of the interaction between radiation and matter.
- **5** The _____ ratio of a particle or system is the ratio of its magnetic dipole moment to its angular momentum.
- 7 ______ spectroscopy is a form of absorption spectroscopy that develops information about the structure of molecular substances from various covalent bond vibrational modes.
- **8** A mass ______ is an intensity versus mass-to-charge ratio plot which represents the distribution of components by mass-to-charge ratio in a sample.
- 11 _____ NMR is the application of nuclear magnetic resonance in NMR spectroscopy with respect to hydrogen.

Aldehydes & Ketones



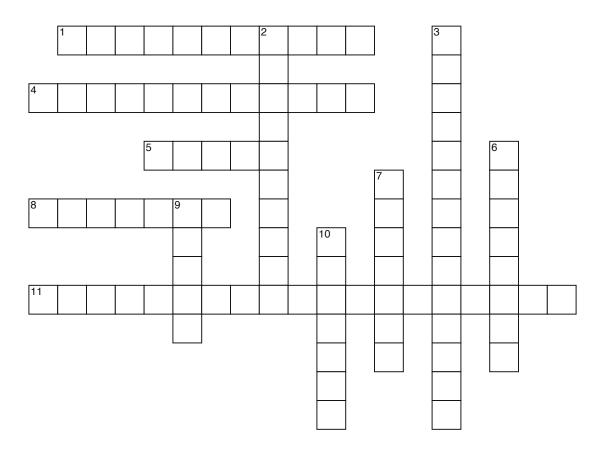
ACROSS

1	A anion is the anion of an enol, formed
-	by loss of a proton from the alpha carbon of a
	carbonyl group.
4	A adduct is a beta-hydroxy ketone or
	aldehyde resulting from the addition of a ketone
	enolate to an aldehyde.
8	A is a molecule with two single bonded
	oxygens are attached to the same carbon atom
	which is also bonded to an alkyl or aryl group and a
	hydrogen.
9	Keto-enol refers to a chemical
	equilibrium between a keto form and an enol.
10	Nucleophilic addition occurs by reaction
	of a nucleophile at the beta position of alpha-beta
	unsaturated carbonyl compounds.
11	In a reaction two or more molecules
	combine to form a larger one.
13	The reaction is an organic reaction by
	an aldehyde or ketone with a cyanide anion or a
	nitrile to form a cyanohydrin.
14	A base is a functional group that
	contains a carbon-nitrogen double bond with the
	nitrogen atom connected to an aryl or alkyl group,
	but not hydrogen.
15	A is a molecule with two single bonded
	oxygens attached to the same carbon atom which
	is also bonded to two alkyl or aryl groups.
16	The Wolff reduction is a chemical

reaction that fully reduces an aldehyde or ketone to an alkane.

1	s are alkenes with a hydroxyl group
	affixed to one of the carbon atoms composing the
	double bond.
2	The carbon in an aldehyde or ketone
	refers to the first carbon after the carbonyl carbon.
3	A group is introduced into a molecule by
·	chemical modification of a functional group in orde
	to obtain chemoselectivity in a subsequent
	chemical reaction.
5	' reagent is usually ammoniacal silver
3	-
	nitrate, an oxidizing agent, which is itself reduced
_	to silver metal. It is used as a test for aldehydes.
6	is an organic reaction that involves the
	formation of an acetal or ketal.
7	The reaction is an organometallic
	chemical reaction involving alkyl- or aryl-
	magnesium halides with electrophiles.
12	A is an unsaturated chemical compound
	or functional group consisting of a conjugated
	system of an alkene and a ketone.

Alcohols



ACROSS

reaction that involves the loss of water from the reacting molecule.

4 ______ substitution is a fundamental class of substitution reaction in which an electron rich nucleophile selectively bonds with or attacks the positive charge of a group or atom called the leaving group.

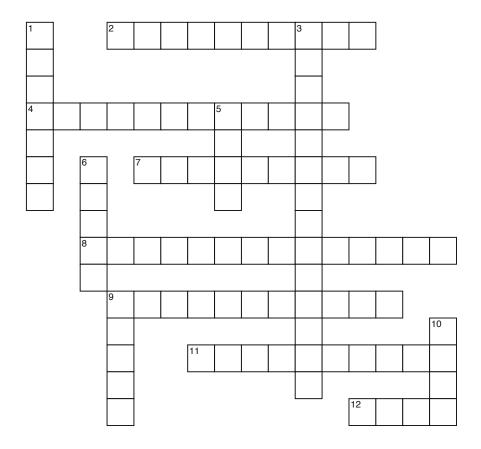
5 ______ is the general name for a class of chemical compounds which contain an oxygen atom connected to two alkyl or aryl groups.

1 A reaction is a chemical

- 8 ______ esterification is the process of forming an ester by refluxing a carboxylic acid and an alcohol in the presence of an acid catalyst.
- 11 ______ is the process of exchanging the alkoxy group of an ester compound by another alcohol.

- 2 Phosphorus _____ is widely used in the laboratory for the conversion of alcohols to alkyl bromides.
- 3 _____ is the general name for a chemical reaction in which two chemicals, such as an alcohol and an acid, form an ester as the reaction product.
- **6** A _____ is the conjugate base of an alcohol.
- **7** A _____ is a cyclic ether with only three ring atoms.
- **9** The Williamson ______ synthesis involves the reaction of an alkoxide ion with a primary alkyl halide via an SN2 reaction.
- 10 _____ chloride is an inorganic compound often used in chlorination reactions in which sulfur is bound to an oxygen and two chlorine atoms.

Carboxylic Acid Derivatives



ACROSS

by reaction with water.

4 A ______ reaction is a chemical reaction in which two molecules or moieties combine to form one single molecule, together with the loss of a small molecule.

7 An acid _____ is an organic compound which has two acyl groups bound to the same oxygen atom.

8 ____ is the general name for a chemical reaction in which two chemicals, such as an alcohol and an acid, form an ester as the reaction product.

9 ___ or ethanoylation describes a

2 _____ is a chemical reaction or process in

which a chemical compound is broken down

11 _____ acids are organic acids characterized by the presence of a carboxyl group.

reaction that introduces an acetyl functional

12 A _____ group is a functional group derived by the removal of one or more hydroxyl group from an oxoacid.

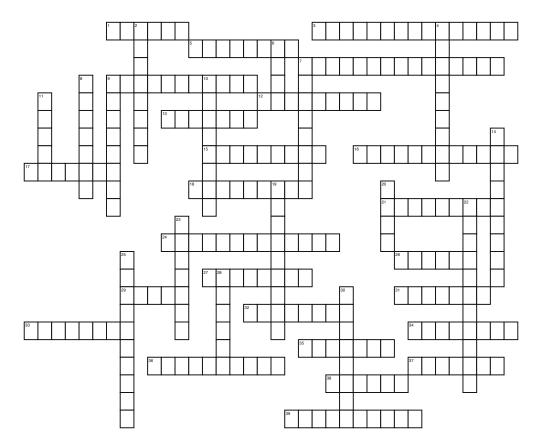
group into an organic compound.

DOWN

1 _____ esterification is the process of forming an ester by refluxing a carboxylic acid and an alcohol in the presence of an acid catalyst. is the hydrolysis of an ester under basic conditions to form an alcohol and the salt of a carboxylic acid. 5 Nucleophilic _____ substitution describes the substitution reaction involving nucleophiles and carboxylic acid derivatives including esters, amides and acid halides. 6 ____s are a class of chemical compounds and functional groups which consist of an inorganic or organic acid in which at least one hydroxyl group has been replaced by an alkoxy group. s are a class of chemical compounds and functional groups in which a carbonyl group carbon is linked to a nitrogen or acid halide is a chemical 10 A compound derived from an acid by replacing

a hydroxyl group with a halide group.

Proteins



ACROSS

- 1 The _____ site of an enzyme contains the catalytic and binding sites.
- 3 ______inhibition is a type of inhibition that reduces the maximum rate of a chemical reaction without changing the apparent binding affinity of the catalyst for the substrate.
- 5 A _______ is an enzyme capable of cleaving the phosphodiester bonds between the nucleotide subunits of nucleic acids.
- 7 _____ is the addition of a phosphate group to a protein molecule or a small molecule.
- 9 _____inhibition is a form of enzyme inhibition where binding of the inhibitor to the enzyme prevents binding of the substrate and vice versa.
- 12 The ______ structure of a protein is the general three-dimensional form of local segments of biopolymers such as proteins and nucleic acids.
- 13 _____ proteins, also called scleroproteins, are long filamentous protein molecules that form one of the two main classes of tertiary structure protein, the other being globular proteins.
- 15 Enzyme _____s are molecules that bind to enzymes and decrease their activity.
- 16 ____s are proteins that contain oligosaccharide chains covalently attached to their polypeptide backbones.
- 17 Protein ______ is the physical process by which a polypeptide arranges into its characteristic threedimensional structure.
- **18** A ______ is a molecule upon which an enzyme acts.
- 21 A _____ amino acid or indispensable amino acid is an amino acid that cannot be synthesized de novo, and therefore must be supplied in the diet
- 24 Post____ modification is the chemical modification of a protein after its initial synthesis on a ribosome. It is one of the later steps in protein biosynthesis for many proteins.
- 26 ____s are proteins that catalyze chemical reactions.
- 27 The ______ structure of a protein or any other macromolecule is its three-dimensional structure, as defined by the atomic coordinates.

- 29 The ______helix is a common motif in the secondary structure of proteins, a right-handed coiled conformation, resembling a spring, in which every backbone N-H group donates a hydrogen bond to the backbone carbonyl group of the amino acid four residues earlier.
- **31** A protein ______ is a group of two or more associated proteins formed by protein-protein interaction that is stable over time.
- 32 _____ proteins, or spheroproteins are one of the two main protein classes, comprising globelike proteins that are more or less soluble in aqueous solutions.
- 33 _____ feedback feeds part of a system's output, inverted, into the system's input; generally with the result that fluctuations are attenuated.
- 34 Enzyme ______ is the study of the rates of chemical reactions that are catalysed by enzymes.
- 35 A ______ bond is a chemical bond that is formed between two amino acids when the carboxyl group of one molecule reacts with the amino group of the other molecule, releasing a

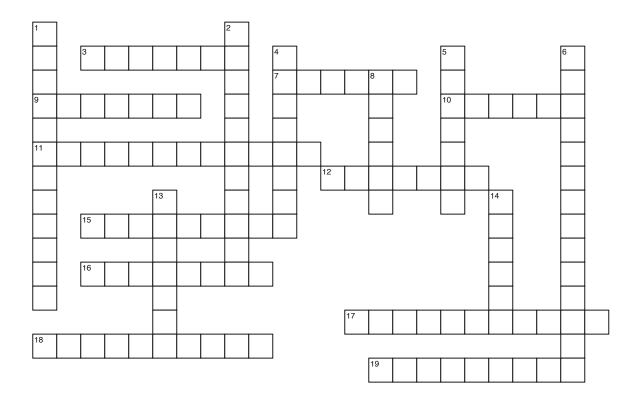
molecule of water.

- 36 A ______ is a chemical compound that is electrically neutral but carries formal positive and negative charges on different atoms.
- 37 ____s are short polymers formed from the linking, in a defined order, of alpha-amino acids.
- **38** The _____ state of a protein is its operative or functional form.
- 39 _____ is the large-scale study of proteins.

- The N- or amino-_____ refers to the end of a protein or polypeptide terminated by an amino acid with a free amine group.
- 4 A membrane _____ protein is a protein involved in the movement of ions, small molecules, or macromolecules, such as another protein across a biological membrane.
- 6 A ______ chain in organic chemistry and biochemistry is a part of a molecule that is attached to a core structure.
- 7 The term _____ refers all the expressed proteins in an organism at a given time point under

- defined conditions.
- s are large organic compounds made of amino acids arranged in a linear chain and joined together by peptide bonds between the carboxyl and amino groups of adjacent amino acid residues.
- g _____ is the main protein of connective tissue in animals and the most abundant protein in mammals, making up about 25% of the total protein content.
- 10 The C- or carboxyl-_____ of a protein or polypeptide is the end of the amino acid chain terminated by a free carboxyl group.
- 11 A _____acid is a molecule that contains both amine and carboxyl functional groups. In biochemistry, this term is often used to refer to the select group of specific forms that serve as the building blocks of proteins.
- 14 A _____ bond is a single covalent bond derived from the coupling of thiol groups.
- 19 Enzyme _____s are molecules that bind to enzymes and increase their activity.
- 20 The _____-pleated sheet is a major form of regular secondary structure in proteins along with the alpha helix. This form of secondary structure consists of strands connected laterally by three or more hydrogen bonds, forming a generally twisted, pleated sheet.
- The ______ point is the pH at which a particular molecule or surface carries no net electrical charge.
- 23 The ______ structure of a biological molecule is the exact specification of its atomic composition and the chemical bonds connecting those atoms, including stereochemistry.
- 25 The ______ structure of a protein is the arrangement of multiple folded protein molecules in a multi-subunit complex.
- 28 _____ catalysis is the catalysis of chemical reactions by proteins.
- 30 A _____ is any enzyme that conducts proteolysis, that is, begins protein catabolism by hydrolysis of the peptide bonds that link amino acids together in the polypeptide chain.

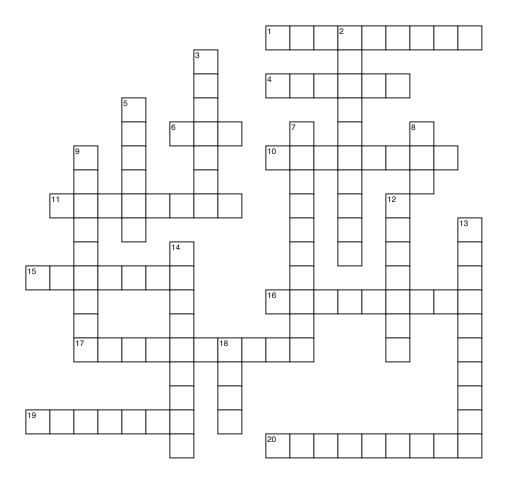
Carbohydrates



ACROSS

ACROSS			starch, the other being amylose.	
		18	A is a hexose with an aldehyde group on	
3	, or malt sugar, is a disaccharide formed		one end.	
	from two units of glucose joined with an alpha(1 -	19	Also called brain sugar,, is a type of	
	4) linkage.		simple sugar which is less sweet than glucose and	
7	is simple pentose sugar which is a		not very water-soluble.	
	component of RNA.		•	
9	is a linear polymer of glucose linked with	DC	OWN	
	mainly alpha (1 - 4) bonds which can consist of			
	several thousand glucose units. It is one of the two	1	A is a sugar composed of two	
	components of starch, the other being amylopectin.		monosaccharides.	
10	is a long-chain polymer of beta-glucose	2	is a deoxy sugar derived from the	
. •	that forms a hard, semitransparent material found	_	pentose sugar ribose by the replacement of the	
	throughout the natural world, for example, in fungi,		hydroxyl group at the 2 position with hydrogen	
	the exoskeletons of arthropods, the radula of	4	(or levulose) is a simple sugar found in	
	molluscs and the beaks of cephalopods.		many foods and is one of the three most important	
11	s or saccharides are a major class of		blood sugars along with glucose and galactose.	
•	biomolecules which are simple compounds,	5	(table sugar) is a disaccharide formed	
	aldehydes or ketones with many hydroxyl groups	·	by the condensation of glucose and fructose.	
	added, usually one on each carbon atom that is not	6	s are the simplest carbohydrates. They	
	part of the aldehyde or ketone functional group.	·	cannot be hydrolyzed into simpler sugars.	
12	is a disaccharide that consists of beta-	8	is a mixture of amylose and	
	D-galactose and beta-D-glucose monomers	·	amylopectin. These are both complex carbohydrate	
	connected by a beta-1-4 glycosidic linkage.		polymers of glucose.	
15	is a polysaccharide derived from beta-	13	, a monosaccharide, is an important	
	glucose. It is the primary structural component of		carbohydrate in biology, used by the living cell as a	
	green plants.		source of energy and metabolic intermediates. It is	
16	is a polysaccharide of glucose which		one of the main products of photosynthesis and	
	functions as the primary short term energy storage		starts cellular respiration in both prokaryotes and	
	in animal cells.		eukaryotes.	
17	is a highly branched polymer of glucose	14	A is a monosaccharide with six carbon	
• •	found in plants. It is one of the two components of	1.7	atoms, twelve hydrogens, and six oxygens.	
	round in planto. It is one of the two components of		atomo, two try arogono, and six oxygens.	

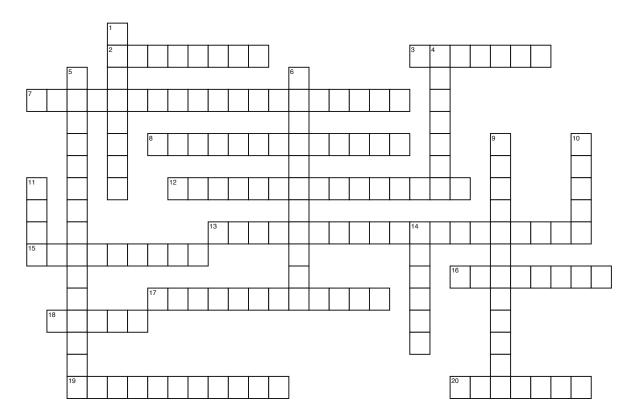
Nucleic Acids



ACROSS

triphosphate is a multifunctional nucleotide that is most important as a molecular currency of intracellular energy transfer. _____ base pairs with adenine in RNA and is replaced by thymine in DNA 6 ______ is a nucleic acid that contains the genetic instructions used in the development and functioning of all known living organisms. is one of the five main nucleobases found in the nucleic acids DNA and RNA. It is a pyrimidine derivative. In Watson-Crick base pairing, it forms three hydrogen bonds with guanine. __ RNA (abbreviated tRNA) is a small RNA chain that transfers a specific amino acid to a growing polypeptide chain at the ribosomal site of protein synthesis during translation. is one of the five main nucleobases found in the nucleic acids DNA and RNA. In base-pairing it binds to cytosine through three 15 _____RNA (rRNA), a type of RNA synthesized in the nucleolus, is the central component of the ribosome, the protein manufacturing machinery of all living cells. s are the parts of RNA and DNA that may be involved in pairing. These include cytosine, guanine, adenine, thymine, uracil, xanthine and hypoxanthine. is one of the four bases in the nucleic acid of DNA along with adenine, quanine, and cytosine. It always base-pairs with adenine. 19 is a nucleoside composed of adenine attached to a ribose moiety via a beta-N9-glycosidic bond. **DOWN** _ is a chemical compound that consists of 3 portions: a heterocyclic base, a sugar, and one or more phosphate groups. is a purine with a variety of roles in biochemistry including cellular respiration, as part of ATP, NAD, and FAD, and protein synthesis, as a chemical component of DNA and RNA is a heterocyclic aromatic organic compound, consisting of a pyrimidine ring fused to an imidazole ring. __ is a heterocyclic aromatic organic compound similar to benzene and pyridine, containing two nitrogen atoms at positions 1 and 3 of the six-member ring is a nucleic acid polymer consisting of nucleotide monomers, which plays a number of important roles in the 8 Ribonucleic acid or processes of translating genetic information from DNA into proteins. ____ is the complex of DNA and protein that makes up chromosomes s are the chief protein components of chromatin, acting as spools around which DNA winds, and playing a role in gene regulation. ____s are glycosylamines made by attaching a nucleobase to a ribose or deoxyribose ring. _____ Ribonucleic Acid (mRNA) is a molecule of RNA encoding a chemical blueprint for a protein product.

18 Two nucleotides on opposite complementary DNA or RNA strands that are connected via hydrogen bonds are called a ___



- 2 _____ is the triterpene which is the biochemical precursor to the whole family of steroids
- 3 A ______ is a terpenoid lipid characterized by a carbon skeleton with four fused rings generally arranged in a 6-6-6-5 fashion.
- 7 ______ is a phospholipid which is a major constituent of cell membranes. This lipid is such a major component of lecithin that in some contexts the terms are used as synonyms.
- 8 A ________ is any member of a group of lipid compounds having 20 carbon atoms, including a 5-carbon ring which are derived enzymatically from fatty acids, having important functions in the animal body especially in immunity.
- 12 _____ fats are fatty acids having more than

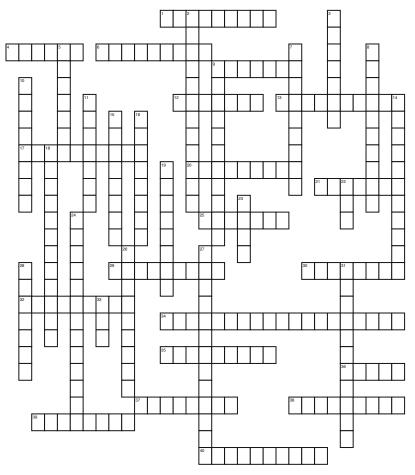
- one double bond present in the fatty acid chain.
- 13 ____s or phosphoglycerides are glycerol-based phospholipids. They are the main component of biological membranes.
- 15 A ______ fat is a fat that consists of triglycerides containing fatty acids with no double bonds between the carbon atoms of the fatty acid chain
- 16 _____ is mostly a mixture of glycolipids, triglycerides, and phospholipids. However, in biochemistry, the term is usually used as a synonym for pure phosphatidylcholine.
- 17 ____s are a class of lipids which are a major component of all biological membranes, along with glycolipids, cholesterol and proteins.
- 18 A ______ acid is a carboxylic acid often with a long unbranched aliphatic tail, which is either saturated or unsaturated

- 19 A ______, more formally known as a diacylglycerol, is a glyceride consisting of two fatty acid chains covalently bonded to a glycerol molecule through ester linkages.
- 20 _____ fatty acids are a family of polyunsaturated fatty acids which have in common a double bond in the third carbon-carbon bond from the terminal end.

- 1 ______ is a common synonym for the chemical compound 2-methylbuta-1,3-diene. It is an important biological material, being a precursor for many important classes such as the terpenes and steroids.
- 4 Including the essential oils of many plants,
 _____s are a large and varied class of hydrocarbons which are derived biosynthetically

- from units of isoprene.
 fats are fatty
 acids having a single
- double bond present in the fatty acid chain, with all of the carbons in the chain single bonded is a sterol
- found in the cell membranes of all physiological tissues and transported in the blood plasma of all animals.
- 9 A ______ consists of glycerol esterified with three fatty acids, although slightly more formally, the form is known as triacylglycerol or triacylglyceride.
- 10 ____s can be broadly defined as any fat-soluble (hydrophobic), naturally-occurring molecules.
- 11 A wide group of molecules, _____ are generally triesters of glycerol and fatty acids.
- _____s, or steroid alcohols are a subgroup of steroids with a hydroxyl group in the 3position of the A-ring.

Biological Membranes



ACROSS

- A _____solution has the lower osmotic pressure of two fluids. The term also describes a cell environment with a lower concentration of solutes than the cytoplasm of the cell.
- transport is the mediated transport of biochemicals, and other atomic/molecular substances, across a membrane which specifically requires the expenditure of cellular energy to move molecules against a gradient.
- 6 ______ is the sponteneous net movement of particles from an area of high concentration to an area of low concentration.
- 9 transport means moving biochemicals and other atomic or molecular substances across membranes in a process that does not require chemical energy.
- 12 Sodium _____s are integral membrane proteins that conduct sodium ions through a cell's plasma membrane.
- 13 A _____ is a chemical compound possessing both hydrophilic and hydrophobic properties.
- 17 A _____ cell environment has a higher concentration of solutes than inside the animal or plant cell.
- 20 Sodium/______-ATPase is an enzyme located in the plasma membrane of virtually every human cell and is common to all cellular life. It helps maintain cell potential and regulate cellular volume.

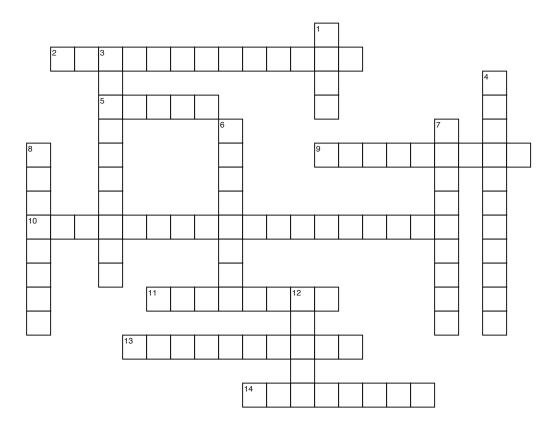
- 21 _____ active transport is directly coupled to ATP cleavage to transport molecules across a membrane.
- 25 A lipid ______ is a membrane or zone of a membrane composed of lipid molecules two molecules thick, a structure which is a critical component of all biological membranes.
- 29 In _____ active transport there is no direct coupling of ATP. Instead the energy derives from the electrochemical potential difference created by pumping ions out of the
- 30 _____ junctions are protein complexes that occur at cell-cell junctions in epithelial tissues. They are usually more basal than tight junctions.
- 32 A second system is a method of cellular signalling where the signalling molecule does not enter the cell but instead utilizes a cascade of events to transduce the signal into a change inside the cell.
- 34 _____ is a phospholipid which is a major constituent of cell membranes. It is such a major component of lecithin that in some contexts the terms are used as synonyms.
- 35 Membrane _____ is the electrical voltage across a plasmalemma.
- 36 _____ junctions, or zonula occludens, are the closely associated areas of two cells whose membranes join together forming a virtual impermeable barrier to fluid.

- 37 A ______ membrane protein is a protein molecule, or assembly of proteins, that is permanently attached to the biological membrane.
- 38 _____ channels are the most common type of ion channel, forming pores selective for that ion spanning cell membranes.
- 39 Transmembrane _____s are integral membrane proteins that bind to a signalling molecule or sometimes to a pair of such molecules on one side of the membrane and initiate a response on the other side.
- 40 _____ is the process by which a cell directs secretory vesicles to the cell membrane and releases their contents.

- 2 ____s are a class of lipids, and a major component of all biological membranes, along with glycolipids, cholesterol and proteins.
- gressure is the hydrostatic pressure produced by a solution in a space divided by a semipermeable membrane due to a differential in the concentrations of solute.
- 5 _____gated calcium channels are a group of ion channels found in excitable cells such as neurons, glial cells, muscle cells, etc.
- 7 A _____, also known as macula adherens, is a cell structure specialized for cell-to-cell adhesion.

 8 _____ membrane proteins are

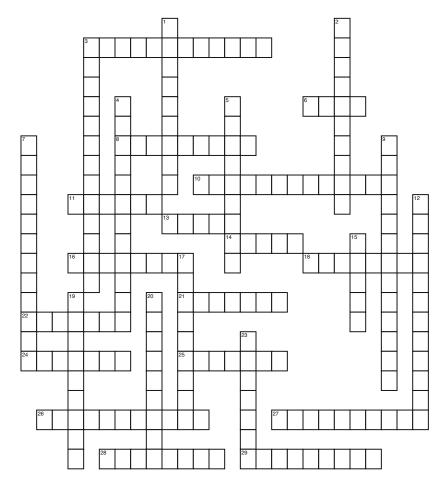
- proteins that adhere only temporarily to the biological membrane with which they are associated.
- 9 _____ is a form of endocytosis in which small particles are brought into the cell suspended within small vesicles which subsequently fuse with lysosomes.
- o _____ is a protein that is the major constituent of the 'coat' of the coated pits and coated vesicles formed during endocytosis of materials at the surface of cells.
- 11 _____-gated potassium channels are transmembrane channels specific for potassium and sensitive to changes in the cell's membrane potential.
- 14 ______ is a process whereby cells absorb material from the outside by engulfing it with their cell membrane.
- 15 A cell ______ is one of a variety of types of structures consisting of protein complexes that provide contact between neighbouring cells, between a cell and the extracellular matrix, or participate in building up the paracellular barrier of epithelia.
- 16 A ______ is a protein on the cell membrane, within the cytoplasm, or within the cell nucleus that binds to a specific ligand, such as a neurotransmitter, hormone, or other substance, and initiates a cellular response.
- 18 ______ is the cellular process of engulfing solid particles by the cell membrane to form an internal phagosome, or food vacuole.
- 19 The cell _____ is a semipermeable lipid bilayer found in all cells.
- 2 channels are poreforming proteins that help to establish and control the small voltage gradient across the plasma membrane of all living cells.
- 23 A cell _____ is a fairly rigid layer surrounding the cells of plants, bacteria, archaea, fungi, and algae, located external to the cell membrane, which provides the cell with structural support, protection, and acts as a filtering mechanism.
- 24 A _____ membrane is a membrane which will allow certain molecules or ions to pass through it by diffusion and occasionally specialized facilitated diffusion.
- 26 The _____ pathway is a series of steps a cell uses to move proteins out of the cell.
- 27 A ______ protein is a protein that spans the entire biological membrane.
- is the spontaneous net movement of water across a partially permeable membrane from a region of high solvent potential to an area of low solvent potential, up a solute concentration gradient.
- 31 Receptor-mediated ______ is a process by which cells internalize molecules into a cell by the inward budding of plasma membrane vesicles which contain proteins with receptor sites specific to the molecules being internalized.
- 33 A ______junction or nexus is a junction between certain animal celltypes that allows different molecules and ions, mostly small intracellular signaling molecules, to pass freely between cells.



_____, also known as murein, is a polymer consisting of sugars and amino acids that forms a mesh-like layer outside the plasma membrane of eubacteria. 5 The _____ membrane refers to the outside membranes of Gram-negative _ is a protein that arranges itself in a hollow cylinder to form the filament in bacterial flagellum 10 A _____ is a large molecule consisting of a lipid and a polysaccharide (carbohydrate) joined by a covalent bond. 11 The cell _____ is the cell membrane and cell wall plus an outer membrane, if one is present. Most fall into two major categories: Gram positive and Gram negative. _ is a general term referring to extracellular polymeric material produced by some bacteria, epithelia and other cells. **14** Gram- bacteria are those that retain a crystal violet dye during the Gram stain process.

1	A cell is a fairly rigid layer
	surrounding a prokaryotic cell, located
	external to the cell membrane, which provides
	the cell with structural support, protection, and
	acts as a filtering mechanism.
3	s are a group of organisms that lack
	a cell nucleus, or any other membrane-bound
	organelles.
4	The space is the space seen
	between the plasma membrane and the outer
	membrane in the gram-negative bacteria.
6	A is a small, dense, functional
	structure found in most known cells that
	assembles proteins and polypeptides used in
	cell division.
7	A is a long, slender projection from
	the cell body, composed of microtubules and
	surrounded by the plasma membrane.
8	In prokaryotes, the (also known as
	the nuclear region, nuclear body or chromatin
	body) is an irregularly shaped region where
	the genetic material is localized.
12	A is a hairlike appendage found on
	the surface of many bacteria. This term and
	fimbria are often used interchangeably.

Eukaryotes



ACROSS

- 3 The _____ system is the system of internal membranes within eukaryotic cells that divide the cell into functional and structural compartments, or organelles.
- 6 Nuclear s are large protein complexes that cross the nuclear envelope, which is the double membrane surrounding the eukaryotic cell nucleus.
- 8 _____ is a gelatinous, semitransparent fluid that fills most cells
- 10 ____s are the thinnest filaments of the cytoskeleton found in the cytoplasm of all eukaryotic cells. These linear polymers of actin subunits are flexible and relatively strong.
- 11 A _____ is an organelle found in eukaryotic cells. It is a thin, tail-like projection extending approximately 5-10 micrometers outwards from the cell body.
- 13 A ______ body is an organelle formed from a centriole, a short cylindrical array of microtubules. It is found at the base of a eukaryotic undulipodium (cilium or flagellum) and serves as a nucleation site for the growth of the axoneme microtubules.
- 14 _____ is the protein which serves as the monomeric subunit of microfilaments, one of the three major components of the cytoskeleton, and of thin filaments which are part of the contractile apparatus in muscle cells.
- 16 A comprises a flattened

- membrane disk which makes up the Golgi apparatus.
- s are organelles that contain digestive enzymes to digest excess or worn out organelles, food particles, and engulfed viruses or bacteria.
- 21 ____s are major organelles found in plants and algae responsible for photosynthesis, storage of products like starch and for the synthesis of many classes of molecules such as fatty acids and terpenes.
- 22 A _____ is one of several members of a small family of globular proteins. The most common members this family are the proteins which makes up microtubules.
- 24 The _____ is a membraneenclosed organelle found in most eukaryotic cells which contains the cell's genetic material.
- 25 The _____ is the inner cytoskeletal structure of eukaryotic cilia or flagella.
- 26 ____s are organelles found in plant cells and eukaryotic algae that conduct photosynthesis. They are generally considered to have originated as endosymbiotic cyanobacteria.
- ____s are ubiquitous organelles in eukaryotes that participate in the metabolism of fatty acids and other metabolites. They have enzymes that rid the cell of toxic peroxides.
- 28 The nuclear _____ or membrane is the double membrane

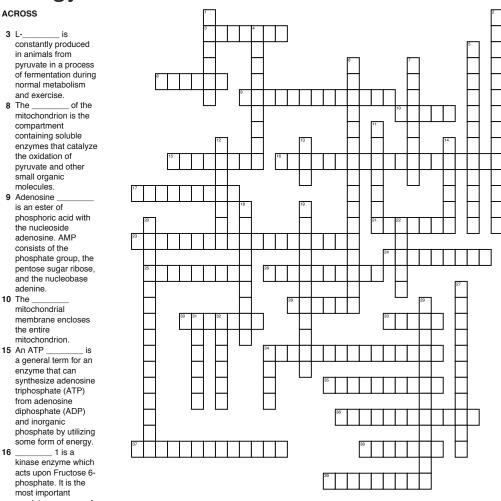
- of the nucleus that encloses genetic material in eukaryotic cells.
- 29 Animals, plants, fungi, and protists are ______s, which are organisms whose cells are organized into complex structures by internal membranes and a cytoskeleton.

- 1 A ______ is a barrel shaped organelle found in most eukaryotic cells with walls usually composed of nine triplets of microtubules.
- 2 The _____ is the main microtubule organizing center of the cell as well as a regulator of cellcycle progression.
- 3 The ____ theory concerns the origins of mitochondria and plastids, according to which these organelles originated as separate prokaryotic organisms which were taken inside the cell as endosymbionts.
- 4 A nuclear signal is an amino acid sequence which acts like a 'tag' on the exposed surface of a protein to target the protein to the cell nucleus through the nuclear pore complex and to direct it into the nucleus via its recognition by cytosolic nuclear transport receptors.
- 5 ____-tubular cluster is an intermediate compartment mediating trafficking between the endoplasmic reticulum and Golgi complex.
- 7 The _____ is a cellular scaffolding or skeleton contained

- within the cytoplasm.
- A is a membraneenclosed organelle that is found in most eukaryotic cells that. In addition to supplying cellular energy, they are involved in a range of other processes.
- 112 _______filaments are cytoskeletal structures formed by members of a family of related proteins. These filaments have a diameter between that of actin microfilaments and microtubules.
- 15 The _____ apparatus is an organelle found in most eukaryotic cells the primary function of which is to process and package the macromolecules such as proteins and lipids that are synthesized by the cell.
- 17 The spindle ______ is a structure of the eukaryotic cytoskeleton involved in mitosis and meiosis, often referred to as the mitotic spindle during mitosis and the meiotic spindle during meiosis.
- The _____ is a sub-organelle of the cell nucleus. Its main function is the production and assembly of ribosome components.
- 20 A ______ is a specialized subunit within a cell, having a specific function, and separately enclosed within its own lipid membrane.
- 23 A ______ is a relatively small and enclosed compartment, separated from the cytosol by at least one lipid bilayer.

Energy Metabolism

Answer key - pg 108



regulatory enzyme of glycolysis.

17 Glucose 6also known as Robison ester, is glucose sugar phosphorylated on carbon 6. Yhe vast majority of glucose

majority of glucose entering a cell will become phosphorylated in this

21 ____ are the internal compartments formed by the inner membrane of a mitochondrion.

wav.

23 (PEP) has the highest energy phosphate bond found in living organisms. It is formed in glycolysis by the action of the enzyme enolase on 2-phosphoglycerate.

24 A _____ organism is any organism that does not require oxygen for growth.

phosphorylation is a type of chemical reaction that results in the formation of ATP by the direct transfer of a phosphate group to ADP from a reactive intermediate.

26 ____s are generally membrane-bound hemoproteins

that contain heme groups and carry out electron transport.

28 Adenylate ______

is a phosphotransferase enzyme that catalyzes the production of ATP from ADP.

33 An electron transport
____associates
electron carriers and
mediating biochemical
reactions that produce
ATP.

34 Oxidative _____ii
a metabolic pathway
that uses energy
released by the
oxidation of nutrients
to produce ATP.

process of most carbohydrate catabolism serving the functions of producing ATP and NADH, pyruvate for the citric acid cycle, and a variety of other compounds which are important for biosynthesis.

36 Ethanol is the biological process by which sugars such as glucose, fructose, and sucrose, are converted into ethanol and carbon dioxide.

37 The _____space is the region between the inner membrane and the outer membrane of a mitochondrion or a chloroplast.

acid is an alpha-keto acid which plays an important role in biochemical processes. It is an output of glycolysis.

(also known as the Neuberg ester) is fructose sugar phosphorylated on carbon 6. The beta-Dform of this compound is very common in cells.

DOWN

1 The electron carrier,

adenine dinucleotide accomodates two equivalents of hydrogen when it is reduced in the citric acid cycle during aerobic respiration.

2 3phosphate is one of the two products of breakdown of fructose 1,6-phosphate in glycolysis, along with dihydroxyacetone phosphate.

triphosphate is a multifunctional nucleotide that is most important as a molecular currency of intracellular energy transfer.

5 _____ is respiration under anaerobic conditions with no external electron acceptor.

6 phosphate (DHAP) is one of the two products of breakdown of fructose 1,6-phosphate in glycolysis, along with glyceraldehyde 3-phosphate.

 glycolysis and the citric acid cycle.

1 A ______ organism is an organism that has an oxygen based metabolism.

12 The malate-aspartate is a biochemical system for translocating electrons produced during glycolysis across the impermeable inner membrane of the mitochondrion for oxidative phosphorylation in eukaryotes.

is an important molecule in metabolism, used in many biochemical reactions. Its main use is to convey the carbon atoms within the acetyl group to the Krebs Cycle to be oxidized for energy production.

alternatively known as a phosphotransferase, is a type of enzyme that transfers phosphate groups from high-energy donor molecules, such as ATP, to specific target molecules.

18 Glucose _____ is an enzyme that catalyzes the conversion of glucose into fructose.

19 _____ is an enzyme that facilitates phosphorylation of glucose to glucose-6phosphate.

20 _____ is the addition of a phosphate group to a protein molecule or a small molecule.

22 The mitochondrial membrane forms internal compartments known as cristae, which allow greater space for the proteins such as cytochromes to function properly and efficiently.

27 ____s is the diffusion of ions across a selectively-permeable membrane, often specifically with reference to the generation of ATP by the movement of hydrogen ions across a membrane during cellular respiration.

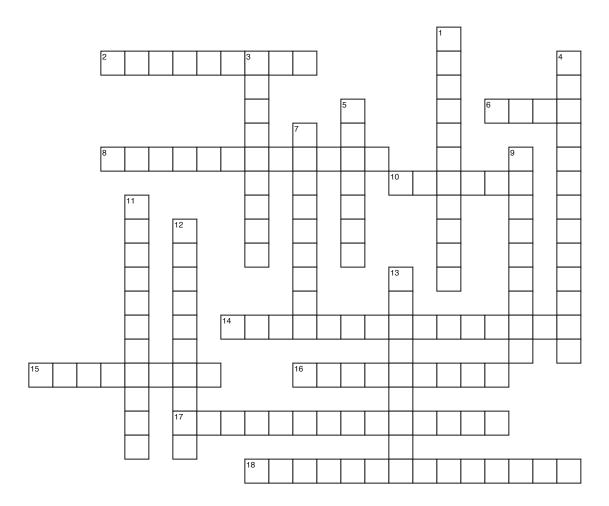
dinucleotide is an important coenzyme found in cells, which plays key roles as a carrier of electrons and a participant in metabolic redox reactions, as well as in cell signaling.

acid fermentation is a form of anaerobic respiration that occurs in some bacteria and animal cells in the absence of oxygen.

2 The _____ acid cycle, also known as the tricarboxylic acid cycle or Krebs cycle, is a series of enzymecatalysed chemical reactions of central importance in all living cells that use oxygen as part of cellular respiration.

34 A proton is an integral membrane protein that is capable of moving protons across the membrane of a cell, mitochondrion, or other subcellular compartment.

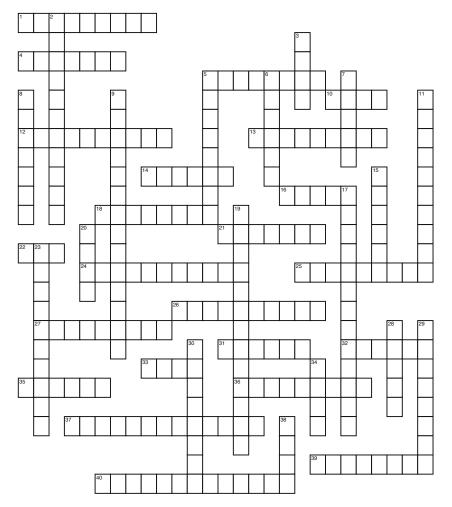
Metabolic Integration



ACROSS

is the process by which an amine group is introduced into an organic molecule. 6 The ___ _ cycle (also known as the ornithine cycle) is a cycle of biochemical reactions occurring in many animals that converts ammonia into a less toxic substance. ___ is the process of glycogen synthesis, in which glucose molecules are added to chains of glycogen. 8 _ cycle is when two metabolic pathways run simultaneously in opposite directions and have no overall effect other than wasting 10 A ___ is the generation of glucose from non-sugar carbon substrates like pyruvate, lactate, glycerol, and glucogenic amino acids. 15 The triterpene isoprenoid compound _____ is the biochemical precursor to the whole family of steroids. ___ pathway is a process that serves to generate NADPH and the synthesis of pentose sugars. (or aminotransfer) is the reaction between an amino acid and an alpha-keto acid in which the amino group is transferred from the former to the latter. is the catabolism of glycogen by removal of a glucose monomer and addition of phosphate to produce glucose-1-phosphate. 18 DOWN pyrophosphate is an intermediate in the mevalonate pathway used by organisms in the biosynthesis of terpenes and terpenoids, and thence a host of other substances such as steroids. _ is an enzyme that catalyzes the conversion of glucose into fructose or the reverse process. _ connects the pentose phosphate pathway to glycolysis in mammals, feeding excess sugar phosphates into the main carbohydrate metabolic pathways. ___ diphosphate glucose is a nucleotide that is used as an activated form of glucose, a substrate for glycosyltransferases. acid is a key organic compound in biochemistry. It is a precursor in the biosynthetic pathway, known as the HMG-CoA reductase pathway, that produces terpenes and steroids. 9 HMG-CoA is the first enzyme of the mevalonate pathway that produces terpenes, terpenoids, steroids and various other biomolecules. is an enzyme of the ligase class that catalyzes the irreversible carboxylation of pyruvate to form oxaloacetate. 11 Pyruvate Oxaloacetate can then either proceed to the citric acid cycle or to gluconeogenesis. 12 The ______ pathway is an important cellular metabolic pathway for the production of dimethylallyl pyrophosphate and isopentenyl pyrophosphate, substances which are the building blocks of a large variety of biomolecules.

____ (also known as Robison ester) is glucose sugar phosphorylated on carbon 6.



- A protein _____, also called a pro-protein or pro-peptide, is an inactive protein that can be turned into an active form by posttranslational modification.
- 4 A gene _____ is the biochemical material, either RNA or protein, resulting from expression of a gene.
- 5 Alternative is the variation mechanism in which the exons of the primary gene transcript, the pre-mRNA, are separated and reconnected so as to produce alternative ribonucleotide arrangements.
- 10 The genetic ______ is the set of rules by which information encoded in genetic material is translated into proteins by living cells.
- 12 RNA ______ is an enzyme that makes an RNA copy of a DNA or RNA template.
- 13 A ______ protein is any of the proteins that, in conjunction with rRNA, make up the subunits of the ribosome.
- 14 The _____ region of a gene is the portion of DNA that is transcribed into mRNA and translated into proteins.
- The codon ATG in DNA, which corresponds to AUG in RNA, is the codon or initiation codon which the amino acid methionine in eukaryotes and a modified

- methionine in prokaryotes.
- 18 A ______ is a small, dense, structure found in most known cells that assembles proteins in a process called translation.
- 21 A ______ transcript is an RNA molecule that has not yet undergone any modification after its synthesis.
- 22 A _____-regulatory element is a region of DNA or RNA that regulates the expression of genes located on that same strand.
- 24 Messenger RNA is decoded in the process of to produce a specific polypeptide according to the rules specified by the genetic code.
- 25 _____ ribonucleic acid is a molecule of RNA encoding a chemical blueprint for a protein product.
- 26 A ______ gene is a gene that codes for any RNA or protein product other than a regulatory element.
- 27 Enzyme ______ is a process in which a molecule, such as a drug, induces the expression of an enzyme.
- 31 The _____ recognition particle is a protein-RNA complex that recognizes and transports specific proteins to the endoplasmic reticulum in eukaryotes and the plasma membrane in prokaryotes.
- 32 ____s are non-coding sections of DNA which are spliced out once

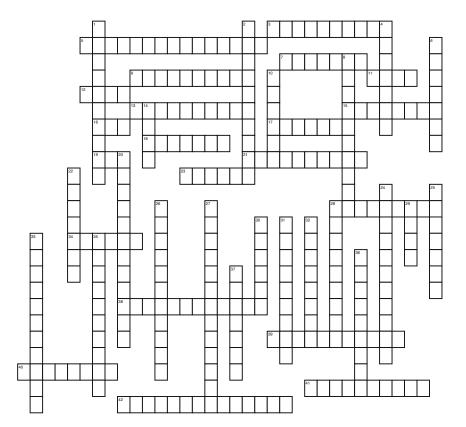
- a DNA sequence has been transcribed as a hnRNA strand.
- 33 A base _____ consists of two nucleotides on opposite complementary DNA or RNA strands connected via hydrogen bonds
- **35** A ______ is an enzyme that can catalyse the joining of two large molecules by forming a new chemical bond.
- 36 Protein _____ or sorting is the mechanisms by which a cell transports proteins to the appropriate positions in the cell or outside of it.
- 37 ______ is a type of eukaryotic gene organization in which genes may not be physically linked, but they are involved in the same process and they are coordinately expressed
- 39 A ______ is a regulatory region of DNA located upstream of a gene, providing a control point for regulated gene transcription.
- 40 A _____ factor is a protein that binds to specific parts of DNA using DNA binding domains as part of the system that controls the transfer of genetic information from DNA to RNA

DOWN

2 _____s are enzymes that cleave nucleotides one at a time from an end of a polynucleotide

- chain.
- 3 A _____ is a locatable region of genomic sequence, corresponding to a unit of inheritance.
- 5 A DNA ______ is a succession of letters representing the primary structure of a DNA molecule or strand.
- 6 The non-____ or template strand is the DNA strand that is read by the RNA polymerase.
- 7 The central of molecular biology is a framework for understanding the transfer of sequence information between sequential information-carrying biopolymers in living organisms.
- 8 A signal ______ is a short portion of a protein dedicated to directing the post-translational transport of a protein.
- 9 _____ refers to the end-to-end chemical orientation of a single strand of nucleic acid.
- 11 A ______ is a section of genetic sequence that marks the end of gene or operon on genomic DNA for transcription.
- 15 A ______ is a functioning unit of key nucleotide sequences including an operator, a common promoter, and one or more structural genes, which are controlled as a unit to produce messenger RNA.
- 17 Post_____ modification is the chemical modification of a protein after its translation.
- 19 ______ is the process by which genetic information from DNA is transferred into RNA.
- 20 A ______ box (also called Goldberg-Hogness box) is a DNA sequence found in the promoter region of most genes in eukaryotes, which is considered to be the core promoter sequence.
- 23 ______ factors are proteins that bind to the small subunit of the ribosome during the initiation of protein synthesis.
- 28 A reading ______ is a contiguous and non-overlapping set of three-nucleotide codons in DNA or RNA
- 9 A ______ is a short region of DNA that can be bound with proteins to enhance transcription levels of genes in a gene-cluster.
- RNA is a small RNA chain that plays a role during translation in shuttling a specific amino acid to a growing polypeptide chain at the ribosomal site of protein synthesis.
- 34 A _____ codon, or termination codon, is a nucleotide triplet within messenger RNA that signals a termination of translation.
- 38 A ____ is any region of DNA within a gene that is transcribed to the final messenger RNA molecule, rather than being spliced out from the transcribed RNA molecule.

Gene Expression Part 2



ACROSS

- 3 In prokaryotic cells, the specific regulatory sequence that, when transcribed into RNA, forms hairpin structures to stop translation when certain conditions are not met.
- is the covalent linkage of a poly(A) tail to a messenger RNA molecule. It is part of the route to producing mature messenger RNA for translation.
- is a molecule that starts gene expression.
- sequence is a promoter, enhancer or other segment of DNA where proteins such as transcription factors bind preferentially.
- DNA is a collective label for the portions of the DNA sequence of a chromosome or a genome for which no function has yet been identified
- 12 N-Formylmethionine, often abbreviated as is a modified form of methionine in which a formyl group has been added to methionine's amino group.
- 13 RNA_ transcribes DNA to synthesize ribosomal 5S rRNA, tRNA and other small RNAs.
- is an RNA molecule that catalyzes a

- chemical reaction.
- 16 5S ribosomal _ is a component of the large ribosomal subunit in both prokaryotes and eukaryotes.
- **17** The factor is a protein that recognises the termination codon or stop codon in a mRNA sequence on the ribosome
- The term RNA describes those molecular processes in which the information content is altered in a RNA molecule through a chemical change in the bases themselves.
- receptor also called docking protein, is a dimer composed of 2 different subunits that are associated exclusively with the rough ER in mammalian cells
- 21 An aminoacyl tRNA _ is an enzyme that catalyzes the esterification of a specific amino acid or its precursor to one of all its compatible cognate tRNAs to form an aminoacyl-tRNA.
- 23 A transcription a molecular structure that occurs during the transcription or replication of DNA when DNA helicase and DNA topoisomerase unzip the DNA double strand.
- is a DNAbinding protein that regulates the expression of one or more genes by decreasing the rate of

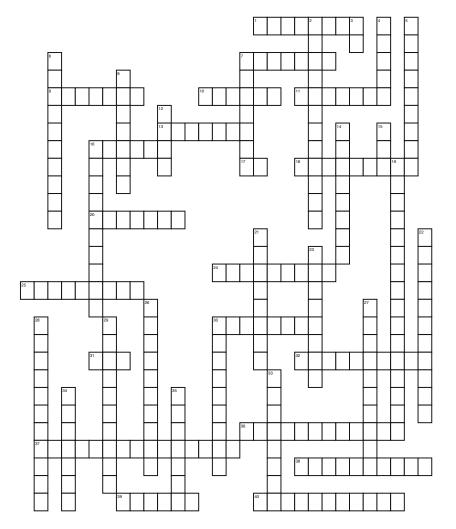
- transcription.
- 34 The Trp is a functional in certain bacteria that controls the production of gene products to increase the production of tryptophan in the absence of tryptophan in the environment.
- 38 RNA is a mechanism for RNA-guided regulation of gene expression in which doublestranded ribonucleic acid inhibits the expression of genes with complementary nucleotide sequences.
- is a protein that decreases gene expression by binding to a transcription factor which contains a DNA binding domain.
- 40 A is a collection of genes under regulation by the same stimulus.
- region is a stretch of DNA sequences located between clusters of genes that comprise a large percentage of the human genome but contain few or no genes.
- gene expression is the activation of genes within specific tissues of an organism at specific times during development.

DOWN

1 RNA transcribes DNA to synthesize ribosomal RNA. 2 Present in eukaryotic cells. RNA II catalyzes

- the transcription of DNA to synthesize precursors of mRNA and most snRNA and microRNA.
- is a collection of genes under regulation by the same regulatory protein.
- 6 A localizing sequence is an amino acid sequence which acts like a 'tag' on the exposed surface of a protein to target the protein to the cell nucleus through the nuclear pore complex.
- refers to the whole set of gene expression in a cell, tissue, organ, organisms, and species.
- 10 The lac is a functional unit of nucleotide sequences conrolling the production of gene products required for the transport and metabolism of lactose in Escherichia coli and some other enteric bacteria.
- reading frame is a portion of an organism's genome which contains a sequence of bases that could potentially encode a protein.
- 20 A single nucleotide is a DNA sequence variation occurring when a single nucleotide in the genome differs between members of a species
- 22 A gene regulatory is a collection of DNA segments in a cell which interact with each

- other and with other substances in the cell, to govern the rates at which the associated genes are transcribed.
- 24 DNA involves the addition of a methyl group to DNA.
- **25** The box is the sequence TATAAT of six nucleotides that is an essential part of a promoter site on DNA for transcription to occur in prokaryotes
- 26 refers to features such as chromatin and DNA modifications that are stable over rounds of cell division but do not involve changes in the underlying DNA sequence of the organism.
- is the set of all messenger RNA molecules produced in one or a population of cells.
- 28 Tryptophan _ DNA binding protein which silences a set of genes involved in tryptophan production.
- -loop intramolecular base pairing is a pattern that can occur in single-stranded DNA or, more commonly, in RNA The structure is also known as a hairpin or hairpin loop.
- base pairing is a process of using modified base pairs in the first base of the anti-codon. It describes how the genetic code makes up for the disparity in the number of codons and tRNA molecules
- is a DNAbinding protein that regulates one or more genes by increasing the rate of transcription by recruiting RNA polymerase to the promoter region.
- 32 Polyribosomes, or s, are a cluster of ribosomes, bound to an mRNA molecule.
- is a protein that increases gene expression by binding to an activator or transcription factor which contains a DNA binding domain.
- factors are a set of proteins that facilitate the events of protein synthesis from the formation of the first peptide bond to the formation of the last one.
- DNA-binding protein which inhibits the expression of genes coding for proteins involved in the metabolism of lactose in bacteria.
- 37 RNAproteins are typically cytoplasmic and nuclear proteins that associate with and facilitate the translation of RNAs.



- over is the process by which two chromosomes, paired up during prophase 1 of meiosis, exchange some portion of their DNA.
- 7 The ______ strand is the DNA strand at the opposite side of the replication fork from the lagging strand.
- 9 ____ is the process by which one diploid eukaryotic cell divides to generate four haploid cells often called gametes.
- 10 The _____ of replication is a particular DNA sequence at which DNA replication is initiated.
- 11 During cell division, the _____ apparatus pulls apart the chromosomes into the two daughter cells.
- 13 A _____ fragment is a relatively short fragment of DNA, with an RNA primer at the five prime terminus, created on the lagging strand during DNA replication.
- 16 A ______ is a nucleic acid strand, or a related molecule, that serves as a starting strand of nucleotides for DNA polymerase to add to.
- 17 The ______ phase is a period in the cell cycle during interphase, after cytokinesis and before the S phase.
- 18 _____ is the stage of mitosis in the eukaryotic cell cycle in which

- condensed chromosomes, carrying genetic information, align in the middle of the cell before being separated into each of the two daughter cells.
- 20 _____ is the process by which a cell duplicates its genetic information in order to generate two, identical, daughter cells.
- a is one of two identical strands of DNA making up a chromosome that are joined at their centromeres, for the process of nuclear division.
- 25 The _____ is made up of two DNA polymerase III core enzymes, which are each made up of three subunits: one with polymerization activity, one with proofreading ability and one that stimulates the proofreading.
- 30 ______, also called syndesis, is the pairing of two homologous chromosomes that occurs during meiosis.
- 31 ______ is the DNA sequence that signals for the origin of replication, sometimes referred to simply as origin.
- 32 DNA _____ III holoenzyme is the primary enzyme complex involved in prokaryotic DNA replication.
- 36 _____ is the phase of mitosis following prophase and preceding metaphase, in eukaryotic somatic cells.

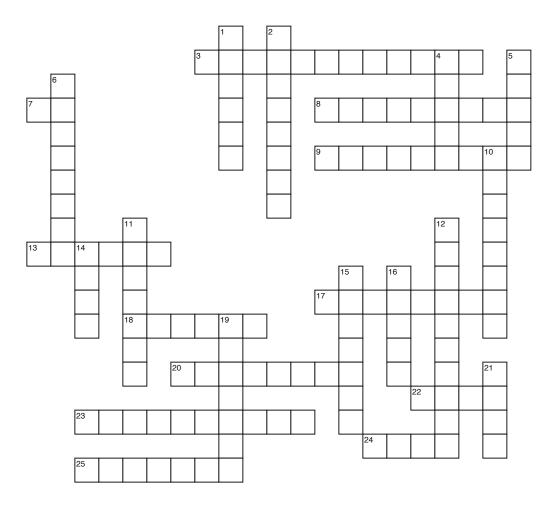
- 37 _____ is a tightly packed form of DNA. Its major characteristic is that transcription is limited.
- 38 DNA I is an enzyme that mediates the process of DNA replication in prokaryotes, an example of a processive enzymean enzyme which catalyzes a series of polymerisations.
- 39 _____ chromatids are identical copies of a chromosome.
- DNA _____ is the process of copying a double-stranded DNA molecule.

- 2 The _____ complex is a protein structure that forms between two homologous chromosomes during meiosis which is thought to mediate chromosome pairing, synapsis, and recombination.
- 3 The phase is the third, final, and usually the shortest subphase during interphase within the cell cycle in which the cell undergoes a period of rapid growth to prepare for mitosis.
- 4 The cell _____ is the series of events that take place in a eukaryotic cell leading to its replication.
- 5 ______ is a stage in either meiosis or mitosis in which nuclei reform and chromatin loses its condensed state.

6	chromosomes are non-
	identical chromosomes that contain
	information for the same biological
	features and contain the same
	genes at the same loci but possibly
	different genetic information at
	those genes.

- 7 The ______ strand is the DNA strand opposite the replication fork from the leading strand.
- Binary ____ is the form of asexual reproduction in most prokaryotes by which one cell divides into two cells of the same size.
- 12 The replication ____ is a structure that forms during DNA replication having two branching prongs, each one made up of a single strand of DNA.
- 14 ______ is the stage of mitosis when chromosomes separate in a eukaryotic cell.
- 15 _____ ligase is a particular type of ligase that can link together strands that have double-strand brooks
- 16 A DNA _____ is an enzyme that assists in DNA replication, catalyzing the polymerization of deoxyribonucleotides alongside a DNA strand.
- 19 _____ replication describes the method by which DNA is replicated which produces two copies each containing one of the original strands and one entirely new strand.
- 21 _____ is a stage of mitosis in which chromatin condenses into a highly ordered structure called a chromosome.
- 22 _____ is the process whereby the cytoplasm of a single cell is divided to spawn two daughter cells.
- 23 Cell _____ is a process by which a cell, called the parent cell, divides into two cells, called daughter cells.
- 26 A _____ is the region in the middle of a chromosome where sister chromatids join in the double chromosomal structure during mitosis, prophase and metaphase.
- is a phase of the cell cycle, defined only by the absence of cell division.
- 28 The _____ is the protein structure in eukaryotes which assembles on the centromere and links the chromosome to microtubule polymers from the mitotic spindle during mitosis and meiosis.
- 29 A ______ is a single large macromolecule of DNA, and constitutes a physically organized form of DNA in a cell.
- 30 The S phase, short for _____ phase, is a period in the cell cycle during interphase, between G1 phase and the G2 phase.
- 33 A ______ is a region of highly repetitive DNA at the end of a linear chromosome that functions as a disposable buffer.
- 34 _____s are the chief protein components of chromatin. They act as spools around which DNA winds, and they play a role in gene regulation.
- 35 DNA ______ is activated by DNA helicase to synthesize a short RNA primer as a starting point for replication.

Transmission Genetics

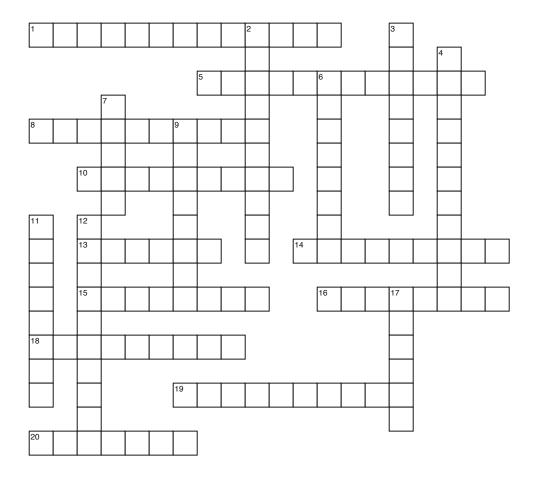


ACROSS

- 3 ______ is the proportion of phenotypic variation in a population that is attributable to genetic variation among individuals.
- 7 The ______ sex-determination system is the sex-determination system found in humans, most other mammals, and some insects.
- 8 A _____ relationship refers to how the alleles for a locus interact to produce a phenotype.
- 9 The _______ describes the ourward physical appearance or manifestation of a trait in an organism, as opposed to its genotype.
- 13 F1 ______ is a term used in genetics and selective breeding to describe the first filial generation offspring resulting from a cross mating of distinctly different parental types.
- 17 A _____ is a non-sex chromosome.
- 18 A ______ is a viable DNA coding that occupies a given locus on a chromosome.
- 20 The ______ describes the genetic constitution of an individual, that is the specific allelic makeup of an individual, usually with reference to a specific character under consideration.
- 22 A _______ breeding organism is an organism having certain biological traits which are passed on to all subsequent generations when bred with another organism of the same type for the same traits.
- 23 _____ is a term used in genetics describing the proportion of individuals carrying a particular variation of a gene that also express the particular trait.
- 24 The ______ type is the typical form of an organism, strain, gene, or characteristic as it occurs in nature.
- 25 Complete ______ is defined as the state in which two loci are so close together that alleles of these loci are virtually never separated by crossing over.

- 1 Gregor ______ (1822 1884) was an Austrian Augustinian priest and scientist often called the father of modern genetics for his study of the inheritance of traits in pea plants.
- 2 A ______ cross is a cross between two individuals identically heterozygous at two loci.
- 4 A _______ is a distinct phenotypic character of an organism that may be inherited, environmentally determined or somewhere in between.
- 5 A _______ is an individual characteristic or trait which can be possessed by an organism, such as eye colour or height, or any other observable characteristic.
- 6 ______ describes the similarity or dissimilarity of DNA between homologous chromosomes at a specific allelic position or gene.
- 10 A _____ chart is a chart which shows the known phenotypes for an organism and its ancestors.
- 11 Genetic ______ occurs when particular genetic loci or alleles for genes are inherited jointly because the loci are on the same chromosome and thus tend to segregate together during meiosis.
- 12 A _____ cross is a cross between individuals who are identically heterozygous at one locus.
- 14 A _____ body is the inactive X chromosome in a female cell.
- 15 The ______ square is a diagram used by biologists to determine the probability of an offspring having a particular genotype.
- **16** A ______ is a fixed position on a chromosome.
- 19 Sex ______ is the phenotypic expression of an allele that is related to the gender of the individual and is found on the sex chromosomes.
- 21 A _____ cross crosses a homozygous recessive with an individual with an unknown genotype.

Mutation

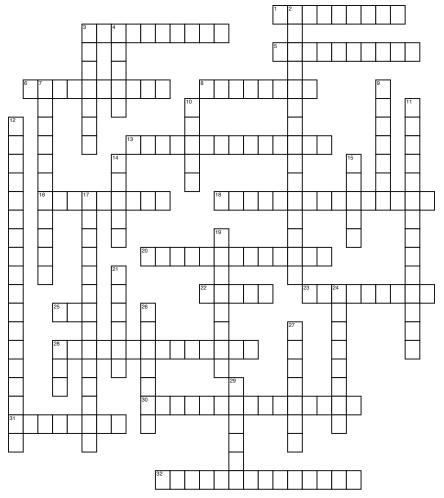


ACROSS

1 Genetic __ $_$ is the process by which a strand of DNA is broken and then joined to the end of a different DNA molecule. 5 _____ is the process by which bacterial DNA is moved from one bacterium to another by a virus. recombination involves the alignment of similar sequences, a crossover between the aligned DNA strands, and breaking and repair of the DNA to produce an exchange of material between the strands. is the addition of one or more nucleotide base pairs into a genetic sequence. 10 Genetic 13 DNA _ refers to a collection of processes by which a cell identifies and corrects damage to the DNA molecules that encode its genome. is a chromosome rearrangement in which a segment of a chromosome is reversed end to end. _ mutations or nonsynonymous mutations are types of point mutations where a single nucleotide is changed to cause substitution of a different amino acid. ___ mutation is a point mutation in a sequence of DNA that results in a premature stop codon, or a nonsense codon in the transcribed mRNA, and possibly a truncated, and often nonfunctional protein product. ___ sequence is a short DNA sequence that acts as a simple transposable element. may occur during meiotic division through a process by which DNA sequence information is transferred from one DNA helix to another DNA helix, whose sequence is altered. __ is a physical or chemical agent that changes the genetic information of an organism. 20 A **DOWN** s are sequences of DNA that can move around to different positions within the genome of a single cell, a process called transposition. __s are changes to the base pair sequence of the genetic material of an organism. ___ is the transfer of genetic material between bacteria through direct cell-to-cell contact. 6 A _ is a mutation in which a part of a chromosome or a sequence of DNA is missing. $_$ mutation, or single base substitution, is a type of mutation that causes the replacement of a single base nucleotide with another nucleotide. __ mutation is any detectable, heritable variation in the lineage of germ cells. _____ over is the process by which two chromosomes, paired up during prophase 1 of meiosis, exchange some portion of their DNA. 11 __ mutation is a genetic mutation caused by inserts or deletes from a DNA sequence of a number of nucleotides not evenly

_ mutations are DNA mutations that do not result in a change to the amino acid sequence of a protein.

Genomics Laboratory



ACROSS

- 1 The ______factor (also known as F factor or sex factor) is a bacterial DNA sequence that allows a bacterium to produce a sex pilus necessary for conjugation.
- The chain reaction is a technique widely used in molecular biology to exponentially amplify a fragment of DNA by in vitro enzymatic replication.
- 5 The term DNA encompasses biochemical methods for determining the order of the nucleotide bases in a DNA oligonucleotide.
- 6 _____ bacteria refers to bacteria which have been genetically engineered.
- 8 chromatography is a chromatographic method of separating biochemical mixtures, based on a highly specific biologic interaction such as that between antigen and antibody or enzyme and substrate.
- 13 _____ is the collective term for laboratory techniques which separate

- analytes dissolved in a mobile phase by passing them through a stationary phase.
- 16 electrophoresis separates biological macromolecules based on their size to charge ratio in the interior of a very narrow tube filled with an electrolyte.
- 18 DNA _____ is an analytical technique used to separate DNA fragments by size though the use of an electric field which forces the fragments to migrate through a gel.
- is a separation process in which a certain quantity of a mixture is divided up in a large number of smaller quantities in which the composition changes according to a gradient.
- 23 _____ refers to the liquid or clear fluid above a sediment or precipitate.
 25 ____-PAGE, officially

- sodium dodecyl sulfate polyacrylamide gel electrophoresis, is a technique used to separate proteins according to their electrophoretic mobility.
- 28 _____ is the genetic alteration of a cell resulting from the uptake and expression of foreign genetic material.
- 30 The _____ is a centrifuge optimized for spinning a rotor at very high speeds, capable of generating acceleration as high as 1,000,000 g.
- 31 lon chromatography is a process that allows the separation of ions and polar molecules based on the charge properties of the molecules.
- an unknown protein is cleaved into peptides by a protease such as Trypsin to form a collection of peptides serving as a unique identifier of the unknown protein.

DOWN

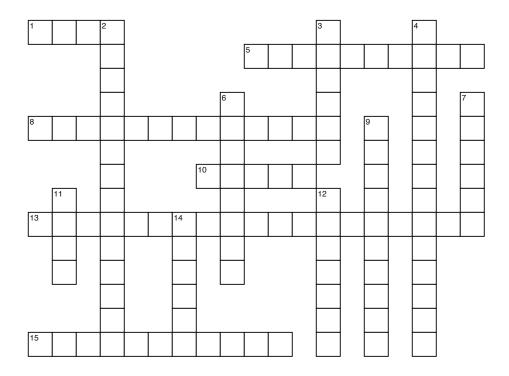
2 Gel _____ is a technique used for the

- separation of biological molecules using an electric field.
- 3 A ______ is a DNA molecule separate from chromosomal DNA which is capable of autonomous replication. It is typically circular and double-stranded.
- 7 A ______enzyme is an enzyme that cuts double-stranded DNA in such a way that the fragments from different chromosomes or genes can be spliced together by ligases.
- g is a material used to form a common type of electrophoresis gel which is derived from the cell membranes of some species of red algae or seaweed.
- degradation is a method of sequencing amino acids in a peptide in which the amino-terminal

residue is labeled and cleaved from the peptide without disrupting the peptide bonds between other amino acid residues.

- 11 _____ is a process that involves the use of the centripetal force for the separation of mixtures.
- are nucleotides lacking a three prime hydroxyl group on their deoxyribose sugar. After one is added by a DNA polymerase to a growing nucleotide chain, no further nucleotides can be added.
- 14 A ______ is a hairlike appendage found on the surface of many bacteria.
- 15 A hybridization is a labeled fragment of DNA of variable length which is used to detect in DNA or RNA samples the presence of nucleotide sequences that are complementary to the sequence in the fragment.
- 17 is a general term in biochemistry that applies to any use of an antibody-based method to detect a specific protein in a sample.
- 19 _____ is the process of separating molecules in solution by the difference in their rates of diffusion through a semipermeable membrane.
- 21 A restriction ______ is a procedure which uses a restriction enzyme to selectively cleave DNA strands into shorter segments, which are more suitable for analytical techniques such as chromatography.
- 24 A ______ is a phage genome inserted as part of the linear structure of the DNA chromosome of a bacterium.
- 26 In _____ sequencing,
 DNA is broken up randomly
 into numerous small
 segments, which are
 sequenced. The process is
 repeated until multiple
 overlapping reads for the
 target DNA are obtained
- 27 Molecular _____ refers to the procedure of isolating a defined DNA sequence and obtaining multiple copies of it in vivo.
- polymerase is a thermostable DNA polymerase named after the thermophilic bacterium Thermus aquaticus from which it was isolated. This enzyme is able to withstand the high temperature required during PCR.
- 29 A cloning _____ is a small DNA vehicle that carries a foreign DNA fragment.

Human Genetics

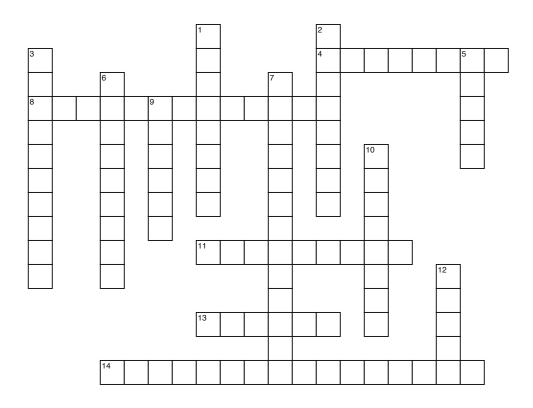


ACROSS

- 1 ______ syndrome or trisomy 21 is a genetic disorder caused by the presence of all or part of an extra 21st chromosome.
- 5 ______ is a change in the number of chromosomes that can lead to a chromosomal disorder.
- 8 A ______ is a section of DNA that consists of a short series of 10-100 bases. These types of sequences occur at more than 1000 locations in the human genome.
- **10** A ______'s peak is a descending V-shaped point in the middle of the hairline. This is a dominant genetic trait.
- 13 ______, also known as PTC, or phenylthiourea, is an organic compound which either tastes very bitter, or virtually tasteless, depending on the genetic makeup of the taster.
- 15 ______'s syndrome or XXY syndrome is a condition caused by a chromosome aneuploidy in which affected males have an extra X sex chromosome.

- 2 _____ is the failure of chromosome pairs to separate properly during meiosis or mitosis.
- 3 _____ syndrome encompasses several chromosomal abnormalities, of which monosomy X is the most common.
- 4 Genetic ______ is a technique used to distinguish between individuals of the same species using only samples of their DNA, exploiting highly variable repeating sequences called minisatellites.
- 6 The recessive trait ______ is a form of hypopigmentary congenital disorder, characterized by a lack of melanin pigment in the eyes, skin and hair. It is a recessive genetically inherited trait.
- 7 _____s are visible indentations of the skin, a dominant genetic trait which shows on some people's cheeks when they smile.
- **9** Sexadactyly or hexadactyly is dominant genetic condition in which a person has six fingers on one or both hands, or six toes on one or both feet.
- 11 Cri du _____ syndrome is due to a partial deletion of the short arm of chromosome number5.
- 12 _____s are small colored spots of melanin on the exposed skin. Having them is genetic and is related to the presence of the melanocortin-1 receptor MC1R gene variant, which is a dominant trait
- 14 _____ X syndrome is a form of chromosomal variation characterized by the presence of an extra X chromosome in each cell of a human female.

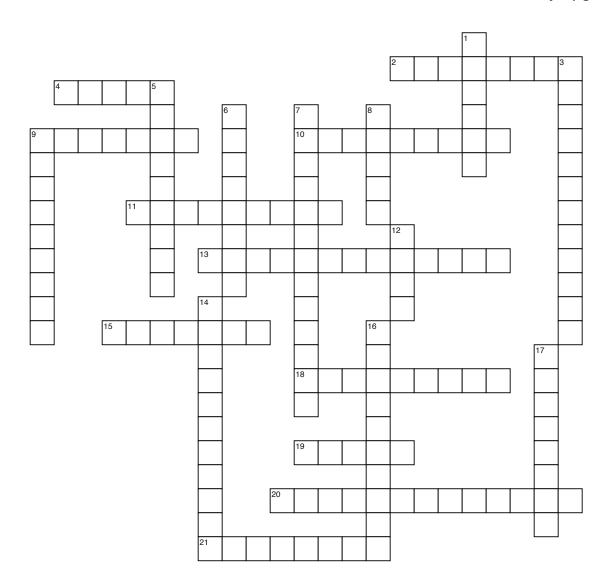
Answer key - pg 115 Viruses



ACROSS

- 4 _____-sense viral RNA is complementary to the viral mRNA and thus must be converted to positive-sense RNA by an RNA polymerase prior to translation.
- 8 A reverse ______, also known as RNAdependent DNA polymerase, is a DNA polymerase enzyme that transcribes single-stranded RNA into single-stranded DNA.
- 11 The _____ cycle is the mode of viral reproduction characterized by the fusion of the nucleic acid of a bacteriophage with that of a host so that the integrated prophage can then be transmitted to daughter cells in subsequent cell divisions.
- 13 Tobacco _____ virus is an RNA virus that infects plants. It was the first virus to be discovered.
- 14 Human ______ virus (HIV) is a retrovirus that can lead to AIDS, a condition in humans in which the immune system begins to fail.

- _____-sense viral RNA is viral RNA which may be directly translated into the desired viral proteins.
 Viral ______s are present in many viruses,
- covering the protein capsid, typically derived from portions of the host cell membranes but including some viral glycoproteins.
- 3 A ______ is an enveloped viruses possessing a RNA genome, which replicates via a DNA intermediate.
- **5** A ______ is a sub-microscopic particle that can infect the cells of a biological organism and which can replicate itself only by infecting a host cell.
- 6 ______, commonly known as flu, is an infectious disease of birds and mammals caused by RNA viruses of the family Orthomyxoviridae.
- 7 A _____ is any one of a number of viruses that infect bacteria.
- 9 A _____ is the protein shell of a virus.
- **10** A _____ is a virus genome that has integrated itself into the DNA of a host cell.
- 12 The _____ cycle is the one of the two cycles of viral reproduction that is typically considered the main method of viral replication, since it results in the destruction of the infected cell and release of virus particles.

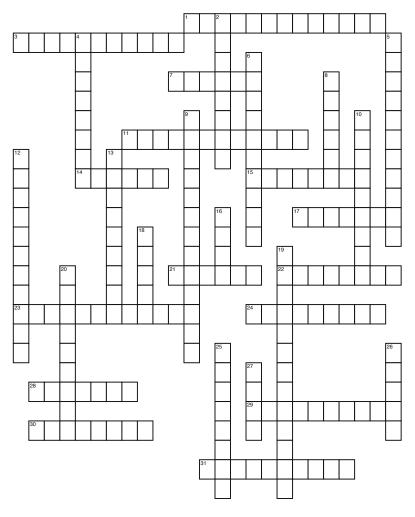


- 2 Gram— bacteria are those that retain a crystal violet dye during the Gram stain process.
- 4 A ______ is a hairlike appendage found on the surface of many bacteria.
- 9 A ______ is a proteinaceous appendage in many gram-negative bacteria that is thinner and shorter than a flacellum.
- 10 A ______ is a dormant, tough, non-reproductive structure produced by a small number of bacteria the purpose of which is to ensure the survival of the bacterium through periods of environmental stress.
- 11 ______ is a protein that arranges itself in a hollow cylinder to form the filament in bacterial flagellum
- **13** A ______ is an organism that is too small to be seen by the human eye.
- 15 The ______, or archaebacteria, are a major group of prokaryotic microorganisms in which many species are extremophiles.
- **18** A _____ is an organism that produces complex organic compounds

- from simple inorganic molecules and an external source of energy.
- 19 A ______ dish is a shallow glass or plastic cylindrical dish that biologists use to culture cells.
- 20 _____ is a phylum of bacteria which obtain their energy through photosynthesis, often referred to as bluegreen algae.
- 21 The cell ______ of a bacterium is the cell membrane and cell wall plus an outer membrane.

- 1 ______ fission is the form of asexual reproduction in prokaryotic single-celled organisms by which one cell divides into two cells of the same size.
- 3 A _______ is an organism that thrives in and may even require physically or geochemically extreme conditions that are detrimental to the majority of life on Earth.
- 5 _____ in microbiology refers to a bacterium with a cell body that twists like a spiral.
- 6 _____ are prokaryotic unicellular

- microorganisms ubiquitous in every habitat on Earth.
- 7 ______ or murein is a polymer consisting of sugars and amino acids that forms a mesh-like layer outside the plasma membrane of eubacteria.
- 8 ______ are microorganisms whose overall shape is spherical or nearly spherical.
- 9 A ______ is a long, slender projection from the cell body, composed of microtubules and surrounded by the plasma membrane.
- A cell _____ is a fairly rigid layer surrounding a cell, located external to the cell membrane.
- 14 The ______ is a bacterial microcompartment made up of a protein shell, that contains enzymes involved in carbon fixation reactions.
- 16 _____s are a group of organisms that lack a cell nucleus or any other membrane-bound organelles.
- 17 The _____ in prokaryotes is an irregularly shaped region within the cell where the genetic material is localized.



- 1 _____ is the process by which the embryo is formed and develops.
- 3 The ______ is known as the primitive gut that forms during gastrulation in the developing blastula.
- 7 The ______ nerve cord is one of the embryonic features unique to chordates, along with a notochord, a post-anal tail and pharyngeal slits.
- 11 ______ is a phase early in the development of animal embryos, during which the morphology of the embryo is dramatically restructured by cell migration.
- 14 A ______ is an embryo at an early stage of embryonic development, consisting of approximately 12-32 cells in a solid ball contained within the zona pellucida.
- 15 ______ is a primary germ layer which forms during gastrulation when some of the cells migrating inward to form the endoderm form an additional layer between the

- endoderm and the ectoderm.

 17 The term ______ pole
 refers to the hemisphere of a
 blastula embryo which
 contains large yolky cells that
 divide very slowly.
- 21 The term ______ pole refers to the hemisphere of a blastula embryo which consists of small cells that divide rapidly, in contrast with the vegetal pole.
- 22 The ______ is the primary germ layer which emerges first during embryogenesis and forms from the outermost of the germ layers.
- 23 _____ is a physiological process involving the growth of new blood vessels from pre-existing vessels.
- 24 The _____ is a flexible, rod-shaped body found in embryos of all chordates which is composed of cells derived from the mesoderm and defines the primitive axis of the embryo.
- 28 ______ is the expansion of one cell sheet over other cells. Takes place during gastrulation.

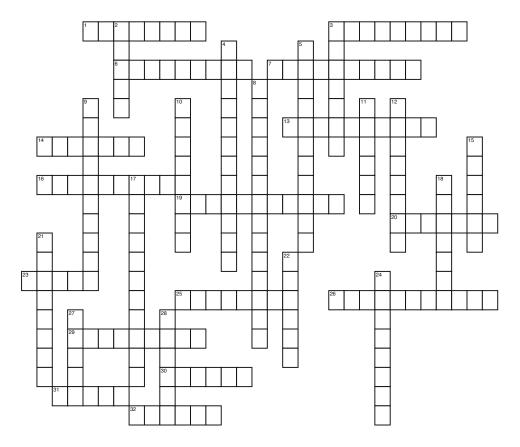
- 29 A _____ is an opening into the archenteron during the embryonic stages of an organism.
- 30 _____ is the division of cells in the early embryo.
- 31 The ______ is the structure formed in early human embryogenesis, after the formation of the blastocele, but before implantation, possessing an inner cell mass, or embryoblast, and an outer cell mass, or trophoblast.

DOWN

- 2 The ______ stage of embryonic development in animals follows the morula and precedes the gastrula stage in the developmental sequence.
- 4 _____ is one of the primary germ layers formed during animal embryogenesis when cells migrating inward along the archenteron form the inner layer of the gastrula.
- 5 A ______ is the fluid-filled central region of a blastocyst.

Embryology

- 6 _____ is the term for the cells formed by cleavage of the ovum in very early embryonic development.
- **8** A _____ is the cell that results from fertilization.
- 9 _____ is the process in animal development by which the ectoderm, endoderm, and mesoderm develop into the internal organs of the organism.
- 10 The _____ streak is a structure that forms during the early stages of embryonic development, characterized as a furrow in the midline of the embryonic disk at the future caudal end of the embryo.
- 12 ____s are cells forming the outer layer of a blastocyst, which provide nutrients to the embryo and develop into a large part of the placenta.
- 13 During the early stages of embryonic development, a shallow groove, the _____ groove, appears on the surface of the primitive streak
- 16 A ______ layer is a collection of cells, a primary tissue layer, formed during animal embryogenesis.
- 18 Formation of neural
 ______ is the first step of
 neurulation in human
 embryology. It is created by a
 flat thickening opposite to the
 primitive streak.
- 19 ______ is fusion of gametes to form a new organism of the same species.
- 20 The _____ knot is the organizer for gastrulation in vertebrates, starting as a regional knot of cells that forms on the blastodisc immediately anterior to where the outer layer of cells will begin to migrate inwards.
- 25 The _____ phase of embryonic development, which follows after the blastula stage, is seen in all animals except the sponges.
- 26 In early embryogenesis, the _____cell mass is the mass of cells inside the primordial embryo that will eventually give rise to the definitive structures of the fetus.
- 27 Following primary and secondary neurulation, the neural _____ is the developing vertebrate embryo's precursor to the central nervous system.



- 1 _____ epithelia are cubeshaped epithelial cells.
 - ____s are the cells that primarily compose adipose tissue, specialized in storing energy as fat.
- 6 The ______ epithelium is a specialized epithelial tissue inside the nasal cavity that is involved in smell.
- 7 ______tissue is largely a category of exclusion rather than one with a precise definition, but most tissues in this category are involved in structure and support, derived from mesoderm, and characterized by the traits of non-living tissue.
- 13 _____ columnar epithelia have several layers, with columnar cells as the outermost of these.
- 14 _____connective tissue is a type of connective tissue which has relatively high tensile strength, due to a relatively high concentration of collagenous fibers.
- 16 _____ is a tissue composed of layers of cells which line the cavities and surfaces of structures throughout the body. It is also the type of tissue of which many glands are formed.
- 19 The _____ is the innermost layer of tissue that lines the chambers of the heart.
- 20 The microscope, often referred to as a light microscope, is a type of microscope which uses visible light and a system of lenses to magnify images of small samples.

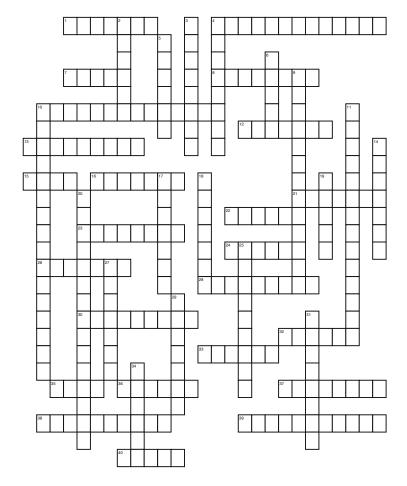
- 23 A _____ is an organ in an animal's body that synthesizes a substance for release, often into the bloodstream (endocrine) or onto a surface (exocrine).
- 25 _____ epithelium is an epithelium characterised by its most superficial layer consisting of flat, scale-like cells.
- 26 The ____ membrane of a polarized cell is the part of the plasma membrane that forms its basal and lateral surfaces, distinct from the lumenal or apical surface.
- 29 _____ is the study of tissue sectioned as a thin slice, viewed under a microscope.
- **30** The basal _____ is a layer on which epithelium sits and which is secreted by the epithelial cells.
- 31 _____ connective tissue has collagen fibers as its main matrix element.
- 32 The extracellular is the extracellular part of animal tissue that usually provides structural support to the cells in addition to performing other functions. It is the defining feature of connective tissue in animals.

- 2 ______adipose tissue is present in many newborn or hibernating mammals as well as migratory birds. Its primary purpose is to generate body heat.
- 3 _____ connective tissue (or loose connective tissue) is the most widely distributed connective tissue type in the body. It can be found in

- the skin as well as in places that connect epithelium to other tissues.
- 4 ______epithelia is a type of tissue consisting of multiple layers of epithelial cells which can contract and expand.
- 5 The _____ is the thin layer of cells that line the interior surface of blood vessels.
- _____ refers to the microscopic examination of tissue in order to study the manifestations of disease.
- 9 _____ cuboidal epithelia is epithelial tissue composed of cuboidally shaped cells arranged in multiple layers.
- 10 The _____ membrane is a structure that supports overlying epithelial or endothelial cells.
- 11 _____ cuboidal eplithelia are epithelial cells with a cuboidal shape arranged in a single layer.
- 12 _____ is a chemical process used in the fields of histology, pathology, and cell biology by which biological tissues are preserved from decay.
- 15 The ____ membrane of a polarized cell is the part of the plasma membrane that forms its lumenal surface, distinct from the basolateral membrane.
- 17 Antonie van ______ (1632 -1723) was a Dutch tradesman and scientist who is commonly known as the Father of Microbiology.
- 18 ______ tissue or fat is loose connective tissue composed of adipocytes.
- 21 _____ epithelia are epithelial cells whose heights are at least twice their width.

- 22 A biological _____ is a collection of interconnected cells that perform a similar function within an organism.
- 24 Simple ______ epithelium is made up of one layer of cells that are relatively thick and protective of the underlying tissues due to its elongated shape.
- adipose tissue is one of the two types of adipose tissue found in mammals, in humans, composing as much as 20% of the body weight in men and 25% of the body weight in women.
- 28 _____ cells are glandular simple columnar epithelial cells whose sole function is to secrete mucus.

Nervous System



ACROSS

- 1 Sodium _____s are integral membrane proteins that conduct sodium ions through a cell's plasma membrane.
- 4 The first neurotransmitter identified, the chemical compound is a neurotransmitter in both the peripheral nervous system and central nervous system.
- 7 ______ neurons begin in the central nervous system projecting their axons outside the CNS and directly or indirectly controlling muscles.
- 8 _____ nerves otherwise known as motor or effector neurons carry nerve impulses away from the central nervous system to effectors such as muscles or glands.
- 10 Epinephrine and _____ are fight or flight hormones released from the adrenal glands, which are also neurotransmitters in the central and sympathetic nervous systems.
- 12 The ______ nervous system represents the largest part of the nervous system, including the brain and the spinal cord.
- 13 The _____ nervous system is the part of the peripheral nervous system that acts as a control system, maintaining homeostasis in the body.
- 15 The ______, or perikaryon, is the bulbous end of a neuron, containing the cell nucleus.
- 16 Chemical ____s are specialized junctions through which

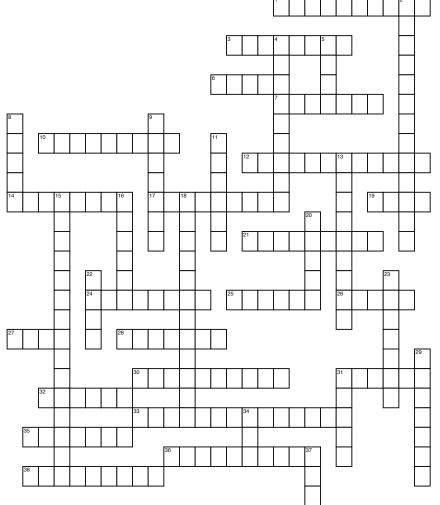
- the cells of the nervous system employ neurotransmitters to signal to each other and to non-neuronal cells such as those in muscles or glands.
- 21 ____ cells are a variety of glial cell that mainly provide myelin insulation to axons in the peripheral nervous system.
- 22 The ______ system is the part of the human brain involved in emotion, motivation, and emotional association with memory.
- 23 In a biological membrane, the ______potential (or Nernst potential) of a particular ion is the membrane voltage at which there is no net flow of ions from one side of the membrane to the other.
- 24 The _____ cord is a thin, tubular bundle of nerves that is an extension of the central nervous system from the brain, enclosed and protected by the bony vertebral column.
- neurons are nerve cells within the nervous system responsible for converting external stimuli from the organism's environment into nerve impulses relayed to the central nervous system.
- 28 _____ conduction is the means by which action potentials are transmitted along myelinated nerve fibers.
 30 Membrane is the
- electrical voltage across a cell's plasma membrane.
- 32 _____ is an electrically

- insulating phospholipid layer that surrounds the axons of many neurons.
- 33 A ______ potential is a spike of electrical discharge that travels along the membrane of a cell.
- 35 ____ matter is a major component of the central nervous system, consisting of nerve cell bodies, glial cells, capillaries, and short axons and dendrites.
- **36** A ______ action is an automatic (involuntary) neuromuscular action elicited by a defined stimulus.
- 37 The telencephalon is the name for the forebrain, a large region within the brain to which many functions are attributed, which many people refer to as the
- 38 The ______ nervous system resides or extends outside the central nervous system serving the limbs and organs.
- 39 The _____ nervous system is a branch of the autonomic nervous system, always active at a basal level and becoming more active during times of stress.
- 40 _____ cells, commonly called neuroglia, are non-neuronal cells that provide support and nutrition, maintain homeostasis, form myelin, and participate in signal transmission in the nervous system.

DOWN

2 ____s are electrically excitable cells in the nervous system that process and transmit information.

- 3 ______feedback is a feedback system in which the system responds to the perturbation with activity in the same direction as the perturbation.
- 4 _____ neurons--otherwise known as sensory or receptor neurons--carry nerve impulses from receptors or sense organs toward the central nervous system.
- 5 The fight-or-_____ response states that animals react to threats with a general discharge of the sympathetic nervous system.
- 6 The ______ or encephalon is the control center of the central nervous system.
- 9 A _____ junction is the synapse or junction of the axon terminal of a motoneuron with the motor end plate.
- 10 ____s are chemicals that are used to relay, amplify and modulate signals between a neuron and another cell.
- 11 _____ is a decrease in the absolute value of a cell's membrane potential.
- nerves are nerves that emerge directly from the brain in contrast to spinal nerves which emerge from segments of the spinal cord.
- 17 The _____ nervous system is the part of the peripheral nervous system associated with the voluntary control of body movements and with reception of external stimuli.
- 18 ______tissue is composed of neurons, which transmit impulses, and the neuroglia, which assist propagation of the nerve impulse as well as provide nutrients to the neuron.
- 19 ____ matter is composed of myelinated nerve cell processes, or axons, which connect various gray matter areas of the brain to each other.
- 20 In a kind of complementary opposition to the sympathetic nervous system, the nervous system is the division of the autonomic nervous system that functions with actions that do not correspond to stress.
- 25 _____ channels are the most common type of ion channel within the physiology.
- 27 The _____ potential of a cell is the membrane potential that would be maintained if there were no action potentials, synaptic potentials, or other active changes in the membrane potential.
- 29 Nodes of ______, also known as neurofibril nodes, are regularly spaced gaps in the myelin sheath around an axon or nerve fiber.
- sare the branched projections of a neuron that act to conduct the electrical stimulation received from other neural cells to the cell body of the neuron.
- 34 A ______ network refers to a network or circuitry of biological neurons.



- The _____ system, or balance system, is the sensory system that provides the dominant input about our movement and orientation in space.
- 3 A sensory is a structure that recognizes a stimulus in the internal or external environment, and in response, initiates sensory transduction by creating graded potentials or action potentials in the same cell or in an adjacent one
- 6 The _____ window is one of two openings along with the oval window that connect the inner ear to the middle ear.
- 7 ____s are small particles, composed of a combination of a gelatinous matrix and calcium carbonate in the viscous fluid of the saccule and utricle.
- 10 _____ refers to the sense of smell.
- 12 The _____ canals are

- three half-circular, interconnected tubes located inside each ear that are the equivalent of three gyroscopes located in three orthogonal planes.
- 14 A _____ is a detectable change in the internal or external environment.
- 17 The _____ is a system of fluid passages in the inner ear, including both the cochlea which is part of the auditory system, and the vestibular system which provides the sense of balance.
- 19 The _____ consists of pigmented fibrovascular tissue known as a stroma, which connects a sphincter muscle to contract the pupil, and a set of dilatory muscles to open it.
- 21 The ______ system is the sensory system that uses taste buds on the upper surface of the tongue to provide information about the taste of food being eaten.
- 24 The _____ are the three smallest bones in the

- human body, contained within the middle ear and serving to transmit sounds to the fluid-filled cochlea.
- 25 The ______ ear is the portion of the ear internal to the eardrum, and external to the oval window of the cochlea
- 26 The ______ ear is the bony labyrinth, a system of passages comprising two main functional parts: the cochlea and the vestibular apparatus.
- 27 _____ cells are the sensory receptors of both the auditory system and the vestibular system in all vertebrates.
- 28 The ______ is a hammer-shaped small bone or ossicle of the middle ear which connects with the incus and is attached to the inner surface of the eardrum.
- 30 The _____ tube is a tube that links the pharynx to the middle ear.
- 31 The ______ is the transparent front part of the eye that covers the iris,

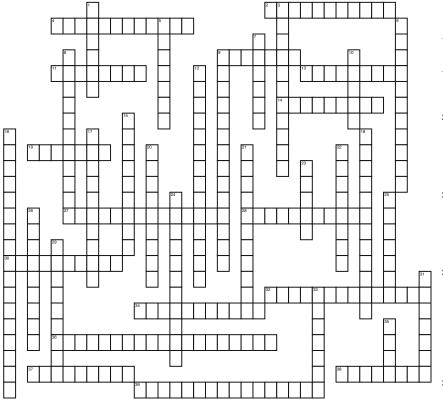
- pupil, and anterior chamber, providing most of an eye's optical power.
- 32 The ______ is a thin layer of neural cells that lines the back of the eyeball.
- 33 A ______ is a sensory receptor that responds to temperature, primarily within the innocuous range.
- 35 A light-sensitive derivative of vitamin A, is the fundamental chromophore involved in the transduction of light into visual signals.
- 36 A is a sensory receptor that sends signals that cause the perception of pain in response to potentially damaging stimulus.
- 38 _____ is an extracellular fluid located within the scala tympani and scala vestibuli of the cochlea.

DOWN

2 _____ is the process by which the eye increases

- optical power to maintain a clear image of an object as it draws near the eye.
- 4 _____ is the fluid contained in the membranous labyrinth of the inner ear.
- 5 Also called the vestibular window, the ____ window is a membrane-covered opening which leads from the middle ear to the vestibule of the inner ear.
- 8 The ______ is the anvilshaped small bone or ossicle in the middle ear.
- 9 The _____ is the auditory portion of the inner ear.
- 11 An olfactory receptor
 _____ is the primary
 transduction cell in the
 olfactory system.
- 3 ______, or visual purple, is a pigment consisting of an opsin protein and retinal cofactor that is responsible for both the formation of the photoreceptor cells and the perception of light.
- 15 A _______ is a sensory receptor that responds to mechanical pressure or distortion.
- 16 The ______ is the stirrup-shaped small bone or ossicle in the middle ear which attaches the incus to the fenestra ovalis.
- 18 ____s detect the pressure of blood flowing through them, and can send messages to the central nervous system to increase or decrease total peripheral resistance and cardiac output.
- or gustation is a form of direct chemoreception producing the ability to detect the flavor of substances such as food and poisons.
- 22 The ciliary ______ is the circumferential tissue inside the eye composed of the ciliary muscle and ciliary processes.
- 23 _____ (or audition) refers to the ability to detect sound.
- 29 The tympanic membrane, colloquially known as the ______, is a thin membrane that separates the external ear from the middle ear.
- 31 The organ of ______, or spiral organ, is the organ in the inner ear of mammals that contains auditory sensory cells, or hair cells.
- 34 ____s are organs of vision that detect light.
- photoreceptor cells in the retina of the eye that can function in less intense light than cone cells can.

Endocrine System



ACROSS

- The ______ glands are small endocrine glands in the neck, usually located behind the thyroid gland.
- 4 Located just below the thalamus, the _____ links the nervous system to the endocrine system via the pituitary gland.
- 9 One of the largest endocrine glands in the body, the controls how quickly the body burns energy, makes proteins, and how sensitive the body should be to other hormones.
- Under the influence of the hypothalamus, the _____ pituitary produces and secretes
 - several peptide hormones which regulate many physiological processes including stress, growth, and reproduction.
- 13 A secondary ____ system is a method of cellular signalling where the signalling molecule does not enter the cell, but rather utilizes a cascade of events that transduces the signal into a cellular change.
- 14 The ______ system is an integrated system of small organs which involve the release of extracellular signaling molecules known as hormones.
- 19 The _____ glands are triangle-shaped endocrine glands that sit on top of the

- kidneys which regulate the stress response through the synthesis of corticosteroids and catecholamines.
- 27 Epinephrine and ______ are a fight or flight hormones released from the adrenal glands, and they are also neurotransmitters within the central and sympathetic nervous systems.
- hormone is secreted by the parathyroid glands, acting to increase the concentration of calcium in the blood, opposite the effect of the thyroid hormone calcitonin.
- 30 _____ is a polypeptide hormone produced by the parafollicular cells of the thyroid which participates in calcium and phosphorus metabolism, lowering blood calcium, countering the effects of parathyroid hormone.
- 32 _____, also called hematopoietin, is a glycoprotein hormone that acts as a cytokine for erythrocyte precursors in the bone marrow.
- 34 _____ is a steroid hormone produced by the outer-section of the adrenal cortex in the adrenal gland which regulates sodium and potassium balance in the blood.
- 36 _____ hormone or corticotropin is a polypeptide hormone produced and secreted by

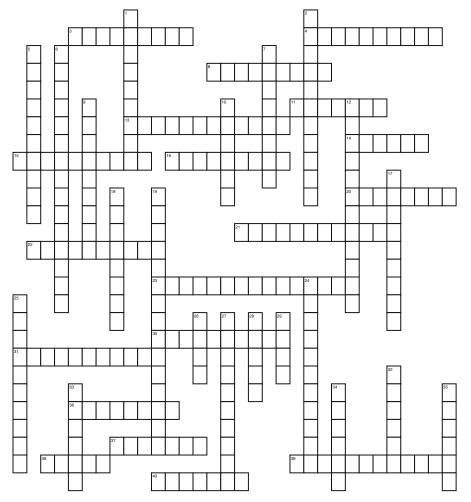
- the pituitary gland. It is an important player in the hypothalamic-pituitary-adrenal axis.
- 37 _____, often abbreviated as T4, is the major hormone secreted by the follicular cells of the thyroid gland.
- 38 _____ is released by the pancreas when the glucose level in the blood is low, causing the liver to convert stored glycogen into glucose and release it into the bloodstream.
- 39 _______(T3) is a thyroid hormone similar to thyroxine but with one less iodine atom per molecule. T3 exhibits greater activity than thyroxine but is produced in smaller quantity.

- The adrenal _____, situated along the perimeter of the adrenal gland, mediates the stress response through the production of mineralocorticoids and glucocorticoids.
- 3 _____ is an oligopeptide in the blood that causes vasoconstriction, increased blood pressure, and release of aldosterone from the adrenal cortex.
- 5 The adrenal _____ is the part of the adrenal gland located in the center of the gland, being surrounded by

- the adrenal cortex.
- 6 Atrial _______ peptide is a polypeptide hormone involved in the homeostatic control of body water, sodium, and adiposity.
- 7 _____ hormone or somatotropin is a protein hormone which stimulates growth and cell reproduction.
- 8 is an oligopeptide in the blood that causes vasoconstriction, increased blood pressure, and release of aldosterone from the adrenal cortex.
- 9 _____ is a glycoprotein hormone produced mainly by the liver and the kidney that regulates the production of platelets by the bone marrow.
- 10 _____ is released by the juxtaglomerular apparatus of the kidneys in response to low blood volume or decreased serum NaCl concentration.
- s are a class of steroid hormones characterised by an ability to bind with cortisol receptors and trigger events which increase and maintain normal concentrations of glucose in blood.
- glands are glands are glands that secrete their product directly into the blood rather than through a duct.
- 16 _____s are a class of steroid hormones that

- include aldosterone which influence salt and water balance.
- 17 The endocrine cells of the pancreas are grouped in the islets of _____.
- 18 Signal ______ refers to any process by which a cell converts one kind of signal or stimulus into another.
- 20 The _____ pituitary gland and is part of the endocrine system, secreting the hormones oxytocin and vasopressin, which are produced in the hyoothalamus.
- 21 Arginine ______, also known as antidiuretic hormone, plays a major role in regulating the body's retention of water, being released when the body is dehydrated.
- 22 _____ is a hormone that also acts as a neurotransmitter in the brain. In females, it is released in large amounts during labor and it is also released during breastfeeding. It is released in both sexes during
- cells are
 endocrine cells in the islets
 of Langerhans of the
 pancreas which are
 responsible for synthesizing
 and secreting the peptide
 hormone glucagon.
- 24 _____releasing
 hormone is a tripeptide
 hormone that stimulates the
 release of thyroidstimulating hormone and
 prolactin by the anterior
 pituitary.
- 25 A ______ is a chemical messenger that carries a signal from one cell, or group of cells, to another.
- 26 _____ is a peptide hormone on the anterior pituitary gland primarily associated with lactation.
- 29 The _____gland, or hypophysis, is an endocrine gland about the size of a pea that sits in a small, bony cavity covered by a dural fold at the base of the
- hormone which causes liver and muscle cells to take in glucose and store it in the form of glycogen and causes fat cells to take in blood lipids and turn them into triglycerides.
- 33 Peptide _____s are a class of peptides that are secreted into the blood stream and have endocrine functions in living animals.
- 35 ____ cells are a type
 of cell in the pancreas in
 areas called the islets of
 Langerhans which make
 and release insulin.

skeleton.



ACROSS

- s are cylindrical organelles, found within muscle cells, which are bundles of actomyosin filaments that run from one end of the cell to the other, attached to the cell surface membrane at each end.
- 4 A _____ is a mononucleate cell that is responsible for bone formation.
- 8 The a star-shaped is the most abundant cell found in bone, derived from osteoblasts after they become trapped within the matrix they secrete.
- 11 _____ muscle is a type of involuntary striated muscle found within the heart.
- 13 The _____ reticulum is a special type of smooth endoplasmic reticulum found in smooth and striated muscle.

- 14 A transverse or Tis a deep
- invagination of the plasma membrane found in skeletal and cardiac muscle cells which allows depolarization of the membrane to quickly penetrate to the interior of the cell.
- 15 A _____ is a type of bone cell that removes bone tissue by removing the bone's mineralized matrix.
- a series of tubes around narrow channels formed by lamellae in compact bone.
- 20 ____ is nitrogenous organic acid which naturally occurs in vertebrates and helps to supply energy to muscle and nerve cells.
- 21 A _____ is a cell originating from a mesenchymal stem cell which forms chondrocytes or cartilage cells.

 22 Also known as

bone, ______bone fills the inner cavity of long bones. It has low density and strength, but very high surface

trabecular, or spongy

- area.
 23 The inorganic mineral
 _____ makes up
 seventy percent of
- bone.

 30 Bone _____ is the process by which osteoclasts break down bone and release the minerals, resulting in a transfer of calcium from bone
- fluid to the blood.

 31 _____s are the only cells found in cartilage.
- 36 A ____ muscle doesn't need to receive impulses from a nerve to make it contract.
- 37 ______ tissue forms the rigid part of the bone organs that make up the skeletal system.
- 38 _____ is the monomeric subunit of microfilaments, one of the three major components of the

- cytoskeleton, and of thin filaments which are part of the contractile apparatus in muscle cells.
- ossification
 is the type of bone
 formation responsible
 for much of the bone
 growth in vertebrate
 skeletons, especially
 in long bones.
- 40 The muscle
 _____'s functions
 are to send
 proprioceptive
 information about the
 muscle to the central
 nervous system and
 to respond to muscle
 stretching.

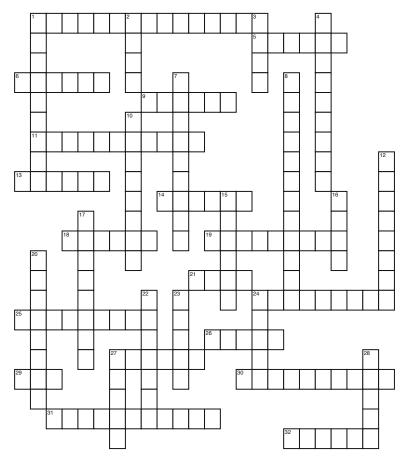
- 1 The ______ is the main or mid section shaft of a long bone.
- 2 A muscle occurs when a muscle fiber generates tension through the action of actin and myosin cross-bridge cycling.
- **5** a ______ is a sheath of connective tissue which groups

- individual muscle fibers into bundles or fascicles.
- ossification
 is the type of bone
 formation responsible
 for the development
 of flat bones,
 especially those
 found in the skull and
 clavicles.
- 7 Also known as compact bone,
 _____ bone is dense and forms the surface of bones.
- 9 A ______ is the basic unit of a muscle's cross-striated myofibril.
- 10 Bone _____ is the soft tissue found in the hollow interior of bones.
- 12 A ______ disc is an undulating double membrane separating adjacent cells in cardiac muscle fibers.
- dense connective tissue composed of collagen fibers and/or elastin fibers which can supply smooth surfaces for the movement of

- articulating bones.

 18 _____ muscle is a type of striated muscle, usually attached to the
- 19 _____ is a phosphorylated creatine molecule that is an important energy store in skeletal muscle.
- 24 As part of the regulation of muscle contraction, in resting muscle fibres, the protein _____ is displaced from its normal binding groove by troponin.
- 25 The _____ is the cell membrane of a muscle cell.
- organs that form part of the endoskeleton of vertebrates, functioning to move, support, and protect the various organs of the body, produce red and white blood cells and store minerals.
- 27 The _____ plate, or growth plate, is the cartilage plate in the long bones of children and adolescents.
- 28 A muscle _____ is a single cell of a muscle.
- 29 A motor _____ is a single alpha-motor neuron and all of the corresponding muscle fibers it innervates.
- s are a large family of motor proteins found in eukaryotic tissues which are responsible for actin-based motility.
- muscle is a type of non-striated muscle, found within the bladder, abdominal cavity, the uterus, male and female reproductive tracts, the gastrointestinal tract and elsewhere.
- 34 A _____ or sinew is a tough band of fibrous connective tissue that connects muscle to bone and is built to withstand tension.
- 35 ____ is
 contractile tissue of
 the body and is
 derived from the
 mesodermal layer of
 embryonic germ cells.

Cardiovascular System



ACROSS

- 1 A ______, also vasopressor or simply pressor, is any substance that acts to cause vasoconstriction and usually results in an increase of the blood pressure.
- 5 The right ______ is one of four chambers in the human heart, receiving deoxygenated blood from the superior and inferior vena cavae and the coronary sinus and pumping it into the right ventricle through the tricuspid valve.
- 6 The left ______ is one of the four chambers in the human heart, receiving oxygenated blood from the pulmonary veins and pumping it into the left ventricle.
- 9 The hepatic _____ vein drains blood from the digestive system and its associated glands.
- 11 _____ are the smallest of a body's blood vessels, connecting arterioles to venules
- 13 Blood s are part of the cardiovascular system and function to transport blood throughout the body, the most important types

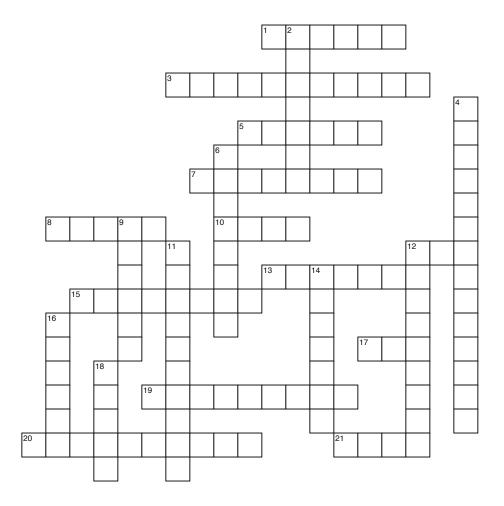
- being arteries and veins. **14** A venous system
- occurs when a capillary bed drains into another capillary bed through veins.
- 18 The _____ valve, also known as the bicuspid valve, is a dual flap valve in the heart that lies between the left atrium and the left ventricle.
- 19 The ______ valve is on the right side of the heart, between the right atrium and the right ventricle.
- 21 A _____ is a blood vessel that carries blood toward the heart.
- 24 The left _____ is one of four chambers in the human heart, receiving oxygenated blood from the left atrium via the mitral valve and pumping it into the aorta via the aortic valve.
- 25 The _____ veins carry oxygen-rich blood from the lungs to the left atrium of the heart.
- 26 Cardiac _____ is the term referring to all or any of the events related to the flow of blood that occur from the beginning of one heartbeat to the beginning of the next.
- 27 A ______ is a muscular blood vessel that carries

- blood away from the heart.
- 29 The bundle of _____ is a collection of heart muscle cells specialized for electrical conduction that transmits the electrical impulses from the AV node to the point of the apex of the fascicular branches.
- 30 _____ arteries are the arteries which are furthest from the heart.
- 31 The _____ is the thin layer of cells that line the interior surface of blood vessels.
- 32 A ______ is a small blood vessel that allows deoxygenated blood to return from the capillary beds to the larger blood vessels called veins.

- 1 The right _____ is one of four chambers in the human heart, receiving de-oxygenated blood from the right atrium via the tricuspid valve and pumping it into the pulmonary artery via the pulmonary valve.

 2 The sinoatrial _____ is one of four chambers in the human for the pulmonary valve.
- the impulse generating pacemaker tissue located in the right atrium of the heart,

- and thus the generator of sinus rhythm.
- 3 Heart ______ is a term used to describe the frequency of the cardiac cycle.
- 4 The _____ arteries carry blood from the heart to the lungs.
- 7 A ______ is a small diameter blood vessel that extends and branches out from an artery and leads to capillaries.
- 3 _____ is a medical condition in which the blood pressure is chronically elevated.
- resistance is a term used to define the resistance to flow that must be overcome to push blood through the circulatory system.
- 12 _____ fibers are specialized myocardial fibers located in the inner ventricular walls of the heart that conduct an electrical stimulus or impulse that enables the heart to contract in a coordinated fashion.
- 15 The _____ valve is one of the valves of the heart. It lies between the left ventricle and the aorta.
- 16 The atrioventricular _____ is an area of specialized tissue between the atria and the ventricles of the heart, which conducts the normal electrical impulse from the atria to the ventricles.
- 17 _____ is the period of time when the heart relaxes after contraction.
- 20 A _____ is a small blood vessel similar to a capillary but with a discontinuous endothelium.
- 22 _____ is the contraction of heart chambers, driving blood out of the chambers.
- 23 The ______ is a muscular organ responsible for pumping blood through the blood vessels by repeated, rhythmic contractions.
- 24 The heart ____s
 maintain the unidirectional
 flow of blood by opening and
 closing depending on the
 difference in pressure on
 each side.
- 27 The _____ is the largest artery in the human body.
- 28 The superior and inferior vena ______ are the veins that return de-oxygenated blood from the body into the heart, emptying into the right atrium.



- 1 The _____ blood group system refers to the five main Rh antigens (C, c, D, E and e) as well as the many other less frequent Rh antigens.
- 3 _____ is a complex process by which blood forms solid clots.
- 5 Blood ______ is the liquid component of blood, in which the blood cells are suspended
- 7 ____s or thrombocytes are the cell fragments circulating in the blood involved in the cellular mechanisms of primary hemostasis leading to the formation of blood clots.
- 8 Chloride ______ is a process which occurs in a cardiovascular system and refers to the exchange of bicarbonate and chloride across the membrane of red blood cells.
- 10 The ______ effect states that in the presence of carbon dioxide, the oxygen affinity of respiratory pigments such as hemoglobin decreases.
- 12 The ______ blood group system is the most important blood type system (or blood group system) in human blood transfusion.
- 13 _____ is an important enzyme present in blood that degrades many blood plasma proteins, most notably

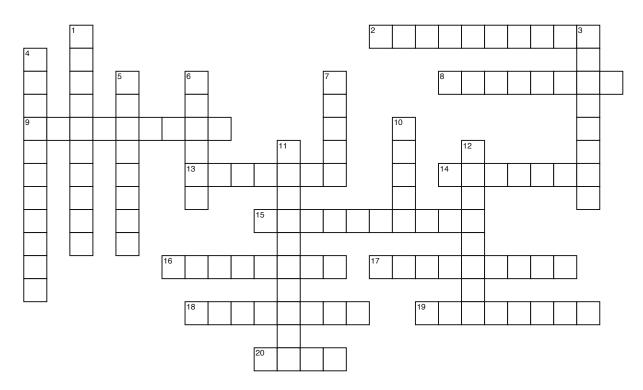
fibrin clots.

- 5 ______ is a serine protease that converts soluble fibrinogen into insoluble strands of fibrin, as well as catalyzing many other coagulation-related reactions.
- 17 ______ blood cells are the most common type of blood cell and the body's principal means of delivering oxygen from the lungs to body tissues via the blood.
- 19 A ______ is a heterocyclic macrocycle derived from four pyrrole-like subunits interconnected via their alpha carbon atoms.
- 20 ______ is the name of a family of hereditary genetic illnesses that impair the body's ability to control coagulation.
- 21 A ______ is a prosthetic group that consists of an iron atom contained in the center of a large heterocyclic organic ring called a porphyrin.

- 2 The ______ effect is a property of hemoglobin that leads to deoxygenated blood's increased ability to carry carbon dioxide.
- 4 _____ is the process by which red blood cells are produced.
- 6 _____ is one of the two types of

- serum proteins, the other being albumin. This term encompasses a heterogeneous series of families of proteins.
- 9 _____ is a protein involved in the clotting of blood which is polymerised to form a mesh that forms a hemostatic plug or clot over a wound site.
- 11 ______ is the iron-containing oxygentransport metalloprotein in red blood cells.
- 12 Carbonic ______ is a family of metalloenzymes that catalyze the rapid conversion of carbon dioxide to bicarbonate and protons.
- 14 Human serum _____ is the most abundant protein in human blood plasma.
- 16 _____-cell disease is a group of genetic disorders caused by an abnormal form of hemoglobin.
- 18 _____ is a specialized biological fluid consisting of erythrocytes, leukocytes, and thrombocytes suspended in a complex fluid medium known as plasma.

Respiratory System



ACROSS

2	Pulmonary	is a
	surface-active I	ipoprotein
	complex formed	d by type II
	alveolar cells w	hich reduces
	surface tension	in the air water
	interface within	alveoli.
8	The	bronchi arise
	from the second	dary bronchi.

- 9 _____ circulation is the portion of the cardiovascular system which carries oxygendepleted blood away from the heart, to the lungs, and returns oxygenated blood back to the heart.
- 13 A ______ is one of the two channels of the nose, from the point where they bifurcate to the external opening.
- 14 The ______, or windpipe, is a tube that extends from the larynx to the primary bronchi.
- 15 ______, also known as inspiration, is the movement of air from the external environment, through the airways, into the alveoli during breathing.
- 16 The left main _____ is smaller in caliber but longer than the right, entering the root

of the left lung opposite the
sixth thoracic vertebra.

- 17 The ______ is a sheet of muscle extending across the bottom of the ribcage which separates the thoracic cavity from the abdominal cavity and performs an important function in respiration.
- 18 _____ ducts are the tiny end tubules of the branching airways that fill the lungs.
- 19 Functional _____ capacity is the volume of air present in the lungs at the end of passive expiration.
- 20 _____ space is air that is inhaled by the body in breathing, but does not partake in gas exchange.

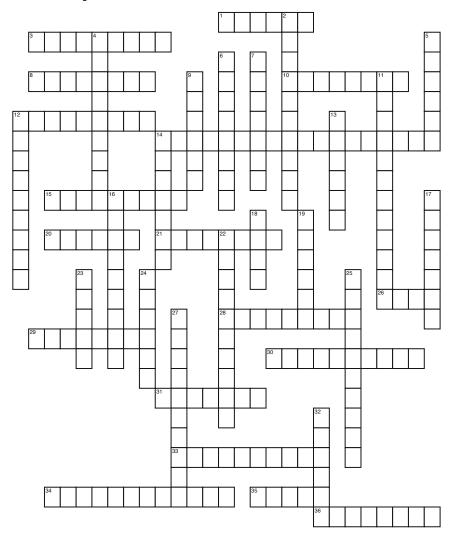
DOWN

- 1 ______, or expiration, is the movement of air out of the bronchial tubes, through the airways, to the external environment during breathing.
- **3** A ______ bronchiole is a bronchiole at the end of the conducting zone.
- **4** The _____ system consists of the airways, the lungs, and

the muscles that mediate the movement of air into and out of the body.

- **5** A ______ is a caliber of airway in the respiratory tract that conducts air into the lungs.
- 6 The ______, colloquially known as the voicebox, is an organ in the neck of mammals involved in protection of the trachea and sound production.
- 7 _____ capacity is the maximum volume of air that a person can exhale after maximum inhalation.
- 10 In humans the respiratory
 _____ is the part of the
 anatomy that has to do with the
 process of respiration.
- 11 The primary _____s arise from the tertiary bronchi.
- 12 The right main ______ is wider, shorter, and more vertical in direction than the left, entering the right lung nearly opposite the fifth thoracic vertebra.

Immunity



ACROSS

- The _____ system is a collection of mechanisms within the body that protects against disease by identifying and killing pathogens and tumor cells.
- is a biogenic amine involved in local immune responses as well as regulating physiological function in the gut and acting as a neurotransmitter.
- The ______ immune response provides the vertebrate immune system with the ability to recognize and remember specific pathogens.
- 10 The _____ duct is the largest lymphatic vessel in the body, collecting most of the lymph in the body and draining into the systemic circulation at the left subclavian vein
- 12 A ______ is a cell that ingests and destroys foreign matter such as microorganisms or debris.
- 14 The major _____ complex (MHC) is a large gene family found in most vertebrates, the most genedense region of the mammalian genome, which plays an important role in the immune system, autoimmunity, and reproductive success.
- 15 A _____ is the detrimental

- colonization of a host organism by a foreign species.
- 20 The ____ immune system comprises the cells and mechanisms that defend the host from infection by other organisms, in a non-specific manner.
- 21 A _____ is a nonparasitic antigen capable of stimulating a type I hypersensitivity reaction in atopic individuals.
- 26 Lymph _____s are components of the lymphatic system that contain white blood cells and act as filters or traps for foreign particles.
- 28 The ______system is a complex network of lymphoid organs, lymph nodes, lymph ducts, tissues, lymph capillaries and lymph vessels that produce and transport lymph fluid from tissues to the circulatory system.
- 29 ____s are a family of small, signaling proteins and glycoproteins particularly important in both innate and adaptive immune responses.
- 30 _____ or immunoglobulins are proteins used by the immune system to identify and neutralize foreign objects, such as bacteria and viruses.
- 31 _____ is an abnormal reaction to a substance foreign to the body that is acquired, predictable and

rapid.

- 33 _____ granulocytes are white blood cells of the immune system that are responsible for combating infection by parasites in vertebrates
- 34 Antigen _____ is a process in the body's immune system by which macrophages, dendritic cells and other cell types capture antigens and then show them for recognition by T-cells.
- 35 _____ blood cells or leukocytes are cells of the immune system which defend the body against both infectious disease and foreign materials.
- 36 The T cell is a molecule found on the surface of T lymphocytes that is generally responsible for recognizing antigens bound to major histocompatibility complex molecules.

DOWN

- s are the most abundant type of white blood cells, phagocytes that quickly congregate at a focus of infection, attracted by cytokines expressed by activated endothelium, mast cells and macrophages.
- 4 _____ signalling is a form of hormonal signalling in which a cell

Answer key - pg 125

- secretes a hormone, or chemical messenger, that binds to receptors on the same cell.

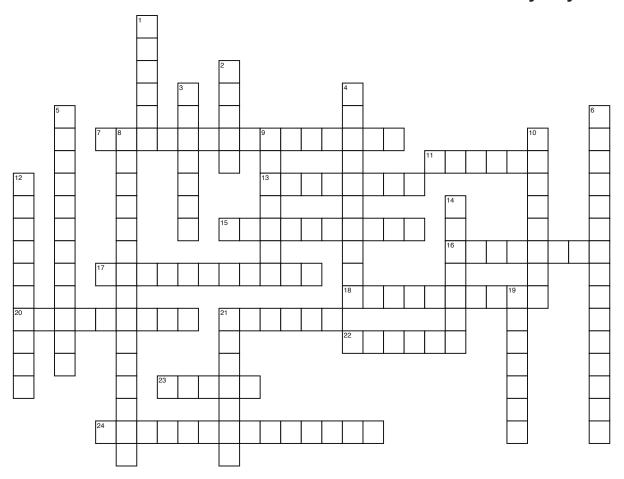
 Becalls are a Becall subtype that are formed following primary infection. They remember
- infections

 6 A ______ or infectious agent is a biological agent that causes disease or illness to its host.

the same pathogen for faster antibody production in future

- 7 A ______ is the part of a macromolecule that is recognized by the immune system, specifically by antibodies, B cells, or T cells.
- **9** Lymph _____s are thin walled, valved structures that carry lymph.
- 11 _____ is the complex biological response of vascular tissues to harmful stimuli, such as pathogens, damaged cells, or irritants.
- 12 _____ signaling is a form of cell signaling in which the target cell is close to the signal releasing cell.
- 13 Natural ______ cells are a form of cytotoxic lymphocyte which constitute a major component of the innate immune system.
- 14 The _____ immune response is the aspect of immunity that is mediated by antibodies, produced by the cells of the B lymphocyte lineage.
- 16 A ______ T cell belongs to a subgroup of T lymphocytes which are capable of inducing the death of infected somatic or tumor cells.
- 17 A ______ or immunogen is a molecule that stimulates an immune response.
- 19 _____ cells are cells of the immune system that secrete large amounts of antibodies.
- 22 ____ T cells, sometimes known as suppressor T cells, act to suppress activation of the immune system and thereby maintain immune system homeostasis and tolerance to self-antigens.
- 23 A _____ is an abnormal growth or mass of tissue.
- 24 The ____ is an organ located in the abdomen of the human body, where it carries out immune functions as well as functions concerned with the destruction of old red blood cells.
- 25 Monocytes and _____s are phagocytes, acting in both innate immunity as well as cell-mediated immunity of vertebrate animals.
- 27 The _______ system is a biochemical cascade which helps clear pathogens from an organism consisting of a number of small proteins found in the blood, normally circulating as inactive zymogens.
- 32 With no cytotoxic or phagocytic activity, T cells are a sub-group of lymphocytes playing an important role in establishing and maximizing the capabilities of the immune system.

Urinary System



ACROSS

- 7 The ______ apparatus is a microscopic structure in the kidney which regulates the function of each nephron.
- 11 The _____s are organs that filter wastes, such as urea, from the blood and excrete them, along with water, as urine.
- 13 _____ is increased production of urine by the kidney.
- 15 The glomerular _____ rate is the volume of fluid filtered from the renal glomerular capillaries into the Bowman's capsule per unit time.
- 16 The ______ tubule is the portion of the duct system of the nephron leading from Bowman's capsule to the loop of Henle.
- 17 The _____ buffering system is the most important buffer solution for maintaining a relatively constant pH in the plasma.
- 18 A ____ is a capillary tuft surrounded by Bowman's capsule in nephrons of the vertebrate kidney which

- receives its blood supply from an afferent arteriole of the renal circulation.
- 20 _____, known by physiologists as micturition ot voiding, is the process of disposing urine from the urinary bladder through the urethra.
- 21 The _____s are the ducts that carry urine from the kidneys to the urinary bladder.
- 22 _______'s capsule is a cup like sac at the beginning of the tubular component of a nephron in the kidney. A glomerulus is enclosed in the sac.
- 23 _____ is a liquid produced through the kidney, and is collected in the bladder and excreted through the urethra.
- 24 ______ exchange is a mechanism used to transfer some property of a fluid to another across a semipermeable membrane.

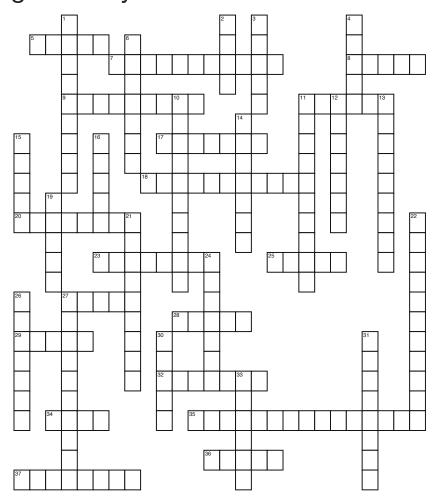
DOWN

1 The renal _____ is the outer portion of the kidney

- between the renal capsule and the renal medulla.
- 2 In the kidney, the loop of _____ is the portion of the nephron that leads from the proximal convoluted tubule to the distal convoluted tubule. The main function of this structure is to reabsorb water and ions from the urine.
- 3 The urinary ______ is a hollow, muscular, and distensible organ that collects urine excreted by the kidneys prior to disposal by urination.
- 4 The ______ duct system of the kidney consists of a series of tubules and ducts that connect the nephrons to the ureter.
- 5 Renal _____ is a mechanism by which the kidneys can regulate the plasma pH.
- 6 The _____ cells are cells that synthesize, store, and secrete the enzyme renin.
- 8 _____ occurs at the barrier between the blood and the filtrate in the renal corpuscle or Bowman's capsule in the kidneys.

- 9 The renal _____ is the innermost part of the kidney.
- 10 Renal _____ are coneshaped tissues of the kidney within the renal medulla, which is made up of 8 to 18 of these conical subdivisions.
- 12 The distal ______ tubule is a portion of kidney nephron between the loop of Henle and the collecting duct system.
- **14** A _____ is the basic structural and functional unit of the kidney.
- 19 The ______ is a tube which connects the urinary bladder to the outside of the body.
- 21 The ______ system is the organ system that produces, stores, and eliminates urine. In humans it includes two kidneys, two ureters, the bladder, and the urethra.

Digestive System



ACROSS

- 5 Gastric ______ is a strong acidic liquid, pH 1 to 3 in humans, released by gastric glands in the lining of the stomach. Its main components are digestive enzymes pepsin and rennin, hydrochloric acid, and mucus.
- 7 _____ is the rhythmic contraction of smooth muscles to propel contents through the digestive tract.
- 8 The _____ intestine, commonly referred to by its Greek name, the colon, is the last part of the digestive system: the final stage of the alimentary canal in vertebrate animals.
- 11 ______ is a digestive protease released by the chief cells in the stomach that functions to degrade food proteins into peptides.
- 17 _____ acid is one of the main secretions of the stomach, an acid solution consisting mainly of hydrochloric acid, and small quantities of potassium chloride and sodium chloride.
- **18** The ______, or cholecyst, is a pear-shaped organ that stores

- about 50 ml of bile until the body needs it for digestion.
- 20 _____ glands are glands that secrete their products into ducts.
- 23 The ______ is a hollow jointed tube about 25-30 cm long connecting the stomach to the jejunum.
- 25 Pancreatic _____ is produced by the pancreas, containing a variety of enzymes including trypsinogen, chymotrypsinogen, elastase, carboxypeptidase, pancreatic lipase, and amylase.
- 27 The ______, also known as the buccal cavity or the oral cavity, is the orifice through which an organism takes in food and water.
- 28 The _____ is the final section of the small intestine.
- 29 The gastrointestinal _____ is the system of organs within multicellular animals that takes in food, digests it to extract energy and nutrients, and expels the remaining waste.
- 32 _____ is the name given to glycoside hydrolase enzymes that break down starch into glucose molecules.
- 34 Released from the gall bladder,
 _____acids serve multiple
 functions including elimination of
 cholesterol from the body,
 elimination of catabolites from the
 liver, and emulsifying lipids and fat
 soluble vitamins in the intestine.

- 35 _____ is a peptide hormone of the gastrointestinal system responsible for stimulating the digestion of fat and protein.
- 36 In addition to being the largest gland in the human body, producing bile for example, the ______ plays a major role in metabolism.
- 27 _____ cells are the stomach epithelium cells which secrete gastric acid and intrinsic factor.

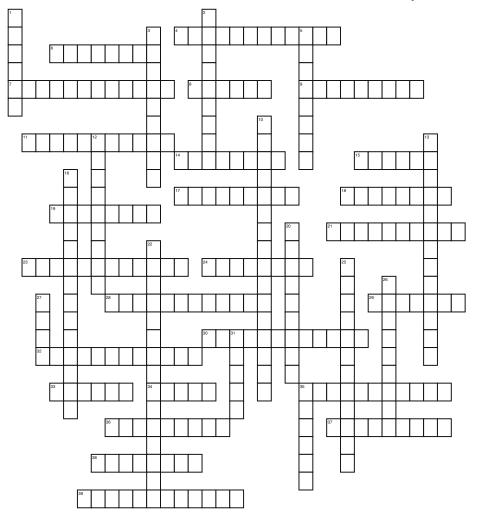
DOWN

- 1 _____ is the process of metabolism whereby a biological entity processes a substance in order to chemically and mechanically convert the substance for the body to use.
- 2 _____ is a bitter, yellow or green alkaline fluid secreted by hepatocytes from the liver with components including cholesterol, lecithin, bilirubin, biliverdin, sodium glycocholate and sodium taurocholate.
- **3** A gastric _____ cell is a cell in the stomach that releases pepsinogen and rennin.
- 4 A ______ is a round mass of organic matter moving through the digestive tract.
- 6 The ______ is the central of the three divisions of the small intestine and lies between the duodenum and the ileum.

Answer key - pg 127

- _____, known scientifically as deglutition, is the reflex in the human body that makes something pass from the mouth, to the pharynx, into the esophagus, with the shutting of the epiglottis.
- 11 The _____ duct is a duct joining the pancreas to the common bile
- 12 The ______ is the part of the neck and throat situated immediately posterior to the mouth and nasal cavity, and cranial, or superior, to the esophagus, larynx, and trachea.
- 13 _____ is a science that examines the relationship between diet and health.
- 14 A ______ is a nutrient that is an organic compound required in tiny amounts for essential metabolic reactions in a living organism.
- 15 _____ is the liquid substance found in the stomach before passing through the pyloric valve and entering the duodenum.
- 16 Intestinal _____ are tiny, fingerlike structures that protrude from the wall of the intestine.
- 19 The _____ is another name for the large intestine.
- 21 The ____ or gullet is an organ in vertebrates which consists of a muscular tube through which food passes from the pharynx to the stomach.
- 22 _____ or chewing is the process by which food is mashed and crushed by teeth.
- 24 Dietary ____s are the chemical elements required by living organisms, other than the four elements carbon, hydrogen, nitrogen, and oxygen which are present in common organic molecules.
- 26 A ______ is an essential human nutrient. In foods of animal origin, the major form of this vitamin is retinol, but all forms have a betaionone ring to which an isoprenoid chain is attached.
- 27 The _____ are structures that increase the surface area of cells by approximately 600 fold, thus facilitating absorption and secretion.
- 30 The _____ intestine is where the majority of digestion takes place, the part of the gastrointestinal tract (gut) between the stomach and the large intestine which includes the duodenum, jejunum, and ileum.
- 31 A _______ is a substance used in an organism's metabolism or physiology which must be taken in from the environment.
- 33 ______ is the watery and usually frothy substance produced in the mouths of humans and some animals.

Reproductive System



ACROSS

- 4 A ______ is the haploid cell that is the male gamete. It joins an ovum to form a zygote.
- 6 Ovarian s are the basic unit of female reproductive biology, roughly spherical aggregations of cells found in the ovary.
- 7 _____ is an event that occurs early in pregnancy in which the embryo adheres to the wall of uterus.
- 8 The _____ or womb is the major female reproductive organ. One end, the cervix, opens into the vagina; the other is connected on both sides to the fallopian tubes.
- 9 _____ is the process by which a mature ovarian follicle ruptures and discharges an ovum.
- 11 The _____ is the inner membrane of the uterus.
- hormone is secreted by the anterior pituitary gland to stimulate the growth of Graafian follicles in women and enhance the

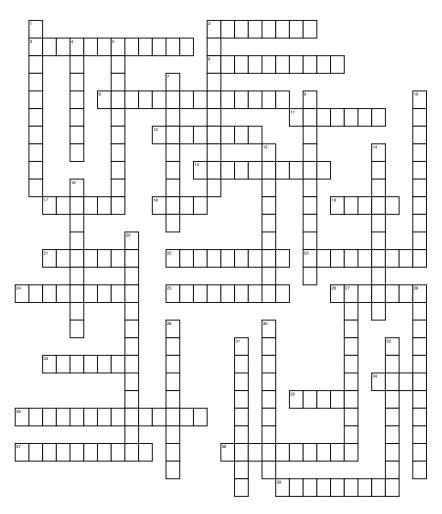
- production of androgenbinding protein by Sertoli cells in men.
- 15 A ______ is a female gametocyte or germ cell involved in reproduction, in other words, an immature ovum.
- 17 The zona _____ is a glycoprotein membrane surrounding the plasma membrane of an oocyte.
- 18 The ____ is an ephemeral organ present during pregnancy, which develops from the same sperm and egg cells that form the fetus, and functions as a fetomaternal organ for filtration and transfer.
- 19 _____ refers to the first menstrual period, or first menstrual bleeding.
- 21 The _____phase, or proliferative phase, is the phase of the menstrual cycle during which follicles in the ovary mature to climax with ovulation.

- luteum and thereby maintaining progesterone levels for pregnancy.
- et m is the generic term for any natural or synthetic compound, usually a steroid hormone, that stimulates or controls the development and maintenance of masculine characteristics in vertebrates.
- 28 A ______ is a male gametocyte which is derived from a spermatogonium.
- 29 A ______ is a kind of germ cell that results from the meiosis of a gametocyte.
- 30 _____ is a steroid hormone involved in the female menstrual cycle, pregnancy and embryogenesis.
- 32 _____ is a phase of the menstrual cycle in which the endometrium is shed.
- 33 The corpus _____ is a temporary endocrine structure in mammals, involved in the production of the progestogens which are needed for the maintenance of a pregnancy.

- 4 A _____ is an organ that makes gametes.
- synthesized and secreted by the anterior lobe of the pituitary gland. In the female, an acute rise of LH triggers ovulation. In the male, LH stimulates Leydig cell production of testosterone.
- **36** A _____ cell is a somatic cell found closely associated with the developing oocyte within the ovary.
- 37 ______ is the creation of an ovum, the female process of gametogenesis.
- organelle that develops over the anterior half of the spermatozoon's head which contains digestive enzymes including hyaluronidase and acrosin.
- male sex hormone and an anabolic steroid. It is rimarily secreted in the testes of males and the ovaries of females, although small amounts are also secreted by the adrenal glands.

DOWN

- Found adjacent to the seminiferous tubules in the testicle, _______ cells can secrete testosterone and are often closely related to nerves.
- The _____ cycle is a recurring cycle of physiologic changes under endocrine control necessary for reproduction that occurs in females.
- 3 The word _____ literally means the permanent physiological, or natural, cessation of menstrual cycles.
- 5 A ______ is an immature ovum. It is a female gametogonium.
- 10 _____ is the maturation of the ovarian follicle, a densely-packed shell of somatic cells that contains an immature occyte.
- 12 Labelled the female hormone but also present in males, _____ is the major estrogen in humans, not only of critical importance for reproductive and sexual functioning, but also affects other organs.
- 13 Fertilization is fusion of gametes to form a new organism of the same species.
- 16 A _____ is an intermediary male gametogonium in the production of spermatozoa.
- 20 The term _____ refers to the haploid male gametid that results from division of secondary spermatocytes.
- 22 _____ is the process by which male spermatogonia develop into mature spermatozoa.
- 25 _____ tubules are located in the testicles, and are the specific location of meiosis, and the subsequent creation of gametes, namely spermatozoa.
- 26 _____ describes the secretion of milk from the mammary glands, the process of providing that milk to the young, and the period of time that a mother lactates to feed her young.
- 27 A ______ is a haploid female reproductive cell or gamete.
- 31 A _____ is the eggproducing reproductive organ found in female organisms
- 35 The _______phase, or secretory phase, is the latter phase of the menstrual cycle, which begins with the formation of the corpus luteum and ends in either pregnancy or luteolysis.



ACROSS

- 2 _____evolution is the independent evolution of similar traits, starting from a similar ancestral condition due to similar environments or other evolutionary pressures.
- 3 A _____ tree, also called an evolutionary tree, is a tree showing the evolutionary relationships among various biological species or other entities that are believed to have a common ancestor.
- 6 _____ speciation is a form of speciation in which new species are formed in isolated peripheral populations.
- 8 _____ is the occurrence of small-scale changes in allele frequencies in a population, over a few generations, also known as change at or below the species level.
- 11 ______, also known as mimetism, describes a situation where one organism, the mimic, has evolved to share common outward characteristics with another organism, the model, through the selective action of a signal-receiver or dupe.
- 12 _____ selection is the selective removal of alleles that are deleterious.
- 15 A population ______ is an evolutionary event in which a significant percentage of a population or species is killed or

- otherwise prevented from reproducing, and the population is reduced by 50% or more.
- 17 Charles Robert _____ (1809 1882) was an English naturalist who proposed and provided scientific evidence that all species of life have evolved over time from one or a few common ancestors through the process of natural selection.
- 18 ____ is the genus that includes modern humans and their close relatives.
- 19 A ______ is a taxonomic group of organisms comprising a single common ancestor and all the descendants of that ancestor.
- 21 The ____ effect is defined as the effect of establishing a new population by a small number of individuals, carrying only a small fraction of the original population's genetic variation.
- 22 _____ Speciation is the genetic divergence of multiple populations inhabiting the same geographic region from a single parent species, such that those populations become different species.
- 23 Heterozygote ______ describes the case in which the heterozygote genotype has a higher relative fitness than either the homozygote dominant or homozygote recessive genotype.
- 24 ______ selection refers to forms of natural selection which work to maintain genetic polymorphisms within a population.

- 25 _____ evolution occurs when two or more biological characteristics have a common evolutionary origin but have diverged over evolutionary time.
- 26 A _____ is one of the basic units of biological classification, often defined as a group of organisms capable of interbreeding and producing fertile offspring.
- **33** A group of organisms is said to have common ______ if they have a common ancestor.
- 34 A gene _____ is the complete set of unique alleles in a species or population.
- 35 Genetic is the statistical effect that results from the influence that chance has on the survival of alleles, which may cause an allele, and the biological traits that it confers, to become more common or rare over successive generations.
- 36 The _____ principle states that the occurrence of a genotype, perhaps one associated with a disease, stays constant unless matings are non-random or inappropriate, or mutations accumulate.
- avolution is the process whereby organisms not closely related, not monophyletic, independently evolve similar traits as a result of having to adapt to similar environments or ecological niches.
- 38 _____ speciation, also known as

geographic speciation, is the phenomenon where large biological populations are physically isolated by an extrinsic barrier and evolve intrinsic reproductive isolation.

39 Adaptive ______ describes the rapid speciation of a single or a few species to fill many ecological niches.

DOWN

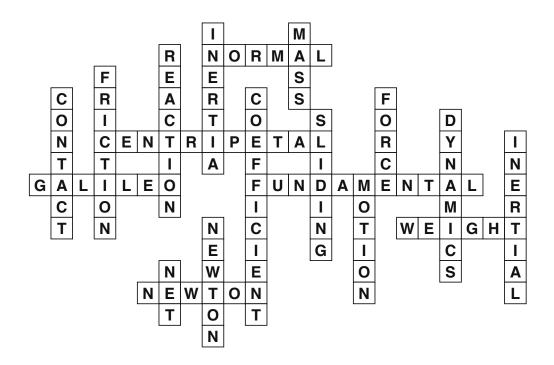
- 1 _____ is the evolutionary process by which new biological species arise.
- genetics is the study of allele frequency distribution and change under the influence of the four evolutionary forces: natural selection, genetic drift, mutation and gene flow.
- 5 _____ is the cessation of existence of a species or group of taxa, reducing biodiversity.
- 7 Under _____, individuals with advantageous or adaptive traits tend to be more successful than their peers reproductively.
- 9 _____ selection occurs when natural selection favors a single allele and therefore allele frequency continuously shifts in one direction.
- 10 ______ selection is a descriptive term used to describe changes in population genetics that simultaneously favor individuals at both extremes of the distribution.
- 13 Natural _____ is the process by which favorable traits that are heritable become more common in successive generations of a population of reproducing organisms, and unfavorable traits less common.
- is a philosophy of classification that arranges organisms only by their order of branching in an evolutionary tree and not by their morphological similarity.
- 16 Allele _____ is a measure of the relative proportion of an allele on a genetic locus.
- 20 Habitat ______ describes the emergence of discontinuities in an organism's preferred environment.
- 27 speciation is a form of speciation that occurs due to variations in mating frequency of a population within a continuous geographical area.
- 28 ______ selection, also referred to as purifying selection, is a type of natural selection in which genetic diversity decreases as the population stabilizes on a particular trait value.
- 29 The genotype _____ is the proportion of genotypes in a population.
- 30 ______ is the change in the inherited traits of a population from generation to generation.
- 31 ______ evolution is the process of evolution at the scale of DNA, RNA, and proteins.
- 32 Reproductive ______ is a category of mechanisms that prevent two or more populations from exchanging genes.

Puzzle Keys Beyond This Page

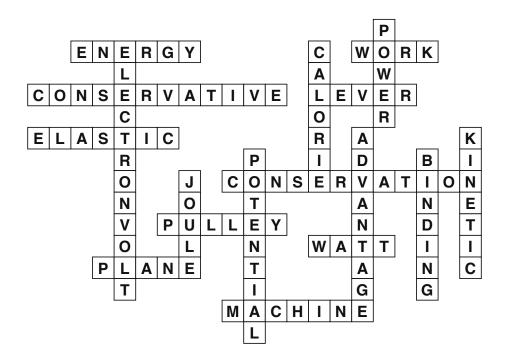
Kinematics



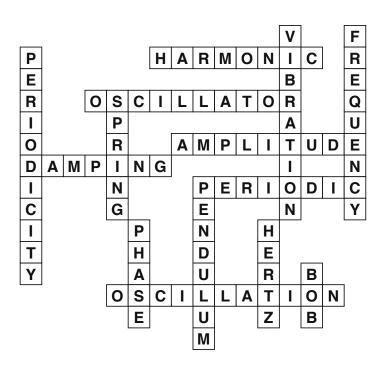
Newton's Laws



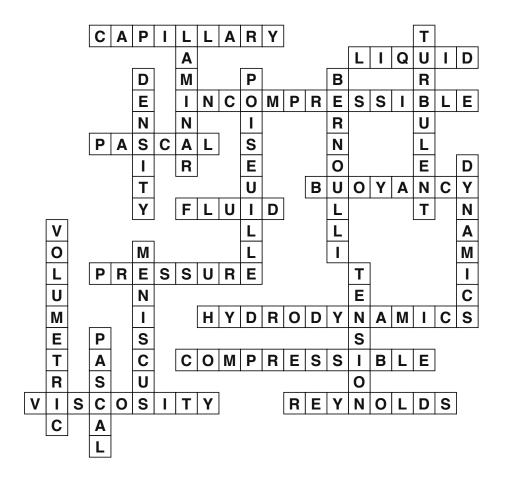
Work & Energy



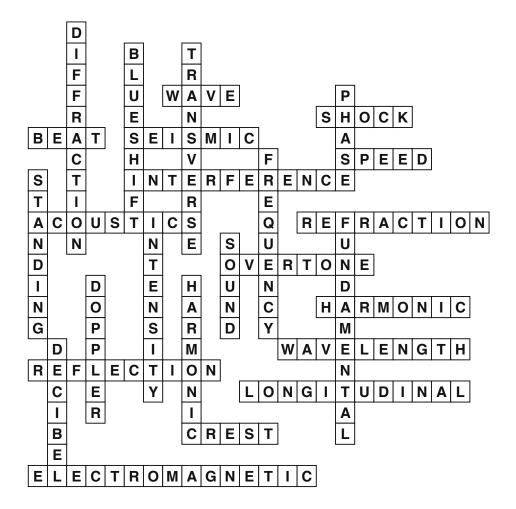
Harmonic Motion



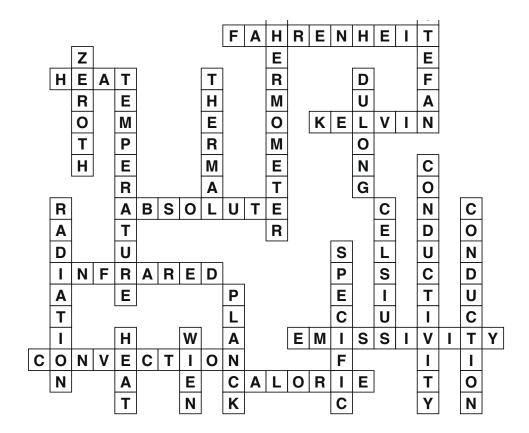
Fluid Mechanics



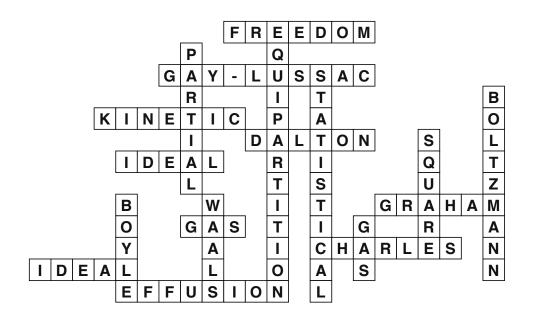
Waves



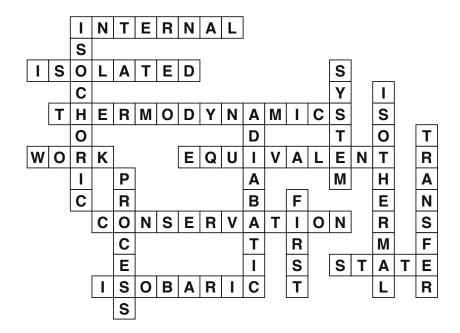
Temperature and Heat Flow



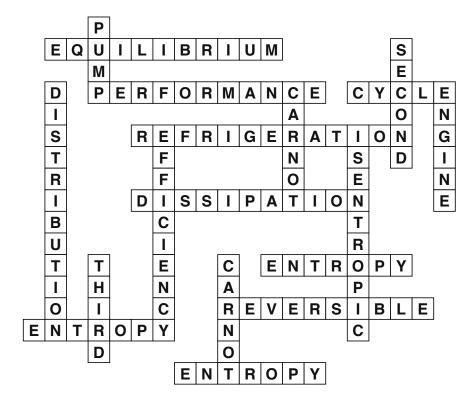
Ideal Gas



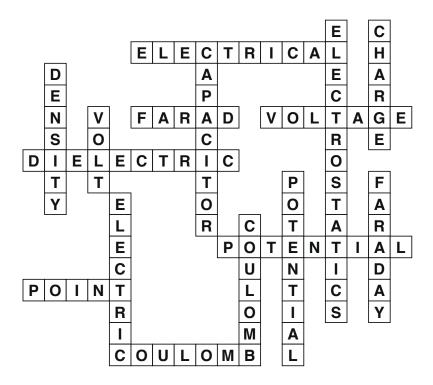
1st Law of Thermodynamics



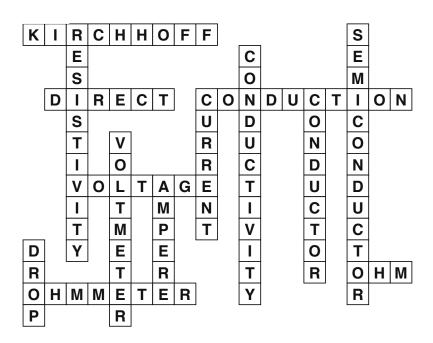
2nd Law of Thermodynamics



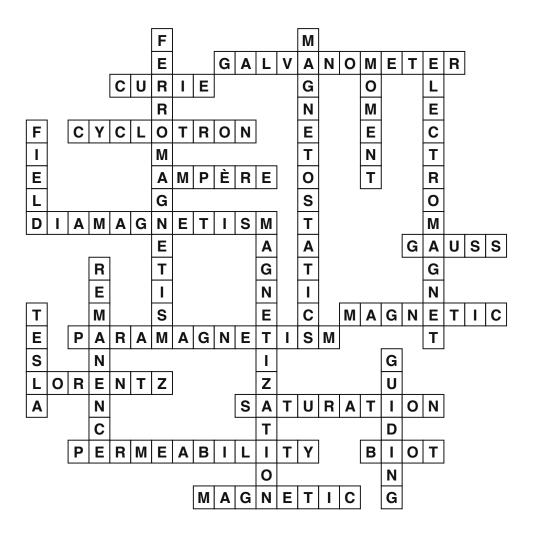
Electricity



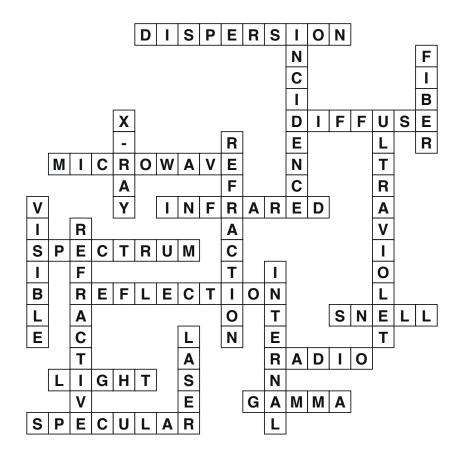
DC Current



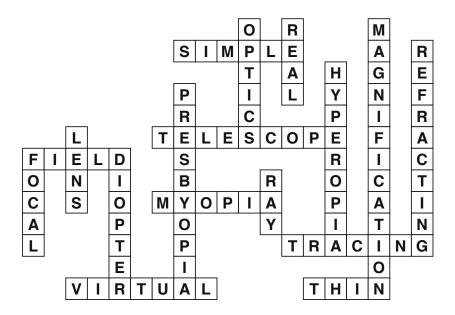
Magnetism



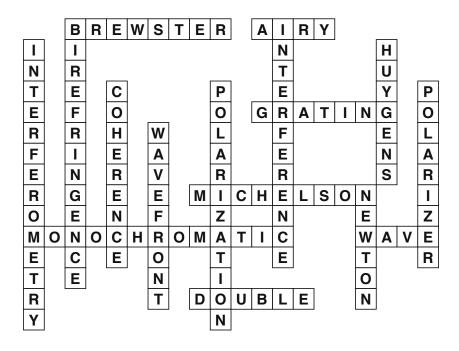
Properties of Light

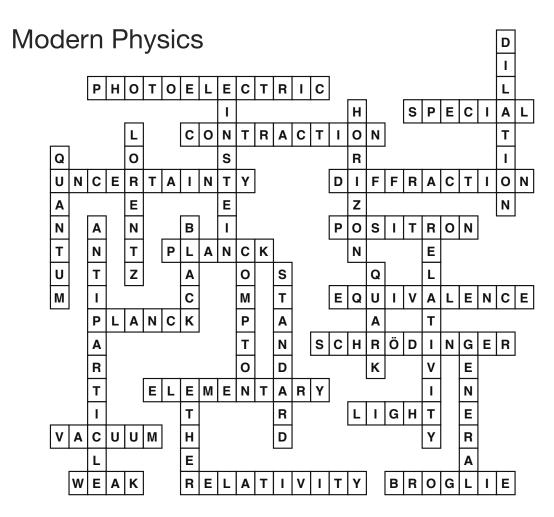


Geometric Optics

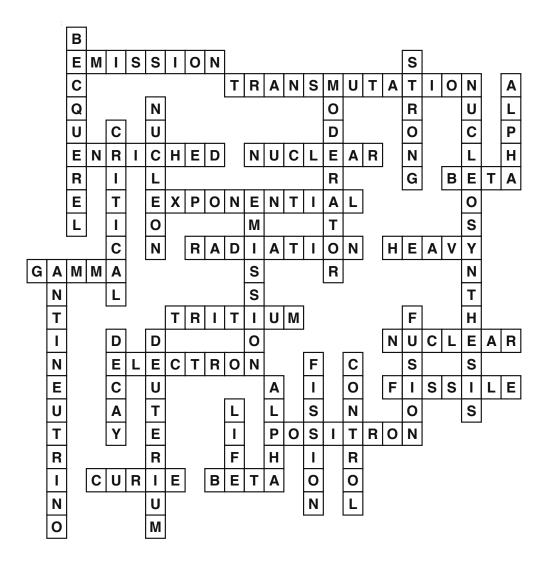


Wave Optics

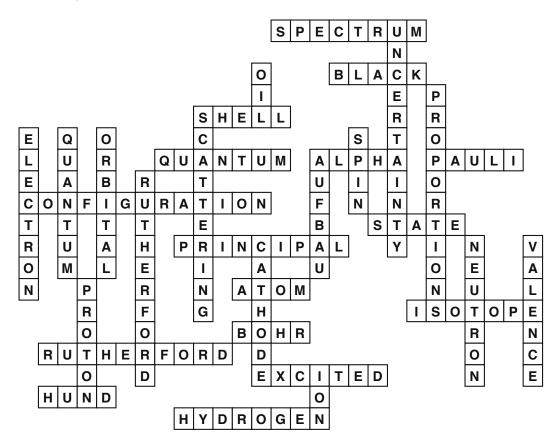




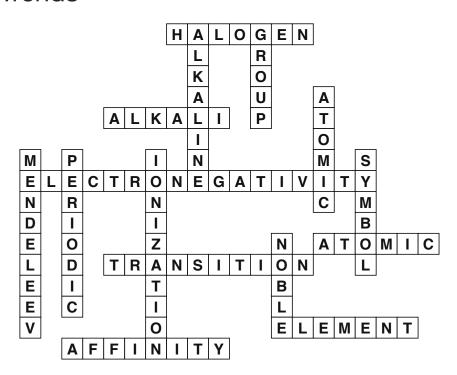
Nuclear Physics



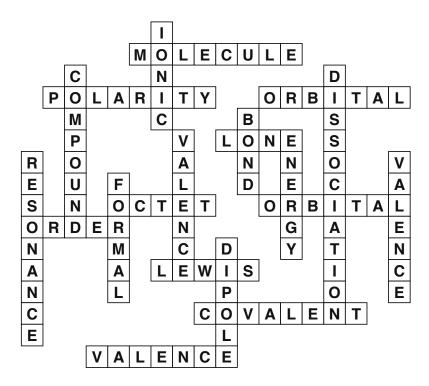
Atomic Theory



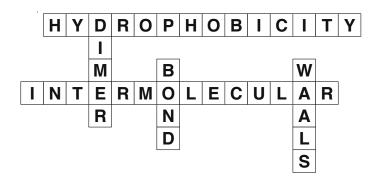
Periodic Trends



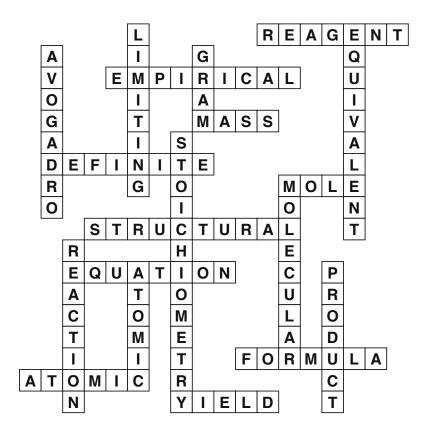
Chemical Bonding



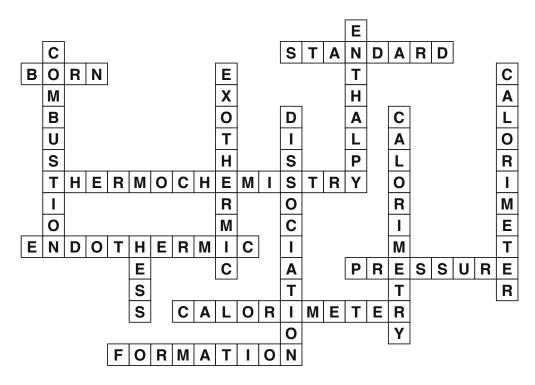
Intermolecular Force



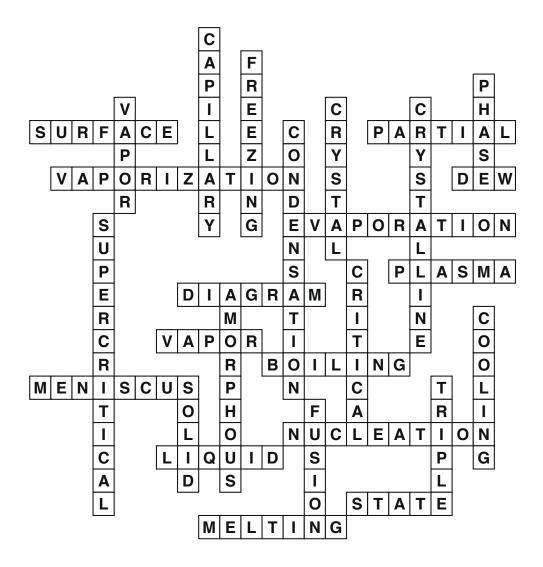
Stoichiometry



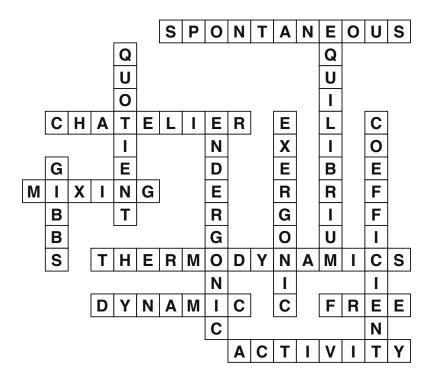
Thermochemistry



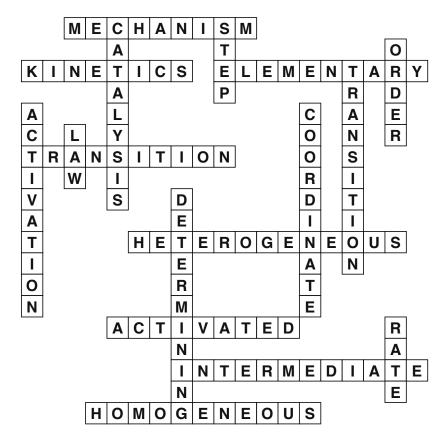
States of Matter



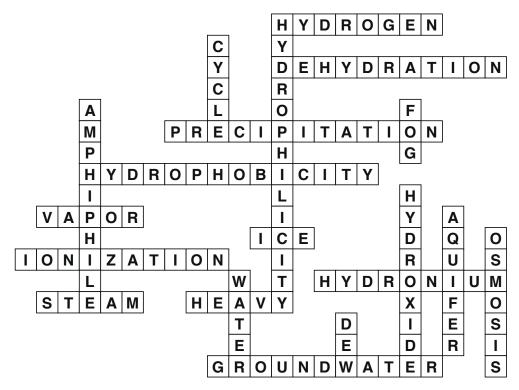
Chemical Thermodynamics



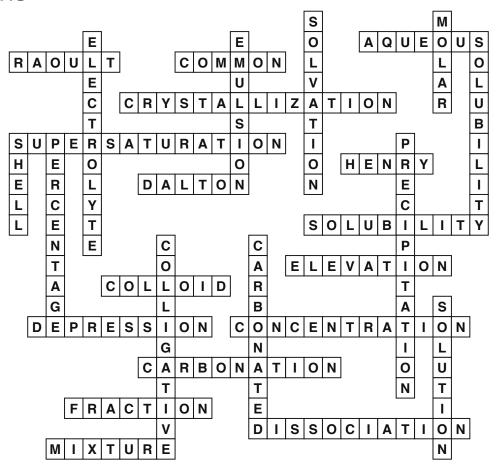
Chemical Kinetics



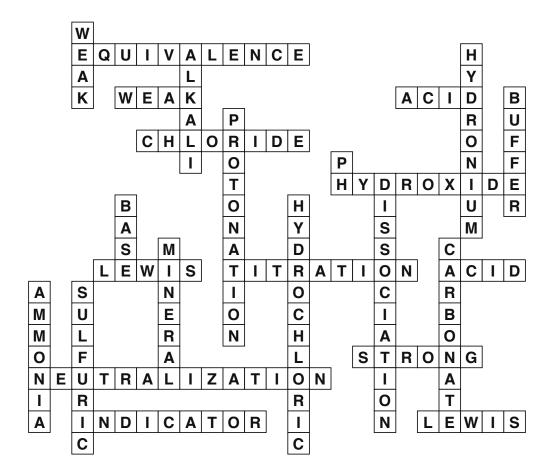
Water



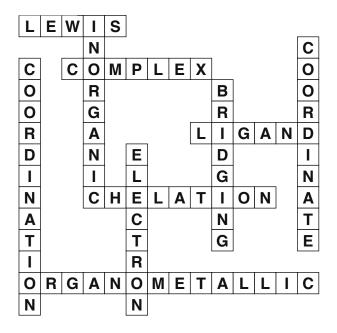
Solutions



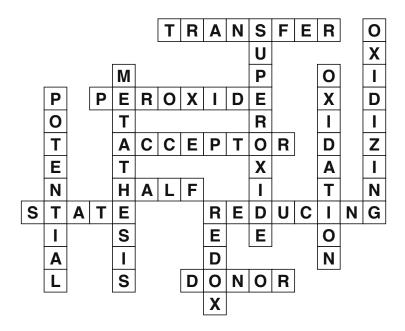
Acids and Bases



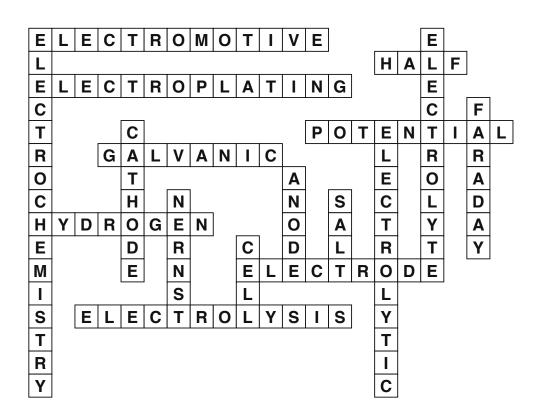
Coordination Chemistry



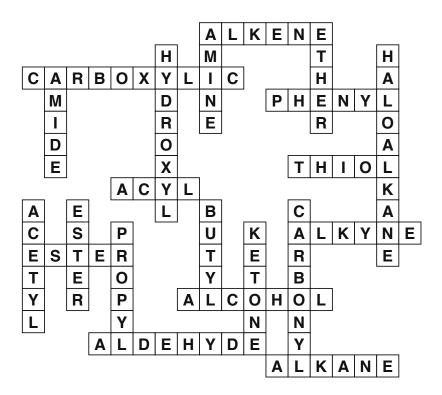
Oxidation-Reduction



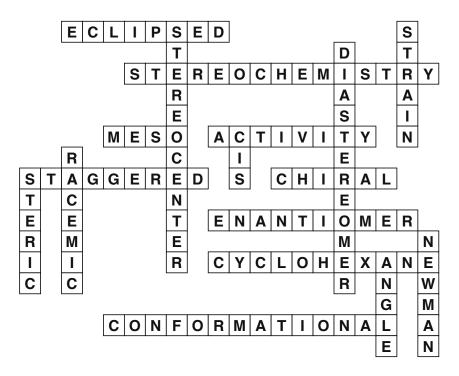
Electrochemistry



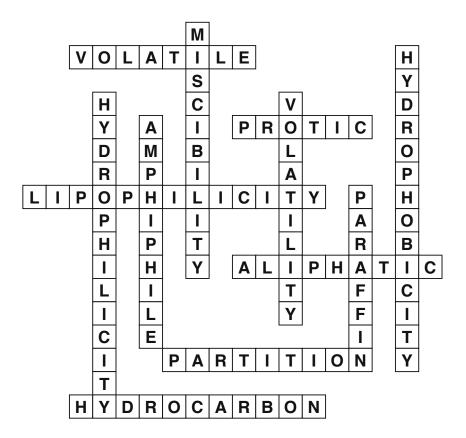
Organic Functional Groups



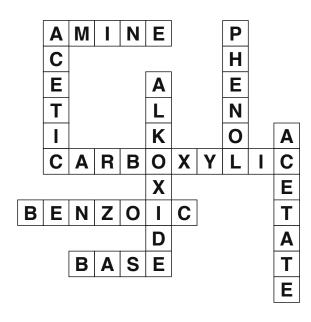
Stereochemistry



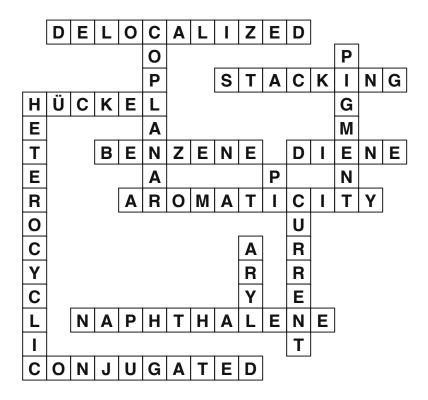
Physical Properties



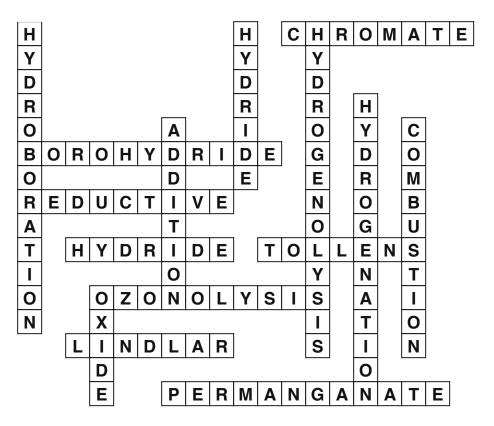
Organic Acids & Bases



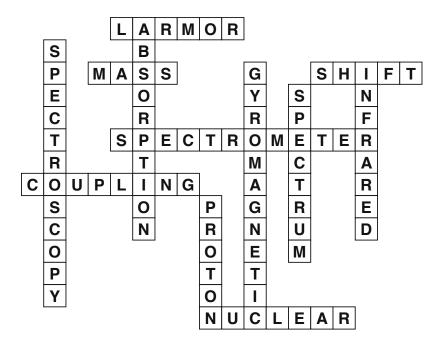
Aromatic Compounds



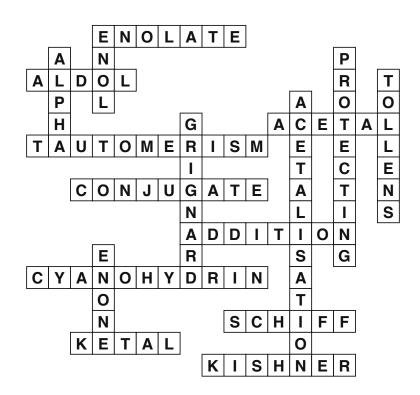
Organic Redox



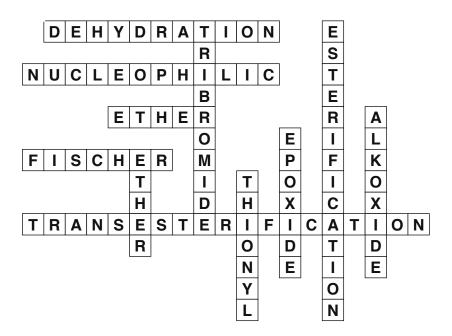
Molecular Spectroscopy



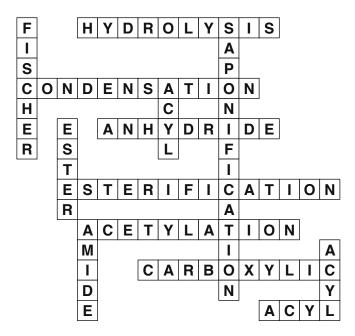
Aldehydes & Ketones



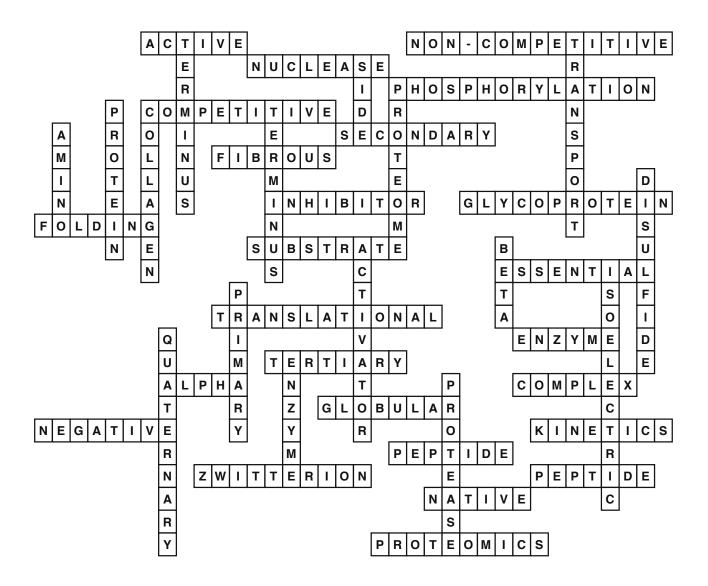
Alcohols



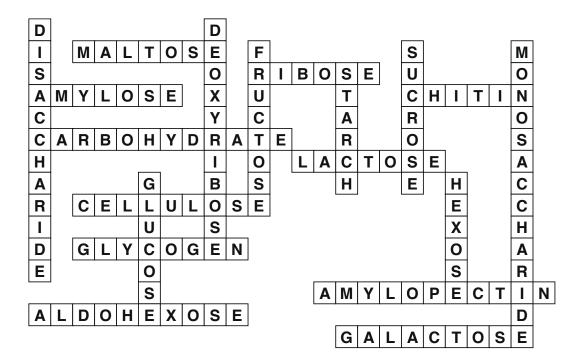
Carboxylic Acid Derivatives



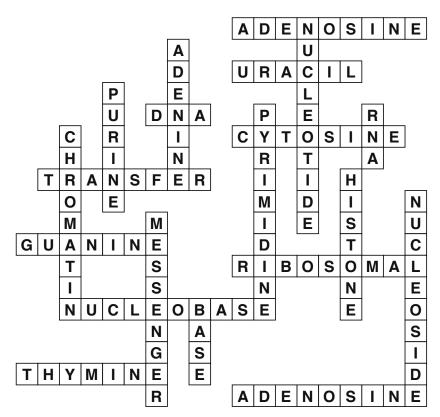
Proteins



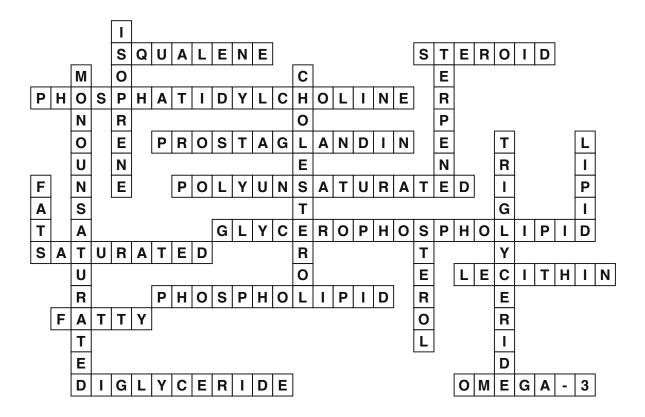
Carbohydrates



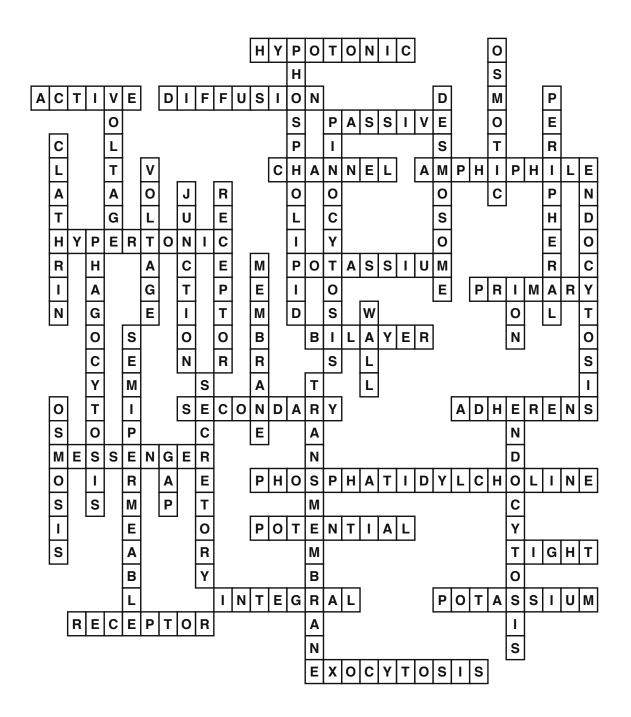
Nucleic Acids



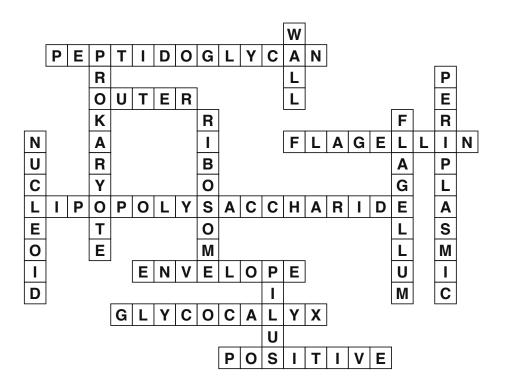
Lipids



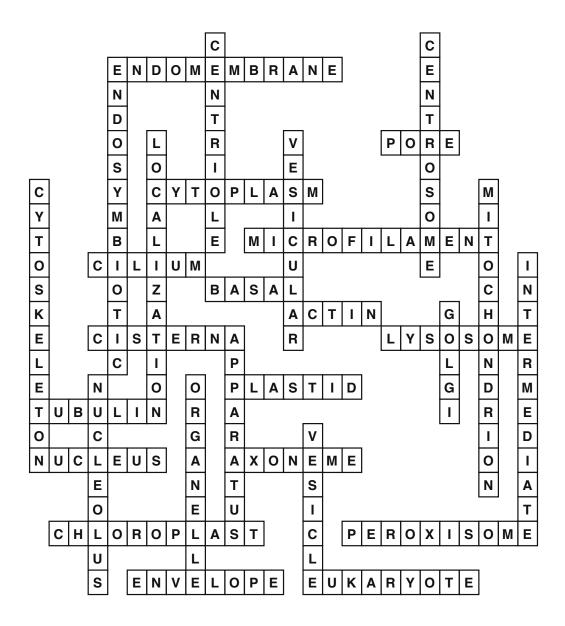
Biological Membranes



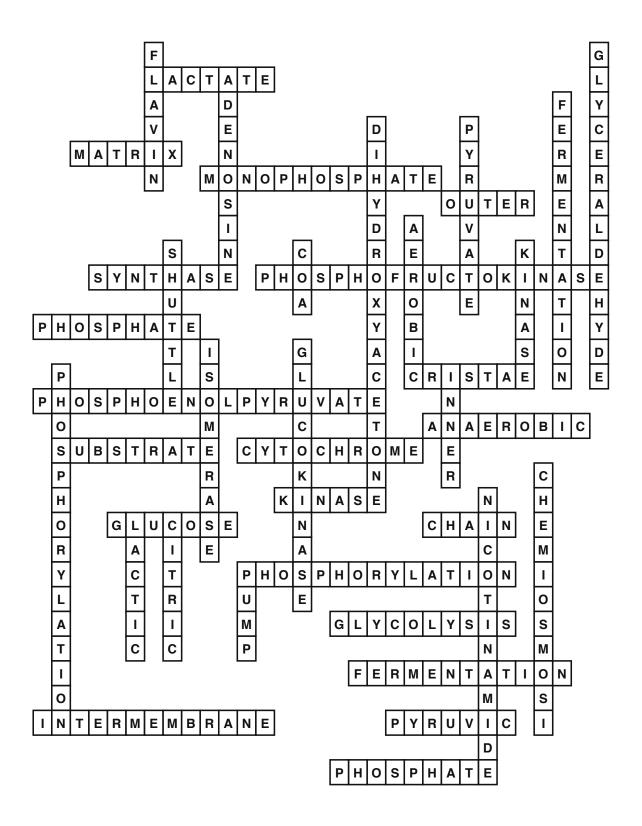
Prokaryotes



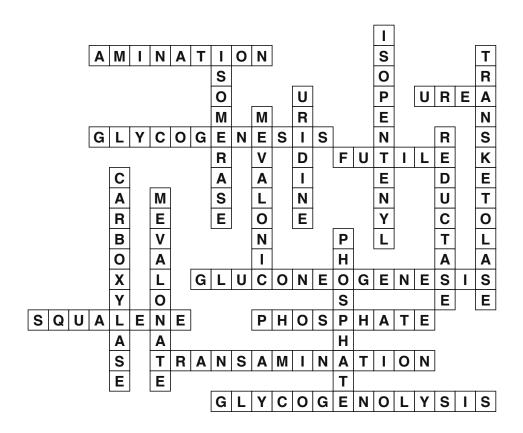
Eukaryotes



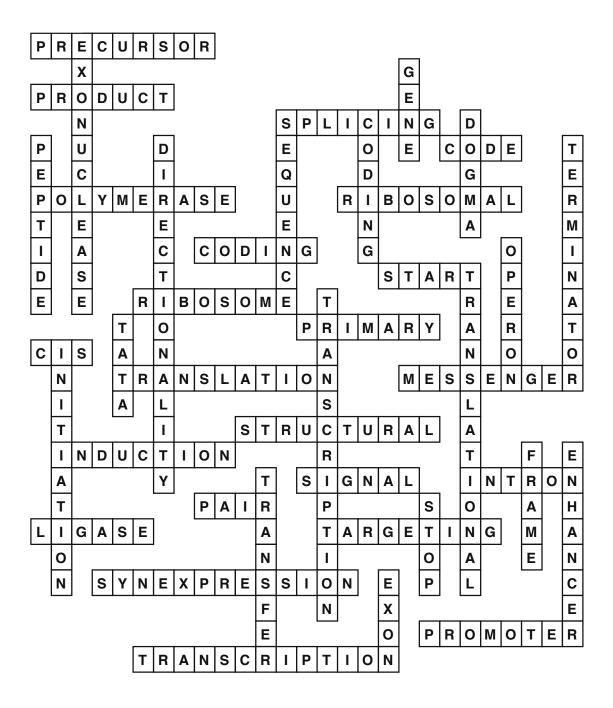
Energy Metabolism



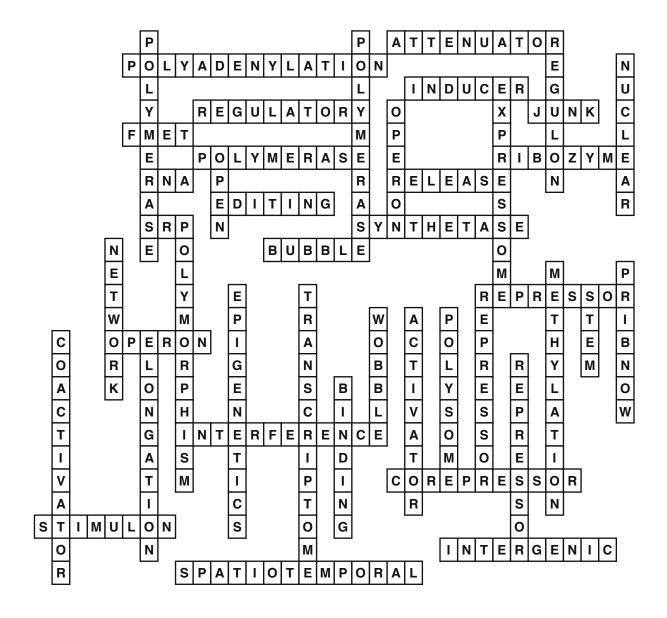
Metabolic Integration



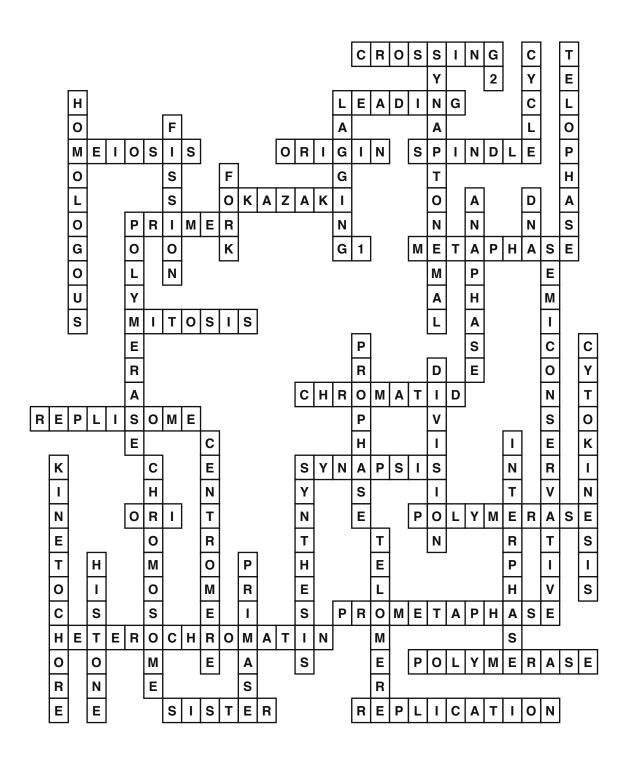
Gene Expression - Part 1



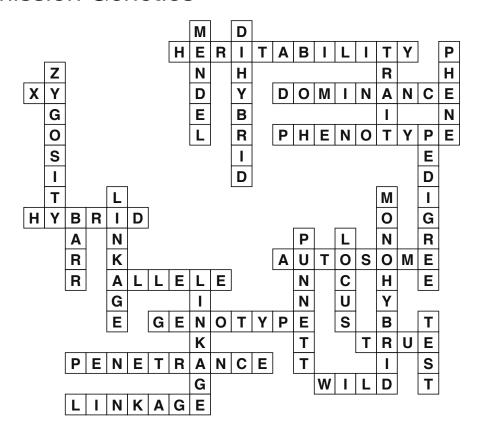
Gene Expression - Part 2



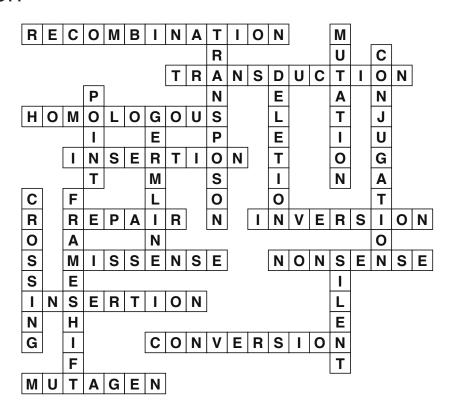
Cellular Reproduction



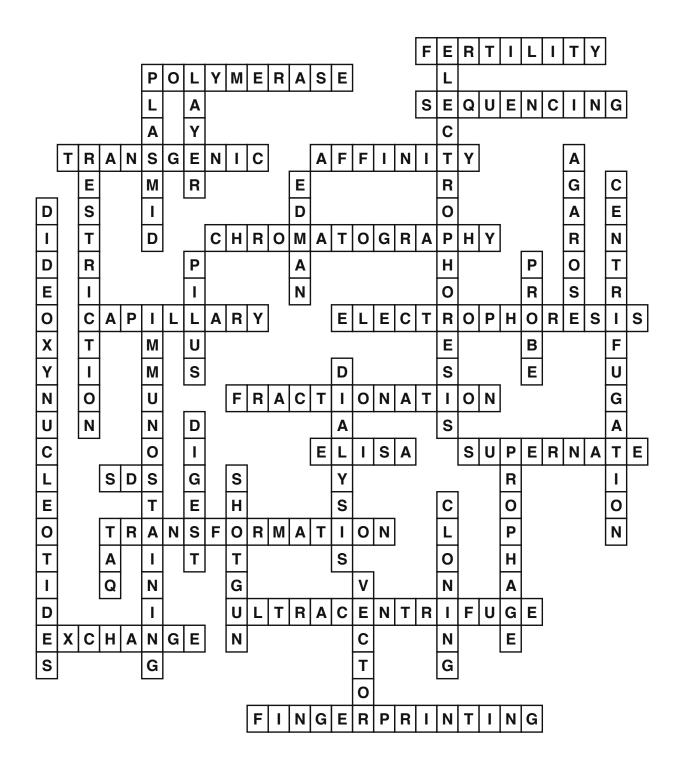
Transmission Genetics



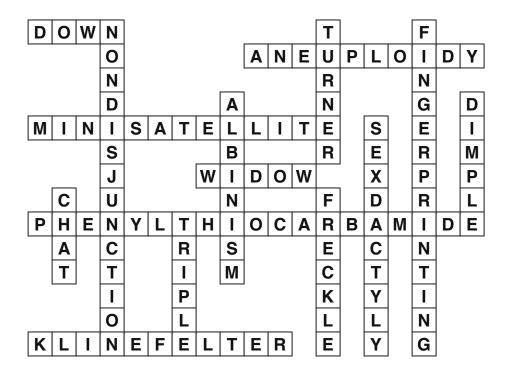
Mutation



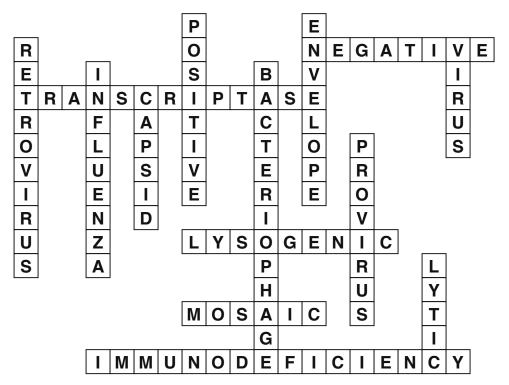
Genomics Laboratory



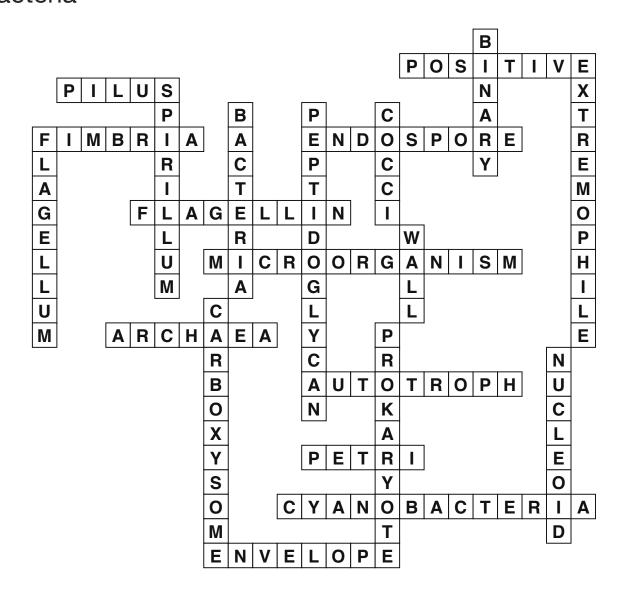
Human Genetics



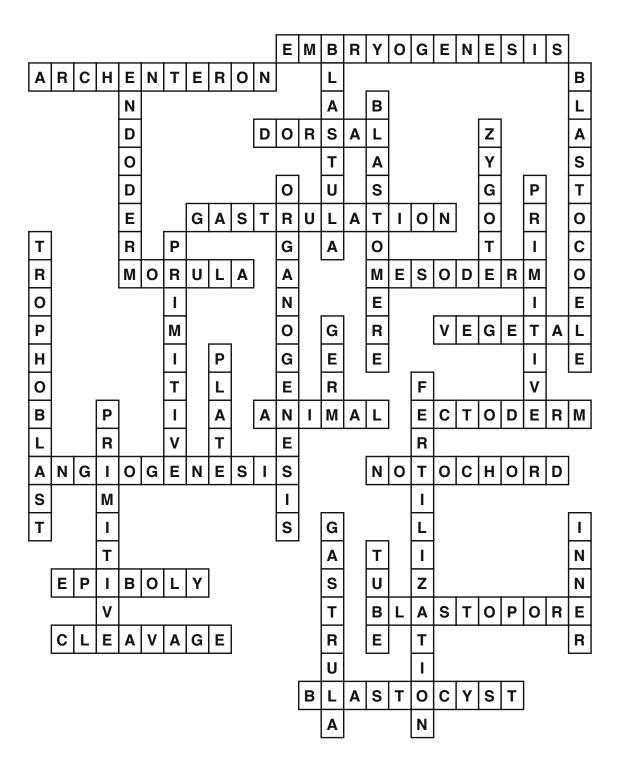
Viruses



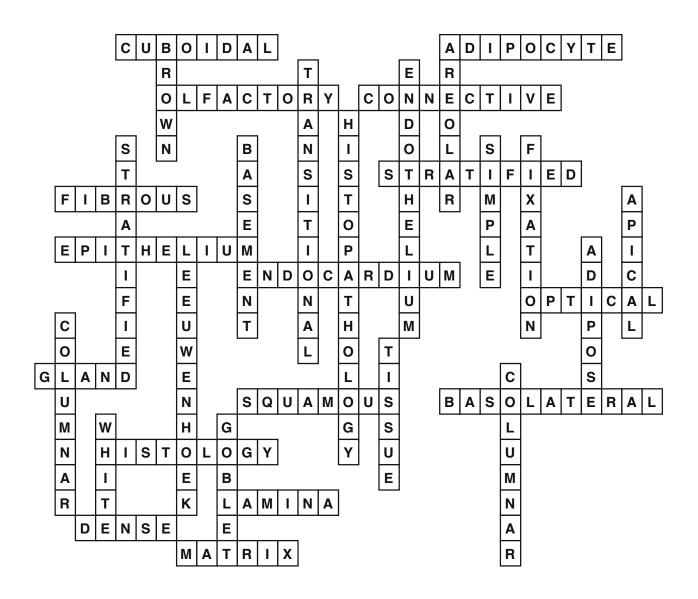
Bacteria



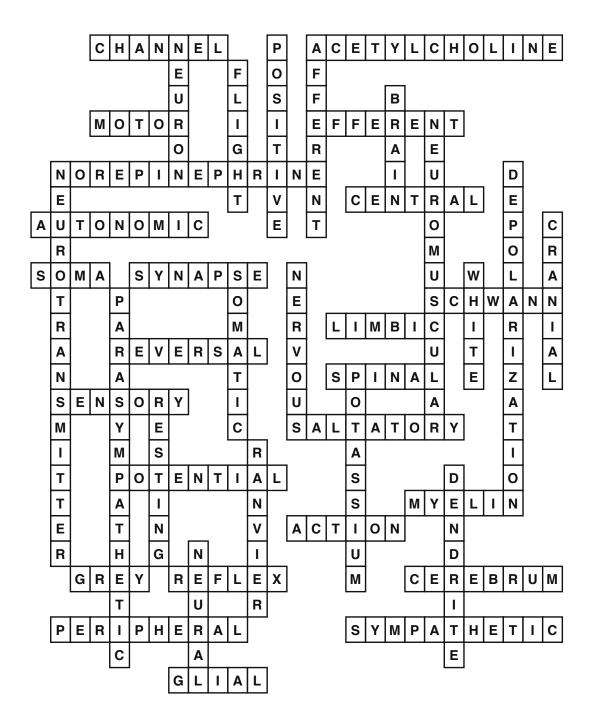
Embryology



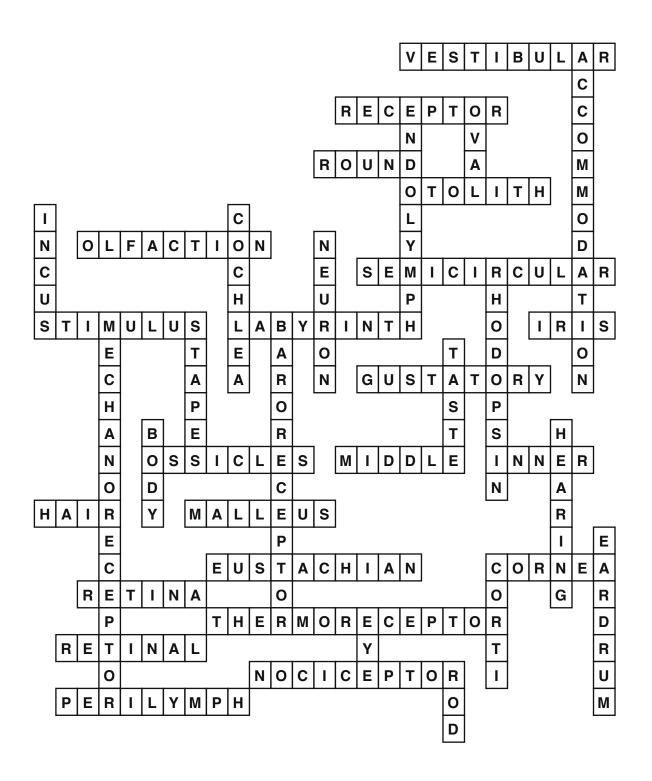
Animal Tissues



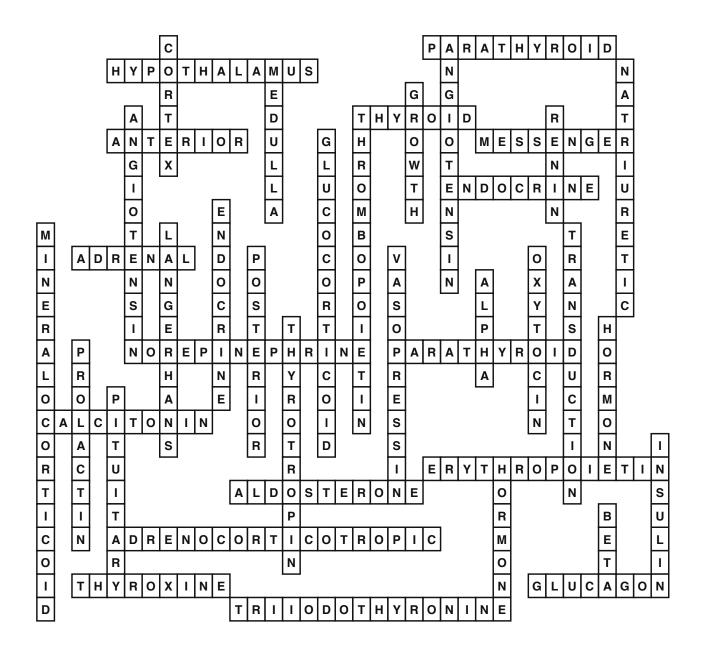
Nervous System



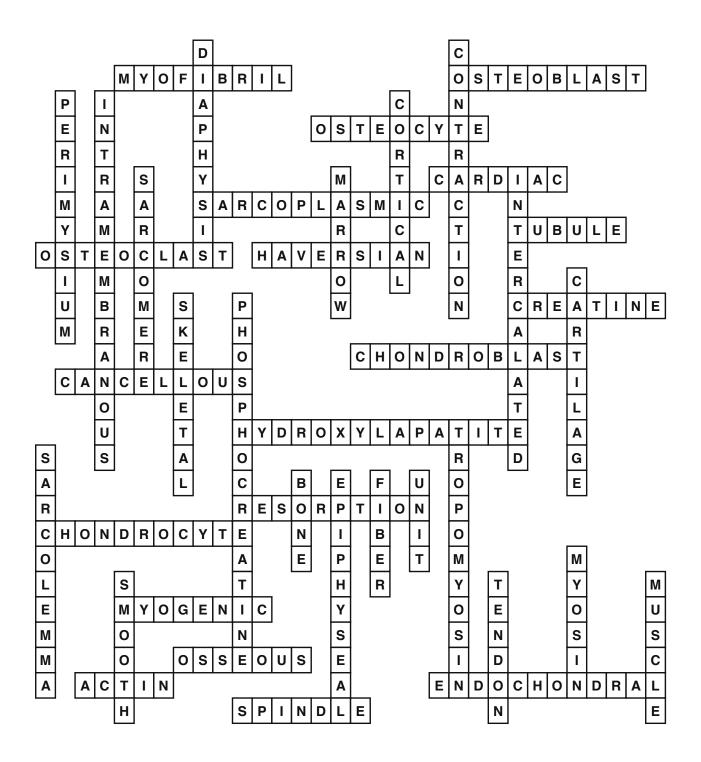
Sensory Systems



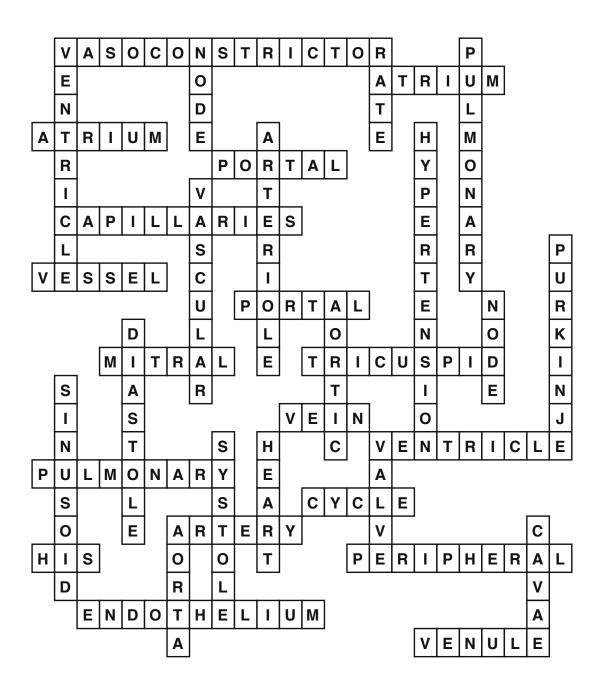
Endocrine System



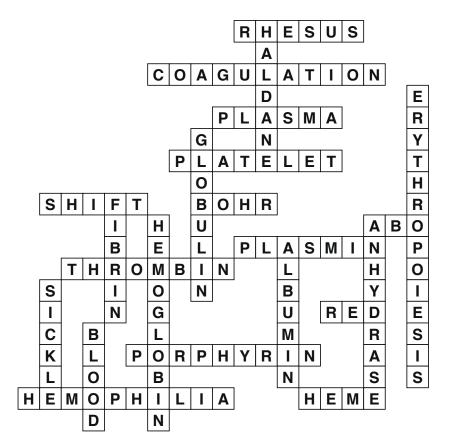
Musculoskeletal System



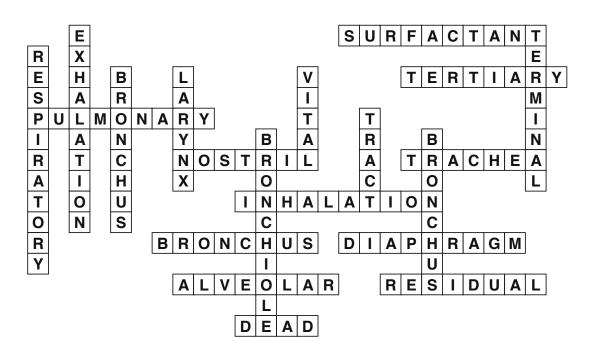
Cardiovascular System



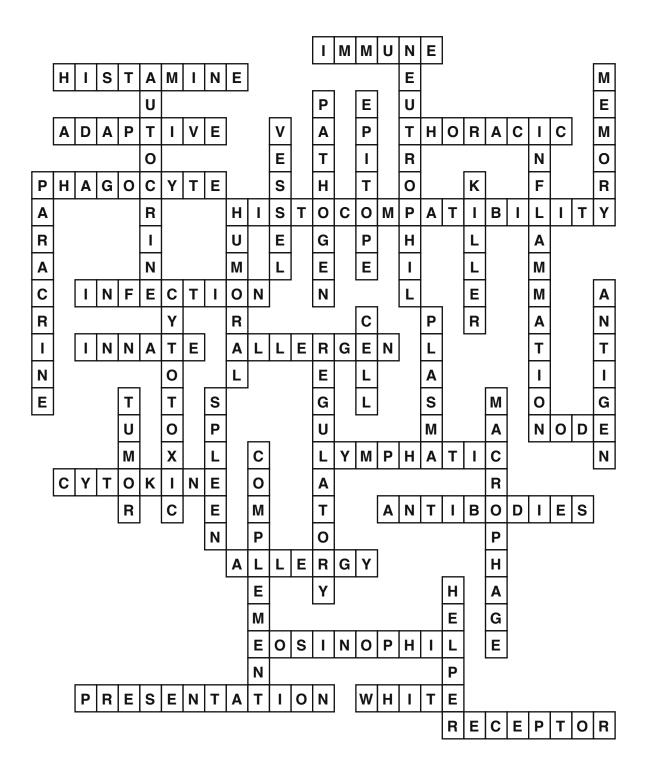
Blood



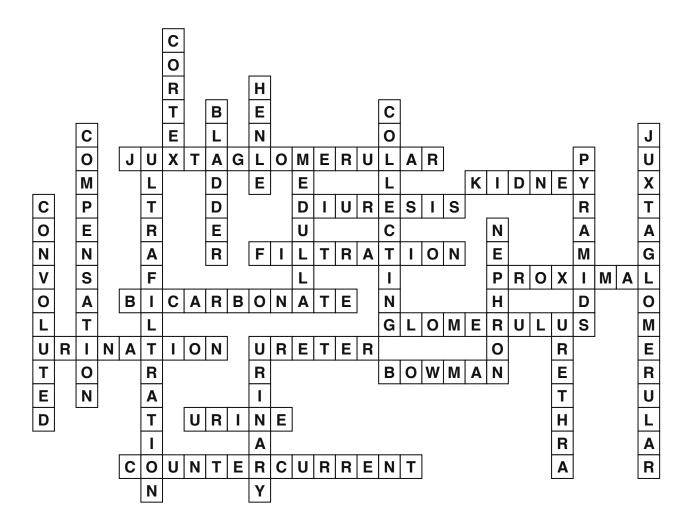
Respiratory System



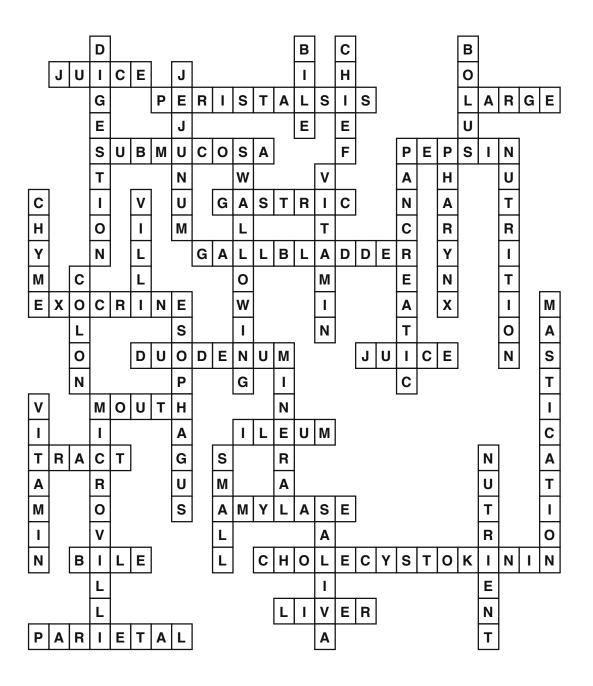
Immunity



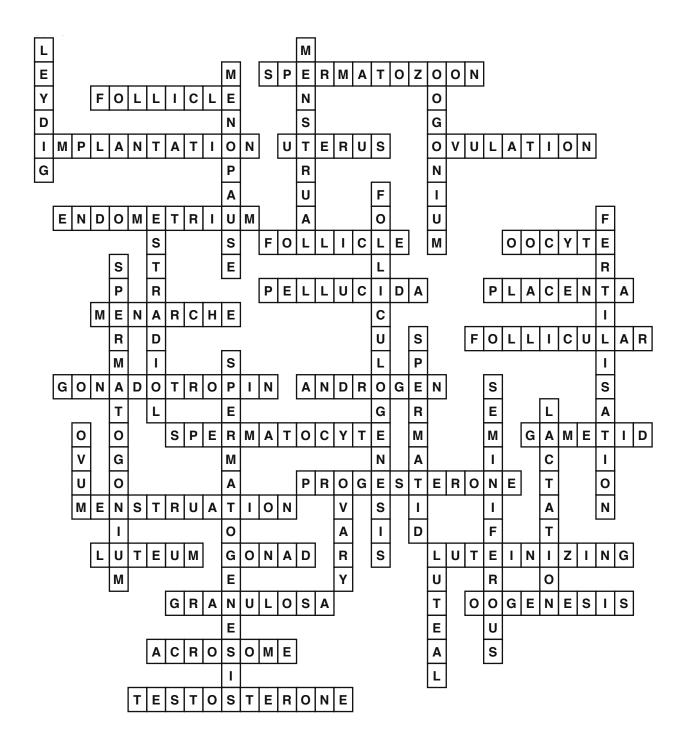
Urinary System



Digestive System



Reproductive System



Evolution

